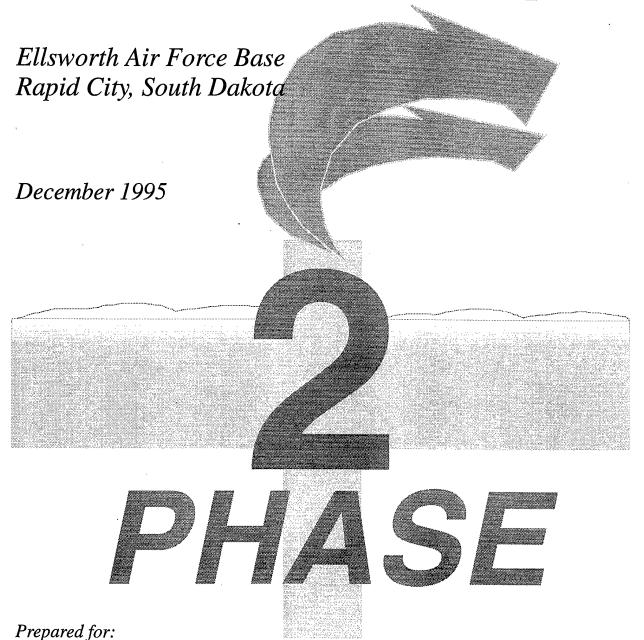
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2-Phase[™] Pilot Test Technology Evaluation Report



U.S. Army Corps of Engineers Omaha District

AGM01-01-0298

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13 December 1995

U.S. Army Corps of Engineers, Omaha District ATTN: CEMRO-ED-EB (Robert Zaruba) 215 North 17th Street Omaha, Nebraska 68102-4978

SUBJECT: Contract No. DACA45-93-D-0027, Delivery Order No. 27, Mod No.2;

Final Ellsworth AFB 2-Phase Pilot Test technology Evaluation Report

Dear Mr. Zaruba:

Enclosed you will find two copies of the final Ellsworth AFB 2-Phase Pilot Test Technology Evaluation Report for the work accomplished at Operable Unit No. 1 and associated response to comments. The only comments received were from the Omaha District Corps of Engineers (COE) staff. Ellsworth AFB staff (Mr. Dell Petersen and Mr. John DeYoe) were contacted and indicated that they would not be providing written comments.

Additional copies of the report have been distributed to Ellsworth AFB, and to Ms. Margaret Calvert at Langley AFB.

If you have any questions regarding this report, please contact me at (916) 857-7281 or Mr. James Machin at (512) 419-5280.

Sincerely,

FRANCIS E. SLAVICH, P.E.

Project Manager

c: Margaret Calvert, ACC/ESVW (2)
Dell Petersen, Ellsworth AFB (7)
James Machin, Radian
Bill BuChans, Radian
John Yackiw, Radian

Gary Dyke, Radian

ELLSWORTH AFB 2-PHASE™ PILOT TEST TECHNOLOGY EVALUATION REPORT

Ellsworth Air Force Base South Dakota

Prepared for:

U.S. Army Corps of Engineers Omaha District ATTN: CEMRO-ED-EB 215 North 17th Street Omaha, Nebraska 68102

Prepared by:

Radian Corporation 1093 Commerce Park Drive, Suite 100 Oak Ridge, Tennessee 37830 Doc. #F950824.1DP51

December 1995

No. 50473 Exp. 6-30-97

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ACRONYMS

ACC Air Combat Command

AFB Air Force Base

BGS Below Ground Surface

BTEX Benzene, Toluene, Ethylbenzene, and Xylenes

DCA Dichloroethane
DCE Dichloroethylene

DNAPL Dense Nonaqueous Phase Liquid

EPA U.S. Environmental Protection Agency

ESVE Enhanced Soil Vapor Extraction FPTA Fire Protection Training Area

GAC Granular Activated Carbon

HQ Headquarters

IRA Interim Remedial Action

LNAPL Light Nonaqueous Phase Liquid
MCL Maximum Contaminant Level
O&M Operation and Maintenance

OU-1 Operable Unit 1
PCE Tetrachloroethylene

PREECA Presumptive Remedy Engineering Evaluation/Cost Analysis

PVC Polyvinyl Chloride

RI Remedial Investigation
SVE Soil Vapor Extraction

TCE Trichloroethylene

TPE Xerox 2-Phase[™] Extraction

USAF U.S. Air Force

F950824.1

VOA Volatile Organic Analysis
VOC Volatile Organic Compound

1.0 INTRODUCTION

In June 1995, Ellsworth Air Force Base (AFB), in Rapid City, South Dakota, and Radian Corporation (Radian) completed a five-day pilot treatability test at the fire protection training area burn pit of Operable Unit 1 (OU-1) using the Xerox 2-Phase Extraction (TPE) technology. This report provides a summary of the methodology used during the test, the test results, and base-specific recommendations.

1.1 Purpose/Objectives

On 5 May 1995, Headquarters (HO) Air Combat Command (ACC) published *United States Air* Force Presumptive Remedy Engineering Evaluation/Cost Analysis (PREECA) [U.S. Air Force (USAF) 1995] as a standardized decision framework specifying the criteria and associated decision logic necessary for implementing a nontime-critical removal action. This decision framework, developed by Radian in conjunction with the U.S. Army Corps of Engineers and the USAF, combines the standard Comprehensive Environmental Response, Compensation, and Liability Act nontime-critical removal action process with the concept of presumptive remedies and a "plug-in" logic tree approach. The result is a "generic" remedy selection document for all USAF installations that facilitates early and substantial risk reduction at USAF sites. PREECA applies only to a closely defined subset of conditions that the USAF has found to be common and that pose sufficient risk to justify nontime-critical removal actions. This methodology was not intended to be used at sites where the need for cleanup actions is not readily apparent.

PREECA focuses on remedies that can satisfy the majority of common USAF contamination situations, namely in situ bioventing, soil vapor extraction (SVE), groundwater pump and treat for containment, and capping. However, REECA is intended to be updated as new, successful remedies are established. The USAF is currently gathering extensive cost and performance data at a number of contaminated

sites for intrinsic groundwater remediation, bioslurping, and 2-Phase[™] extraction. As part of this effort, HQ ACC has contracted with Radian through the Omaha District Corps of Engineers to evaluate the TPE technology for inclusion in the USAF PREECA. Radian, in conjunction with the USAF, developed an initial remedy profile for TPE as part of the original PREECA efforts.

This report presents the results of the TPE pilot test conducted at Ellsworth AFB in June 1995. It compares the pilot test results to PREECA's initial remedy profile for TPE and demonstrates that TPE is an effective technology for use at Ellsworth AFB. In addition, it presents data on additional objectives for the pilot test, which were to:

- Demonstrate the contaminant removal effectiveness of the TPE technology;
- Determine the feasibility of installing a full-scale system;
- Collect sufficient engineering data to facilitate the design, installation, and operation of a full-scale extraction and treatment system; and
- Assist in the prevention of contaminant migration, thereby minimizing the threat of exposure to human health and the environment.

TPE was selected for testing at the OU-1 burn pit because the low saturated zone permeabilities limit the effectiveness of conventional pump and treat systems to capture groundwater contaminant plumes. Ellsworth AFB is in the process of implementing an interim remedial action at OU-1 that will consist of SVE and groundwater pump and treat (dual-phase). A large complement of information exists for OU-1, including the remedial investigation (RI) report (EA Engineering 1994a) and the data from two previous treatability studies. The TPE technology is designed to enhance control of groundwater plumes in low- to moderate-

permeability formations, as well as to remove contaminants from the saturated and vadose zones.

1.2 Site Background

OU-1 is located in the southwestern portion of Ellsworth AFB as shown in Figure 1-1. This site was used as a fire protection training area (FPTA), resulting in significant soil and groundwater contamination. Previous field activities in the area have included installation and sampling of wells, two soil vapor surveys, an SVE pilot test, water level measurements, aquifer testing, and groundwater recovery and treatment. Data collected from these activities, in addition to data from this project, have been used to characterize the subsurface features and the nature and relative extent of contamination at the site.

1.2.1 Subsurface Features

The OU-1 area is underlain by approximately 15 to 18 ft of fill and native soil (alluvium) that overlies shale bedrock (Pierre Shale). The native soil consists primarily of fine grained sands, silts, and clays of low permeability, but much of the FPTA has been filled with coarser grained sand and gravel of higher permeability. The upper 10 to 15 ft of the shale is weathered and consists of variably fractured light olive gray to dark olive gray clay, which increases in competence with depth. The permeability of the weathered and fractured shale is low. Figure 1-2 is a geologic cross-section demonstrating the distribution of soil types in the test area.

Based on the lithologic information from inspection of soil samples, the relative permeability of the soils can be inferred. The upper 3 to 5 ft of sandy clay and clayey gravel have lower permeabilities. Silt, sand, and gravel deposits between approximately 3 and 18 ft represent higher permeability materials. Data from an SVE test at the site conducted by EA Engineering, Science, and Technology indicate a relative permeability to air of 200 to 300 Darcy in the vadose zone alluvium. Finally, weathered

and fractured shale beginning at approximately 15 to 18 ft are representative of low permeability materials.

The primary TPE well (existing monitoring well MW 930101) within the burn pit was completed in both the alluvium and shale bedrock and was screened from 12 to 42 ft below ground surface (BGS). Depth to groundwater is 16.5 ft BGS (based on water level measurements from June 1995). The saturated alluvial thickness ranges from 0.5 to 1.5 ft, with the remainder of the saturated zone occurring in the weathered and fractured shale. Hydraulic conductivity in the saturated zone is very low $(9.5 \times 10^{-6} \text{ cm/sec})$ based on previous slug tests at MW 930101. The sustained pumping yield for this well has not been measured but is low, based on recharge times following well purging during previous sampling.

An existing enhanced soil vapor extraction (ESVE) well was used for a secondary TPE test following the primary test at MW 930101. This ESVE well was installed in the saturated fractured shale zone to a depth of 35 ft BGS with a 10 ft screened section (24 to 34 ft). Depth to groundwater was 17.1 ft BGS.

Additional piezometers and vapor points were installed during a previous treatability study in 1994. Data on depths and screen intervals is presented in Section 2.0 of this report.

Groundwater flow direction is generally to the south and southwest.

1.2.2 Nature and Extent of Contamination

Both vadose zone and capillary fringe soil contain high concentrations of JP-4; benzene, toluene, ethylbenzene, and xylenes (BTEX), and chlorinated volatile organic compounds (VOCs). The following general concentration ranges of these contaminants were found in FPTA area soils:

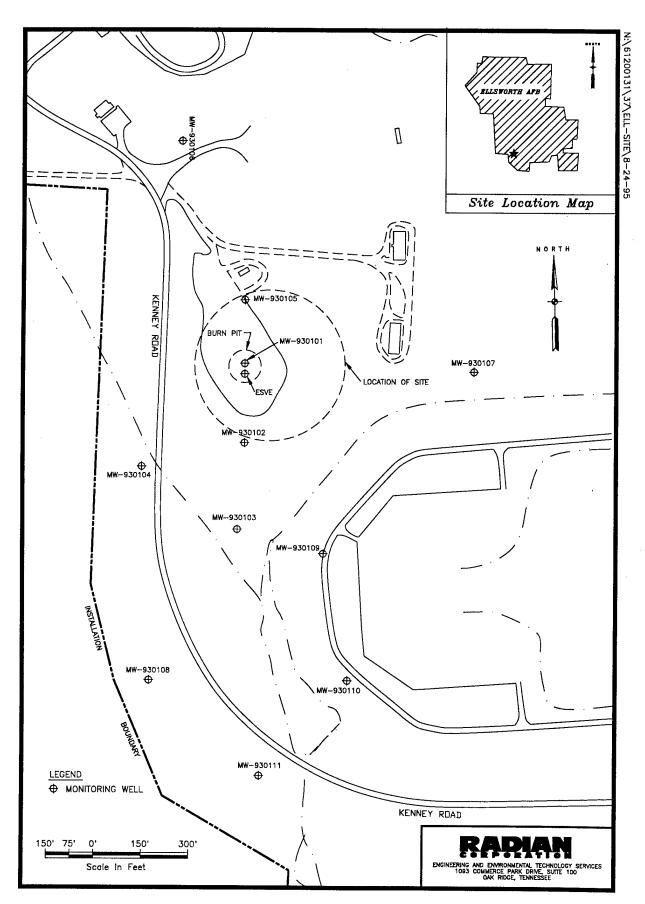


Figure 1-1. OU-1 Site, Ellsworth AFB

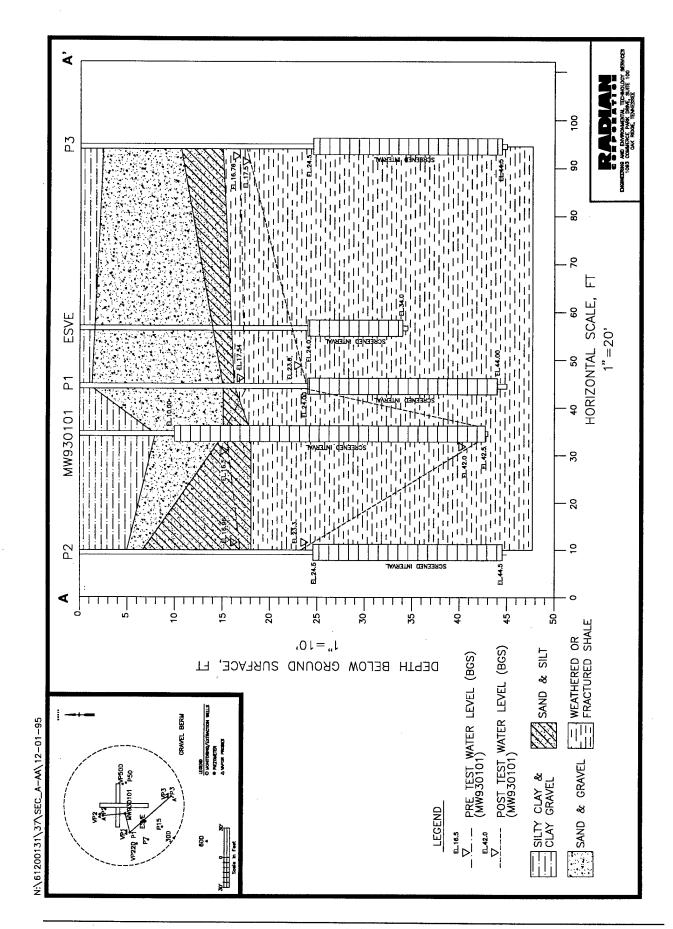


Figure 1-2. OU-1 Conceptual Cross Section

Contaminant	Vadose Zone (μg/kg)	Capillary Fringe (µg/kg)
JP-4	100,000s to 1,000,000s	100,000s to 1,000,000s
Total BTEX	ND to 100,000s	1,000s to 10,000s
Total chlorinated VOCs	ND to 10s	ND to 100s

ND = Nondetect

The RI indicated that light nonaqueous phase liquids (LNAPLs) (JP-4 and the related BTEX compounds) were found in soil at the south end of OU-1; JP-4 is present in much higher concentrations than other contaminants. In general, the distribution of JP-4 reflects the extent of organic contaminants in soil.

Trichloroethylene (TCE), benzene, tetrachloroethylene (PCE), 1,1-dichloroethylene (1,1-DCE), and 1,2-dichloroethane (1,2-DCA), were reported at or above their established Safe Drinking Water Act Maximum Contaminant Level (MCL) in one or more groundwater samples at OU-1. Dense nonaqueous phase liquids (DNAPLs) were not found in the OU-1 groundwater; however, LNAPLs were found at the northern and southern ends of OU-1.

1-5

2.0 2-PHASE[™] EXTRACTION TEST METHODOLOGY

The following information on the technical approach and the sampling and analytical methodologies is a summary of the Ellsworth AFB 2-Phase [™] Extraction Pilot Test Work Plan (Radian Corporation 1995). Additional details are contained in that document.

2.1 <u>Test Procedures</u>

A four-day pilot test was conducted on MW 930101. Three new piezometers, P-1, P-2, and P-3, and three vapor point clusters, each consisting of a shallow and a deep point, were constructed for the test. Existing monitoring wells in the area surrounding MW 930101 were also used to measure changes in subsurface conditions. Additionally, following the four-day test, a oneday pilot test was conducted on a well previously installed for an ESVE study. The same piezometers, vapor points, and monitoring wells were used for that test. The locations of the test wells and monitoring points are shown in Figure 2-1. Well, piezometer, and vapor point characteristics are summarized in Table 2-1. Well logs are included in Appendix B.

2.1.1 Installation of Piezometers and Vapor Probes

2.1.1.1 Piezometers

The piezometers (P-1, P-2, and P-3) were installed in order to monitor the response of the aquifer to the test. Piezometers were located at distances of 11.5, 25.2, and 51.8 ft from extraction well MW 930101. The locations were chosen such that data from the wells would supplement data collected from existing wells previously installed in the area. Figure 2-1 shows the locations of the new and existing wells.

The piezometers were installed on 20 and 21 June 1995 using a hollow stem auger drilling rig with 7.5-in. outside diameter augers. Soil samples were collected from selected intervals so that lithologic logs could be prepared (Appendix

B). The piezometers were constructed with 2-in. diameter polyvinyl chloride (PVC) well casing and screen. The well casing, sand pack, and bentonite seal were installed through the augers to ensure the stability of the well bore. The details of the wells are contained in the completion logs in Appendix B. In general, 20-ft long screens were placed within the weathered Pierre Shale at a depth of 25 to 45 ft below the ground surface.

After the wells were completed, they were developed to remove silt and clay and ensure communication with the aguifer. The wells were first surged with a 2-in., vented, surged block to loosen up the fine material from the sand pack so that it could be removed. The piezometers were then purged using a disposable bailer. Water quality was monitored during development by visually observing the silt and clay content of the water and by pH meter measurements. Development was judged complete when the pH was stable and turbidity of the water had decreased to the satisfaction of the supervising geologist. All wells were bailed essentially dry within an hour and allowed to recharge. Development logs are contained in Appendix B.

2.1.1.2 Vapor Probes

Three sets of two nested vapor probes (a total of six probes) were installed at locations chosen to supplement the existing probes at the site. The new probes were labeled VP-1S, VP-1D, VP-2S, VP-2D, VP-3S, and VP-3D and are shown on Figure 2-1.

The vapor probes were installed on 21 and 22 June 1995 using a hollow stem auger drilling rig and 10-in. outside diameter augers. Soil samples were not collected from the borings for the probes as they were all located within approximately 5 ft of one of the new piezometers. The probes were constructed using 1-in. diameter PVC casing. Slots were cut in the bottom 2 ft of the probes with a hacksaw to allow communication with the vadose zone. Two probes were installed per boring at depths

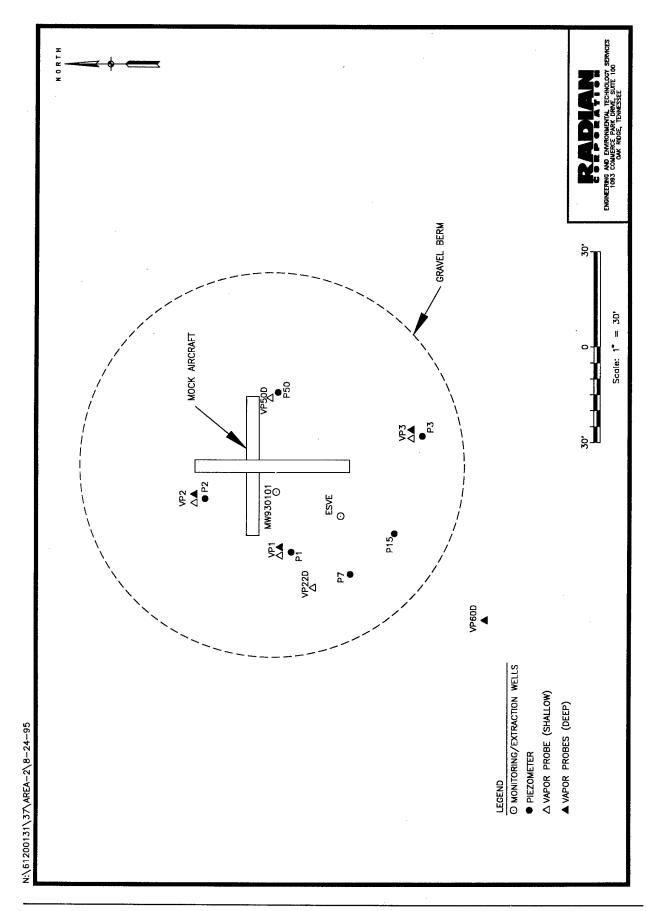


Figure 2-1. OU-1, Test Well and Monitoring Wells, Ellsworth AFB

Table 2-1
Summary of Wells and Monitoring Point Characteristics

Well/Piczometer ID	Used to Monitor	Total Depth (ft BGS)	Screened Interval (ft BGS)	Approximate Distance from MW 930101 (Test Well)
VP-1D	Induced Vacuum	15	13–15	15.3
VP-1S	Induced Vacuum	5.5	3–5	15.3
VP-2D	Induced Vacuum	15	13–15	30.5
VP-2S	Induced Vacuum	5.5	2.5–5	30.5
VP-3D	Induced Vacuum	16	13–15	45.4
VP-3S	Induced Vacuum	5.5	3–5	45.4
P-1	Water Level	44.5	2444	11.5
P-2	Water Level	45	24.5-44.5	25.2
P-3	Water Level	45	24.5-44.5	51.8
P-7	Water Level	35	15–22	22.8
P-15	Water Level	35	15–22	37.4
P-50	Water Level	23	15–22	39.7
VP-22D	Induced Vacuum	14	11.5–13.5	21.9
VP-50D	Induced Vacuum	14	11.5–13.5	39.7
VP-60D	Induced Vacuum	14	11.5–13.5	81.4
MW 930101	Extraction Well	43	12–42	0
ESVE Well	Extraction Well	35	24–34	22.5

BGS = Below Ground Surface

of 15 ft and 5 ft. The slotted intervals were separated by a bentonite seal to eliminate communication through the sand pack.

2.1.2 Test Equipment

The test was conducted using a trailer-mounted, 25-horsepower, high-vacuum extraction unit capable of producing an air flow rate of 300 cubic feet per minute (acfm) at 25 in. mercury. The system is shown in schematic in Figure 2-2. Extracted groundwater was treated using liquid-phase granular activated carbon (GAC) before discharge to a 5000-gal temporary storage tank; extracted vapor was discharged to the atmosphere.

After the wastewater analytical results were received and reviewed by the base environmental staff, the wastewater was transported and discharged to the sanitary sewer. A general schematic of a TPE well is shown in Figure 2-3. Procedures followed during the testing are summarized in the work plan described in Section 2.0.

In the work plan, Radian proposed performing this pilot test using an existing monitoring well (MW 930101). Initial results from MW 930101 suggested that the majority of contaminant removal was from the vadose zone. This is because the well screen in MW 930101 was open across both the upper gravelly vadose zone and the lower fractured shale saturated zone. When steady state flow conditions were reached three days after the start of the test, Radian requested permission to deviate from the work plan from the Omaha District, U.S. Army Corps of Engineers, and Ellsworth AFB environmental staff. Permission was received to move the test from MW 930101 to ESVE after the fourth day of the TPE test.

The screened interval in ESVE is in the fractured shale in the saturated zone and is isolated from the upper sandy vadose zone. Therefore, data from this portion of the test were useful in assessing the performance of TPE strictly in the fractured shale.

2.2 <u>Sampling and Analytical</u> Methodologies

All sampling and analytical procedures (except where noted) were conducted in accordance with procedures and protocols described in the U.S. Environmental Protection Agency (EPA)-approved Ellsworth AFB Quality Assurance Project Plan. Sampling locations and frequency are summarized in Table 2-2.

2.2.1 Sampling Methodology

System parameters and ambient air conditions were measured through various vacuum gauges, meters, and thermometers included on the TPE trailer. Groundwater drawdown in the observation wells was measured using an electronic water level meter, and induced vacuum was measured using Magnehelic® gauges. Data collected were recorded on field data tables (Appendix C).

Baseline groundwater samples from MW 930101 were collected prior to TPE testing in 40-mL volatile organic analysis (VOA) vials using a dedicated Teflon® bailer. Prior to collecting the baseline samples, three well volumes of water were purged from the well. Approximately one hour after ending the test, post-test groundwater samples were collected using the dedicated bailer.

Water samples collected during the test were taken directly from the TPE trailer knock-out pot with VOA vials. All VOA vials were iced and stored in a dedicated cooler until shipped to Energy Laboratories, Inc., in Rapid City, South Dakota.

Vapor samples were collected using disposable syringes and evacuated vials provided by Microseeps Inc., Pittsburgh, Pennsylvania. Once the samples were collected, they were stored at ambient conditions until shipped to the Microseeps laboratory for analysis.

Quality control samples were also collected in the field. Duplicate water and vapor samples were collected at a 10% frequency by the methods

previously described. Trip blanks accompanied the VOA vials throughout shipping and handling.

2.2.2 Analytical Methodology

Groundwater samples were analyzed for VOCs by EPA Method SW-8260. Soil vapor samples were analyzed for VOCs by Microseeps Analytical Method AM 4.02.

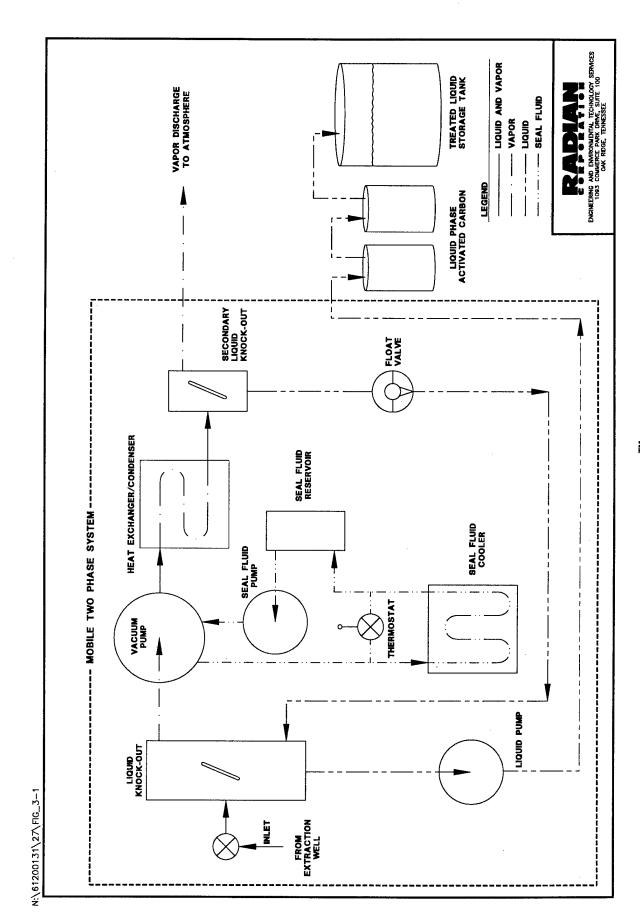


Figure 2-2. 2-Phase System Schematic

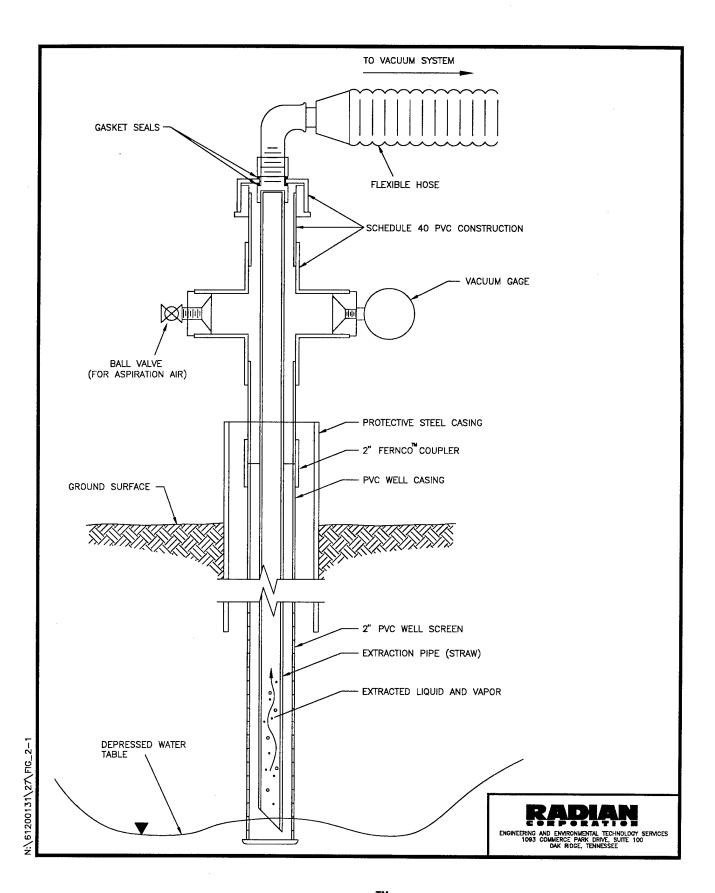


Figure 2-3. Schematic of a 2-Phase[™] Extraction Well Configuration

Table 2-2 Frequency of Sample Collection and Source Monitoring

	Shedule									
		Ambient		Measure	Groundwater	Water Levels at	0.975	Induced Vacuum		Water Samples from
Day	Hour	Barometric Pressure	Ambient Temperature	Water Level at Test Well	Sample from Test Well	Sample from Groundwater Test Well Piezometers	Vapor Samples	at Soil Vapor Monitoring Probes	System Parameters	Knock-Out Pot
0	Before	Х	X	Х	X	×				
1	0.25	X	X	X		X	×	X	×	
	1		X			X	X	×	×	
-	2		X			X	X	X	×	×
1	4		X			X	X	X	×	×
-	8		X			X		X	×	×
2	0	×	X			X	X	X	×	×
2	4		X			X	X	X	×	×
2	8		X			X	*	X	×	X
3	0	×	×			X	X	X	×	×
3	4		×			X	X	X	X	
3	8		×			X	*	X	X	X
4	0	×	×			X	X	X	X	X
4	4	,	×			X	X	X	X	X
4	4.25			×		X		X	X	
4	4.25			X		X		X	X	X
4	8		×			X	X	X	X	×
5	0	×	X			X	X	X	X	X
5	4		×			X	X	X	X	X
5	4.25		×	×	X					

Note: Groundwater/water samples analyzed for VOCs by Method SW-8260. Vapor samples analyzed for VOCs by Microseeps Analytical Method AM 4.02.

3.0 TEST RESULTS AND CONCLUSIONS

A critical step toward adding another presumptive remedy to the PREECA process is to compare that remedial technology's test results, referred to here as the "site-specific profile," to its PREECA remedy profile and determine the extent to which the two profiles match. The remedy profile comprises the performance data (including site selection criteria, process and methodology descriptions, and the acceptable range of quantitative results) by which the effectiveness of the presumptive remedy will be judged.

Radian performed a four-day test on MW 930101 and a one-day test on the ESVE well. Table 3-1 summarizes the results achieved using the TPE system at MW 930101 and the ESVE well. The results of these two tests are described in Section 3.4.

Table 3-1
Summary of Results

System Parameter	MW 930101	ESVE
Groundwater Extraction Rate	0.22 gpm	0.11 gpm
Soil Vapor Extraction Rate	21-57 scfm	14 scfm ^a
Contaminant Removal Rate	58–122 lb/day	0.07 lb/day
Radius of Influence (Groundwater)	>50 ft	b

^aMeasurement includes aspiration air.

^bRadius of influence results were inconclusive and will be discussed in Section 3.2.

gpm = gallons per minute

scfm = standard cubic feet per minute

Based on the results of the pilot-scale TPE test conducted at Ellsworth AFB OU-1, Radian has constructed a site-specific profile for OU-1. A comparison of this site-specific profile to the PREECA's TPE remedy profile is presented in Table 3-2. Note that the OU-1 profile compares

favorably with the corresponding TPE remedy profile.

3.1 System Operation

Physical and analytical data were analyzed to determine the following:

- Baseline VOC concentrations in groundwater;
- The major VOC constituents in the vapor and water streams;
- Average groundwater and soil vapor extraction rates;
- Average VOC extraction rates and total pounds of VOCs removed;
- The relationship between time and VOC concentrations;
- The relationship between time and vapor and water flow rates; and
- The relationship between distance and groundwater drawdown and induced vacuum, including radii of influence.

3.2 Radii of Influence and Production Rates

The following sections describe groundwater production rates and vapor radius of influence.

3.2.1 Groundwater

The groundwater flow rate was measured using a totalizing flow water meter and is plotted along with the vapor flow rate on Figure 3-1 for both the MW 930101 and ESVE tests. Water table drawdown was measured in piezometers P-1, P-2, P-3, P-7, and P-50 during the tests (Appendix D). A plot of drawdown versus time is presented in Figure 3-2, and a water table contour map for the end of the MW 930101 test is shown in Figure 3-3.

Figure 3-1. Vapor and Liquid Flow Rates

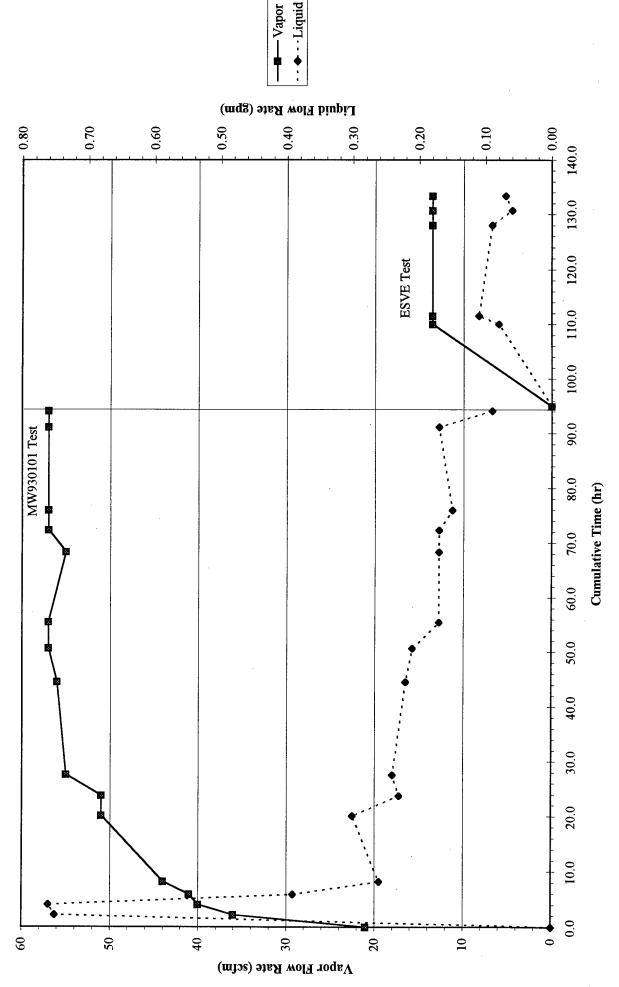
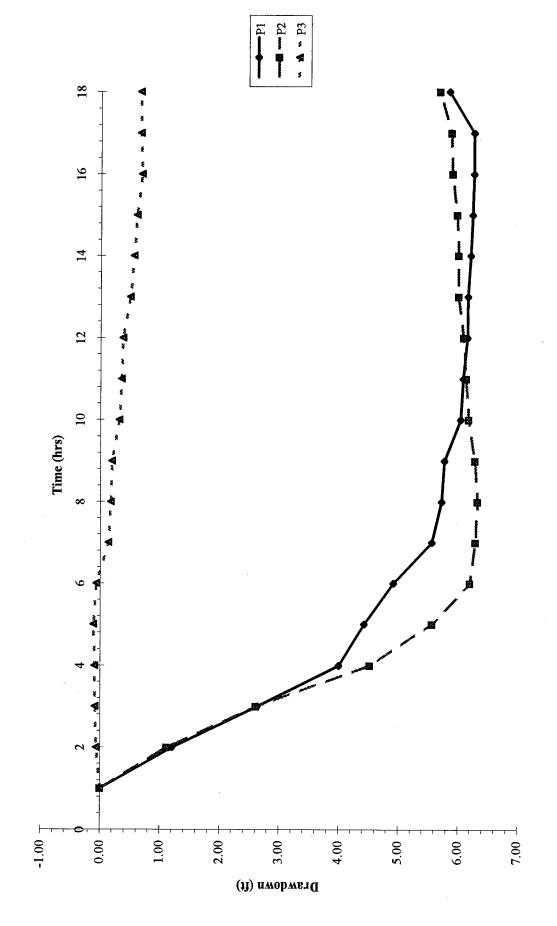


Figure 3-2. Water Level Drawdown Over Time



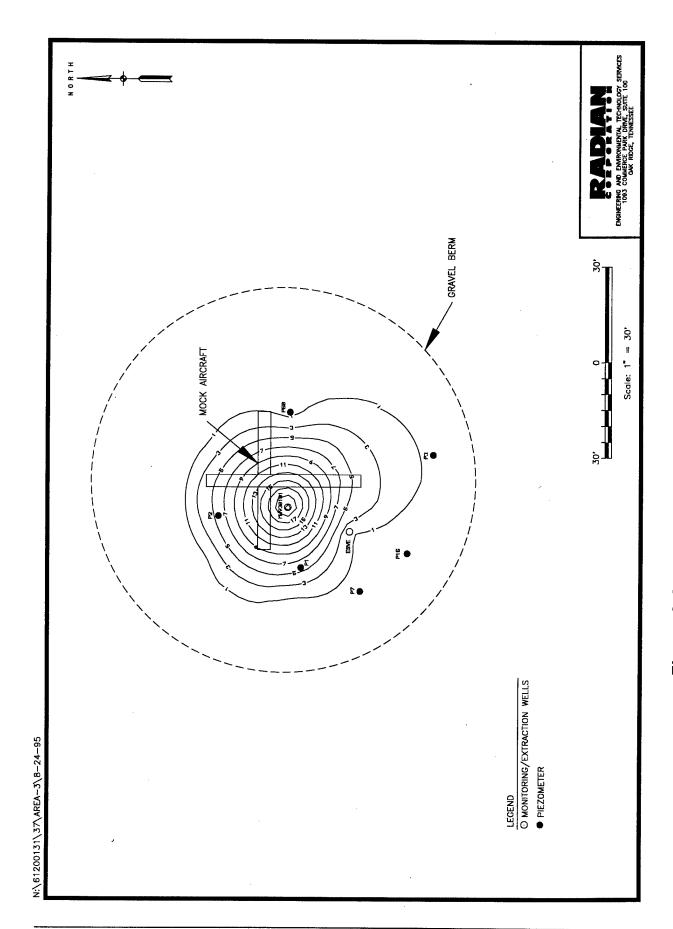


Figure 3-3. Water Level Drawdown Contours

Table 3-2
2-PhaseTM Extraction Technology Selection Criteria

Criteria Parameter	PREECA Remedy Profile Guideline	Ellsworth AFB OU-1 Profile
Contaminant	Halogenated VOCs, TPH, and/or BTEX for	Halogenated VOCs, TCE, BTEX,
	sites where expedited action is required	TPH
Contaminant location	Saturated zone or saturated and vadose zone	Saturated and vadose zone
Contaminant concentration	Significantly greater than MCLs	Significantly greater than MCLs for TCE
Depth of contamination	<150 BGS ⁸	<43 ft BGS
Henry's Law Constant of majority of contaminants	>0.01 at 20°C (dimensionless) ^b	0.297 at 20°C
Vapor pressure of majority of contaminants	>1.0 mm Hg at 20°C	58 mm Hg at 20°C
Hydraulic conductivity	<1 x 10 ⁻⁴ cm/sec (i.e., silts and clays with	9.5 x 10 ⁻⁶ cm/sec (weathered and
(saturated zone)	minimal interlayered sands and gravels)	fractured shale)
Groundwater production rate	<15 gpm for 4-in. well casing	<0.5 gpm for a 2-in. well casing
Average air permeability of vadose zone and distribution of contaminants	Groundwater Only: Case 1: No contamination present in vadose zone. Air permeability in vadose zone is not a determining factor. High concentrations (significantly higher than MCLs) of contaminants in saturated zone. Vadose Zone and Groundwater Contamination: Case 2: Low to moderate concentrations of contaminants in vadose zone. Low or high air permeability in vadose zone. High concentrations of contaminants in saturated zone. Case 3: Low or high air permeability in vadose zone. High concentrations of contaminants in vadose zone and saturated zone. Case 4: High concentrations of contaminants in vadose zone. Low air permeability in vadose zone. Low to moderate concentrations of	 Case 3: Moderate to high concentrations of VOCs in vadose zone and low to moderate concentrations of VOCs in saturated zone, and Low air permeability in the saturated zone (clays, silts, and shale), and high air permeability in vadose zone (sand, gravel, and silt).

^aTPE may be implemented at sites where depth of contamination is greater than 150 BGS; however, it has been shown to be more effective when implemented at shallower depths.

AFB = Air Force Base

BGS = Below Ground Surface

BTEX = Benzene, Toluene, Ethylbenzene, and Xylenes

MCL = Maximum Contaminant Level

OU = Operable Unit

TCE = Trichloroethylene

TPH = Total Petroleum Hydrocarbon

VOC = Volatile Organic Compound

^bDimensionless Henry's Law Constant in the form: (concentration in gas phase)/(concentration in liquid phase)

During the MW 930101 test, the groundwater flow rate surged at the beginning of the test due to evacuation of the well and sand pack, as well as from dewatering of the sand and gravel deposits near the well. After the initial surge, the water production rate continued to fall slowly during the duration of the test (Figure 3-1). The water production rate fluctuated in the range of 0.16 to 0.20 gpm during the later portion of the test, representing quasi-equilibrium conditions for the test. The radius of influence is estimated to be between 30 and 50 ft based on the available data. The drawdown cone is somewhat asymmetrical with greater drawdowns measured toward the southeast.

During the ESVE test, the groundwater flow rate fluctuated in the range of 0.04 and 0.12 gpm (Figure 3-1). Water table drawdown was also measured in the piezometers at the site during the test. However, the water table actually rose in P-1 and P-2 during test. There are two reasons for the water table rise: (1) the base of the screened interval in the ESVE well was 8.5 ft higher than in MW 930101 and (2) the water table did not have time to return to static conditions prior to starting the ESVE test. Therefore, no conclusions can be reached regarding the radius of influence from the ESVE test.

3.2.2 Vapor

The vapor flow rate was measured using rotometers located at the skid and is plotted along with the vapor flow rate on Figure 3-1 for both the MW 930101 and ESVE tests. Induced vacuum was measured in vapor probes VP-1S, VP-1D, VP-2S, VP-2D, VP-3S, VP-3D, VP-22D, VP-50D and VP-60D (Appendix C). A plot of induced vacuum for the end of the MW 930101 test is shown in Figure 3-4 for the deep vapor probes.

During the MW 930101 test, vapor flow steadily increased during the first 30 hours of the test. The flow rate then stabilized at approximately 57 scfm for the duration of the test. The increase in flow during the first 30 hours of the test occurred

as the formation was dewatered and the relative permeability to vapor increased. No aspiration air was required to lift the water from the well; therefore, all vapor flow was from the subsurface. A large pressure drop was maintained in the straw during the test indicating that a higher vapor flow rate might have been achieved in a larger diameter well. The radius of influence of the vapor is greater than 80 ft based on the available data (Figure 3-5).

During the ESVE test, very little, if any vapor flow was generated from the subsurface. Vacuum was not observed in any of the vapor probes, except for VP-3S, during ESVE test (Appendix C). Due to the low permeability of the saturated zone, aspiration air from the surface was required to lift groundwater from the well. It is likely that if the ESVE test were run for a longer time period, more vapor flow would have been generated from the subsurface as the saturated zone was dewatered.

3.3 <u>VOC Recovery</u>

Tables 3-3 and 3-4 summarize analytical results for the VOCs detected in the samples collected during the two tests. BTEX, DCE, ketones, and TCE were the primary contaminants found at the site (see Appendices D and E for the analytical laboratory results and chain-of-custody forms). Results of VOC sampling at MW 930101 included:

- The baseline concentration (before the test) of nonchlorinated VOCs in groundwater from MW 930101 was 2230 μg/L.
- The post-test concentration of nonchlorinated VOCs was 1906 μg/L.
- . The nonchlorinated VOC concentration in the extracted water (collected from knock-out pot) averaged 1027 μ g/L in the MW 930101 test.
- The total VOC concentration in extracted vapor ranged from 4200 to 7308 ppmv in the MW 930101 test.

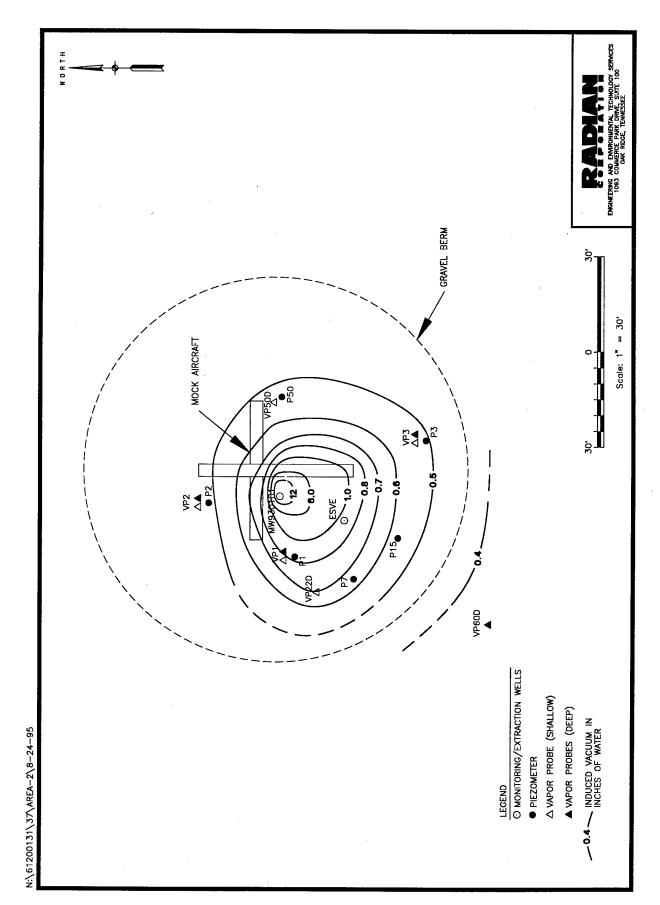
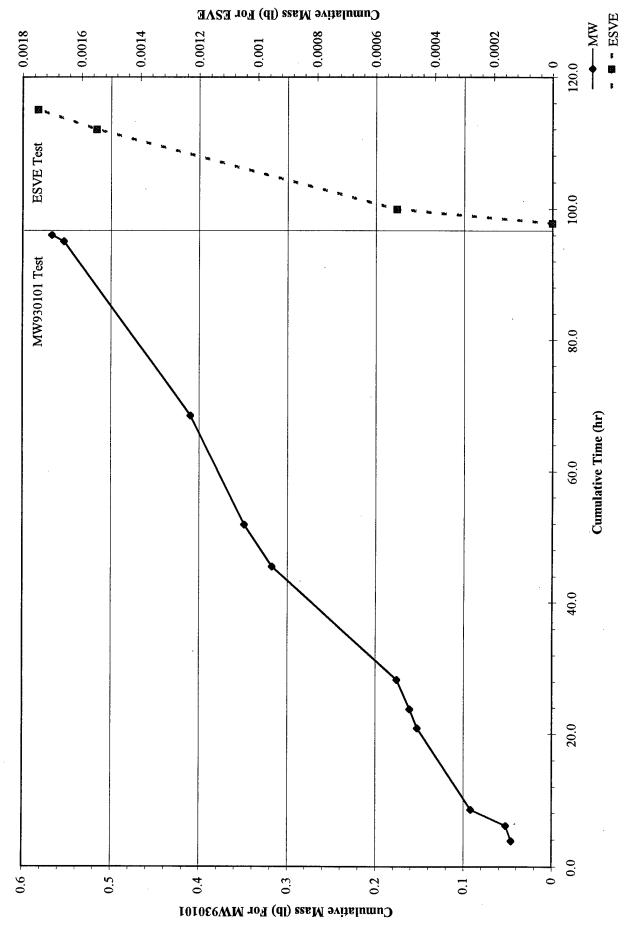


Figure 3-4. Induced Vacuum in Deep Vapor Probes at end of MW930101 Test, Ellsworth AFB

Figure 3-5. VOC Removal Over Time (Water)



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Table 3-3

Summary of Water Data Concentration in Micrograms per Liter (µg/L)

7. 7. 7. 8.							Sam	Sample ID						
			71		MW	MW 930101 Test						ESVE Well Test	/ell Test	
	MW 930101		Influent Influent	Influent	Influent	Influent	Influent	Influent	Influent	Influent	Inflight	Influent	Inducat	MW
Contaminant ²	Pre-Test	1	2	- 1		S	9	7	8	9		12		Post-Test
Acetone	380	1	240	330	400	380	490	740	730	1200	1300	57		1300
Benzene	870	220	270	360	330	270	260	290	240	320	230	7.5	1.8	470
2-Butane (MEK)	_	_	_	120	130	130	091	230	220	360	350	78	17	350
cis-1,2-Dichloroethene	3900	002	790	1000	910	790	810	810	700	098	630	63	30	1900
Ethylbenzene	280	\$9	110	130	150	130	130	130	130	160	130	11	4	56
Toluene	100	28	53	95	130	130	140	180	200	230	180	=	4.1	220
Trichloroethene	18	91	61	56	31	29	28	32	33	44	39	5	6.1	120
m- and p-Xylene	840	180	260	320	390	380	400	440	430	006	420	46	24	1000
o-Xylene	140	13	18	42	50	40	44	82	08	110	120	93	3.3	160
Total purgeable material	45,400	41,300	19,300	112,100	42,600	24,300	28,100	80,500	56,700	41,600	65,600	3,180	1,190	41,500

*Only analytes with confirmed hits above detection limits are reported.

Note: All influent samples were taken from the knock-out pot prior to carbon treatment.

Table 3-4

Concentrations in Parts per Million by Volume (ppmv) Summary of Vapor Data

		Trichloro-	Tetrachloro-)TO-	m- and p-		Unidentified CS-C10	Civil
Sample ID	Benzene	ethylene	ethylene	Ethylbenzene	Xylene	o-Xylene	Compounds	Dichloroethylene
V1 (Pre-test) ^a	38	9.0	0.04	1.3	1.1	0.4	3629	6119
V2	63	1.8	0.15	5.3	8.6	3.5	6351	71.0
V3	<i>L</i> 9	2.4	0.22	7.0	19	6.2	6933	74.0
٧4	<i>L</i> 9	2.6	0.23	7.2	22	8.9	6785	74.9
VS	73	3.0	0.26	7.3	22	6.7	7194	81.0
9/	89	3.5	0.34	8.5	28	8.7	6750	75.9
77	19	3.4	0.34	3.3	28	8.9	6255	70.2
8/	59	3.5	0.35	8.1	28	8.7	6209	68.2
V8 Duplicate	54	3.2	0.33	7.6	26	8.4	5586	63.2
6/	50	3.7	0.42	8.3	29	9.2	5386	60.1
V10	41	3.1	0.37	7.3	27	8.8	4430	47.7
V11	50	4.1	0.52	8.7	32	10	5355	57.0
VI3	38	3.6	0.52	7.3	30	10	4268	44.5
V14 (Post-test) ^a	40	3.6	0.36	4.3	91	4.7	3936	49.3
VI5 ^b	0.26	0.012	0.010	0.21	1.40	0.82	52	0.2
V15 Duplicate ^b	0.25	0.010	0.009	0.19	1.29	0.82	49	0.2
V16 ^b	0.25	900'0	0.005	0.07	80.0	0.07	6	0

BTEX and chlorinated compounds were detected in the groundwater and vapors collected from the ESVE well, but at greatly reduced concentrations compared to MW 930101. The lower concentrations were attributable to the elimination of air flow from the highly contaminated vadose zone. Results of VOC sampling at the ESVE well included:

- The concentration of nonchlorinated VOCs in the extracted water at the start of the ESVE well test was 170 μg/L.
- The concentration of nonchlorinated VOCs in the extracted water at the end of the ESVE well test was 37 μg/L.
- The total VOC concentration in the extracted vapor averaged 30 ppmv in the ESVE well test.

3.3.1 Extraction Results

Results of the MW 930101 test included:

- Approximately 428 lb of total VOC compounds were extracted from MW 930101 in 96 hours of testing (i.e., approximately 100 lb/day). The vast majority of the compounds were extracted in the vapor phase.
- Average groundwater extraction rate was 0.2 gpm.
- Average vapor extraction rate from the formation was 50 scfm.
- The TPE extraction system transferred over half of the VOCs in the groundwater to the vapor phase, resulting in decreased concentrations in the water phase and reduced treatment cost
- A small amount of product (LNAPL) was found in the first carbon drum following the test. The material was extracted from the saturated zone or capillary fringe. The presence of this product in the liquid phase likely increased the concentration of VOCs in

the effluent samples. A much greater stripping effect would have likely been measured without this product.

Results of the ESVE well test included:

- Average groundwater extraction rate was 0.1 gpm.
- Average vapor extraction rate was approximately 13 scfm.
- Approximately 0.1 lb of total VOCs were extracted from the ESVE well in 24 hours.

3.3.2 VOC Removal Over Time

Graphs showing VOC removal over time at the two test wells are provided in Figures 3-6 and 3-7. In general, steady concentrations in both extracted vapor and water were achieved after approximately 20 hours of testing. Average off-gas vapor and effluent water concentrations for the MW 930101 test were:

- 26,000 ppbv nonchlorinated VOC in extracted vapor and
- 1906 μg/L nonchlorinated compounds in extracted groundwater.

Average off-gas vapor and effluent water concentrations for the ESVE well test were:

- 2,690 ppbv nonchlorinated compounds in extracted vapor and
- 172 μg/L nonchlorinated compounds in extracted groundwater.

Ninety-three percent of the total VOCs removed were from the vapor phase and the remaining seven percent were in the water phase.

3.4 <u>Conclusions</u>

The two tests conducted at OU-1 produced very different results that provided valuable information on the application of TPE that a more traditional test would not have provided.

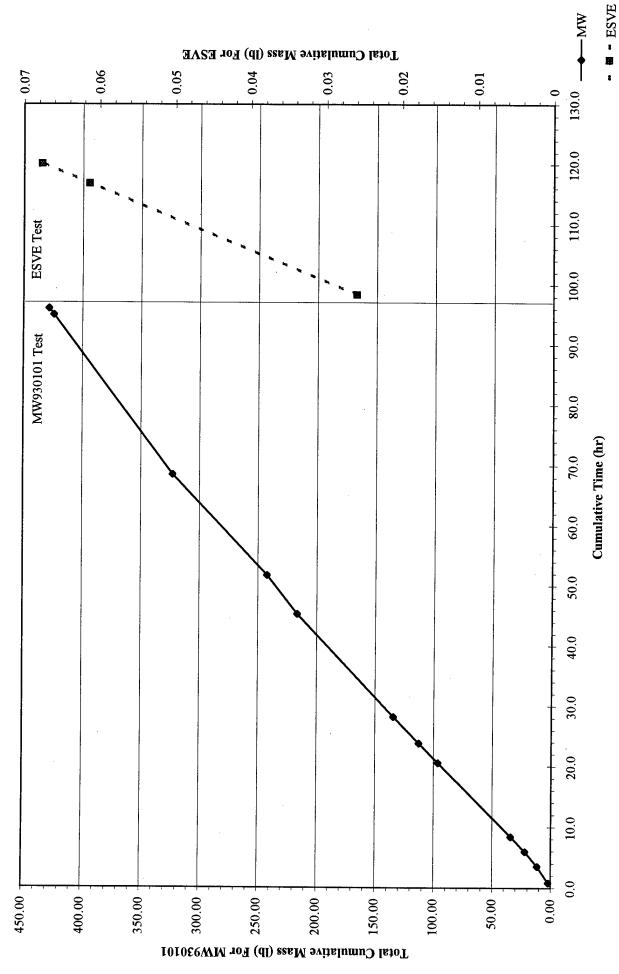
--- ESVE

MM

Cumulative Mass (lb) For ESVE

F3-6 PIL_TEST.XLS

Figure 3-7. Total Mass of VOCs Removed Over Time (water and vapor)



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F3-7 PIL_TEST.XLS

The conclusions from each test are discussed below and are followed by overall conclusions.

3.4.1 MW 930101 Test

The TPE test on MW 930101 demonstrated that TPE is effective in simultaneously removing volatile contaminants from both the vadose zone and groundwater. Approximately 100 lb/day of VOCs were removed during the 96-hour test, primarily from the unsaturated zone (vadose zone).

An average flow rate of 0.2 gpm and a drawdown of 26.4 ft were generated during the TPE test. A small amount of product was also extracted during the test. Well MW 930101 has a 32.5-ft screen that was open across both the saturated zone and vadose zone (10 to 42.5 ft BGS, water table 15.6 ft BGS) (see Figure 1-2). The saturated zone consists primarily of low permeability weathered and fractured shale (estimated hydraulic conductivity of 9.5×10^{-6} cm/s) (EA Engineering 1994a).

A vapor flow rate of up to 57 scfm was generated from the formation during the TPE test under a well head vacuum that decreased from 11 to 1 in. mercury during the test. These vapor flow rates compare with vadose zone rates of between 60 and 125 cfm generated during the previous SVE test at the site. The unsaturated zone at the site consists primarily of a mixture of sand and gravel fill of moderately high permeability (200–400 Darcy) (EA Engineering 1994b). Therefore, it appears that the majority of the vapor flow into the well was from the sand and gravel fill interval in the vadose zone.

Because of the relatively high vapor flow rates (for a TPE test), no aspiration air was required to lift the groundwater through the straw. An advantage of this is that vapor concentrations are not diluted by aspiration air. This maximizes the air flow through the formation and also increases the cost-effectiveness of vapor treatment. Performance at this high vapor flow was hindered by the 1-in. diameter straw. This caused a large head loss and concomitantly low vacuum

applied at the formation (approximately 1 in. mercury). The straw diameter was limited by the well diameter (2 in.). A larger well diameter would allow a larger straw diameter, which would increase the vacuum on the formation.

3.4.2 ESVE Well Test

The short duration (24-hour) TPE test on the ESVE well demonstrated that VOCs could be effectively removed from groundwater in the tight saturated zone formation.

The average groundwater flow rate from ESVE was approximately 0.11 gpm, although this is only 50% of the flow rate obtained from MW 930101, it is similar to flows obtained near the end of the test. Also, the well screen in the ESVE well is located entirely within the weathered and fractured shale in the saturated zone and was effectively isolated from the permeable alluvium in the vadose zone (24 to 34 ft BGS) (see Figure 1-2). The placement of the 10-ft long well screen in ESVE resulted in a drawdown of 17.9 ft. The resulting lower extraction rate in ESVE compared to MW 930101 was a result of the shorter well screen and 8.5 ft less of drawdown.

The vapor phase flow from the well was 14 scfm during the test under a well head vacuum of 20 in. mercury. The majority of vapor-phase flow was aspiration air. Aspiration air was required since vapor flow from the formation was low during the brief test period, and air was required to lift the groundwater and create two-phase flow in the straw. Air flow from the formation would likely increase, and aspiration air could be reduced over time as the system dewaters the weathered and fractured shale and flow pathways open up. The short duration of the test did not allow sufficient time to fully evaluate this effect.

The contaminant removal rate from ESVE during the short duration of the test was approximately 0.07 lb/day. Because the well screen was isolated from the vadose zone, the contaminants removed during this portion of the test were exclusively from the saturated zone.

3.4.3 Overall Conclusions

The TPE tests were conducted in extraction wells located within the former FPTA burn pit. The first portion of the test demonstrated that TPE could be used to simultaneously remove VOCs from contaminated soils in the vadose zone and VOCs from the groundwater. Some liquid hydrocarbon product was also removed. The second portion of the test demonstrated that VOCs could also be effectively removed from contaminated groundwater without pulling significant amounts of vapors from the vadose zone. It also confirmed that the majority of contamination is present in the permeable vadose zone but that there is significant recoverable contamination in the tight saturated zone. The selection of the well screen location can be used effectively as a design criterion depending upon the results desired. The low groundwater flow rate suggests that a conventional pump and treat system, which relies strictly on gravity flow, would not likely be cost-effective at recovering significant quantities of groundwater and contaminants at this site. Conventional SVE. which uses low vacuum, would likely be effective at removing a significant mass of contaminants from the gravel fill in the vadose zone. SVE would have no impact on the saturated zone contamination. Dual phase extraction, which combines SVE with pump and treat, would not likely gain much over straight SVE within the source area. However, pump and treat would likely be effective as a long-term strategy to control groundwater plume migration at the downgradient edge of the plume.

The test results suggest that a TPE removal action focused on the primary area of contamination would be very successful. A 4-in. diameter TPE well with a 10-ft screen length targeting the saturated zone and capillary fringe zone would be very effective at extracting groundwater and VOCs from the saturated zone and deep vadose zone. The capillary fringe is an area between the water table and unsaturated zone, which is frequently a major source of contamination when fuel hydrocarbons are

present. This would likely achieve a significant mass removal rate of subsurface VOCs, especially in the source area.

It also suggests that at this site TPE plus SVE, or a nested TPE system (multiple wells screened in different zones), could remove significant quantities of VOCs from both the saturated zone and the vadose zone, which is also contaminated and is more permeable. This combination of technologies would be most likely to achieve overall remedial goals for the site.

4.0 ELLSWORTH AFB REMEDIAL ACTION ENHANCEMENT

Ellsworth AFB has produced the feasibility study for OU-1 and is in the process of drafting the proposed plan and record of decision. Ellsworth AFB has chosen to implement early actions to address the further spread of contamination from this operable unit. The preferred alternative for an interim remedial action (IRA) at this site is a combination of SVE and groundwater pump and treat, or dual-phase extraction. This was based on the data from two previous treatability studies; a groundwater recovery and treatment system (1990) and an SVE pilot test (1994).

The Final Remedial Design Work Plan for OU-1 states that the overall strategy at the site is to control migration of contamination in the former FPTA and extract and treat contaminated groundwater downgradient of this source.

The IRA currently under design for OU-1 combines SVE and groundwater pump and treat. The groundwater component of this system is primarily focused on migration control by intercepting the plume downgradient of the source area. The SVE component will address much of the vadose zone contamination, particularly in the gravel fill placed in the fire pit area when the FPTA was constructed. The Final Design Analysis for OU-1 states that the system will not address portions of the deep vadose zone in native silty clay, which underlies the fill and may be a significant source of vadose zone and groundwater contamination.

A TPE system could enhance the IRA at OU-1 by addressing the groundwater and contaminated capillary fringe zone as well as the deep vadose zone in the source area on an accelerated schedule. The combination of the current IRA and the TPE system could have a synergistic effect that would likely be more effective than either could accomplish alone. The pump and treat should be effective in controlling downgradient migration, and the SVE should be effective in remediating the shallow vadose zone. Also, an SVE well can cause upwelling of the

groundwater around the well. The upwelling can interfere with the SVE in some cases, particularly if the well screen is close to the water table, as is the case at this site. With TPE operating concurrently, it will draw the water table down from the SVE wells. This would dewater and expose more sediments to air, increasing the efficiency of the SVE.

The TPE system would likely achieve a much higher mass removal of contaminants in the fractured and weathered shale (saturated zone) than could be achieved with a conventional pump and treat system. It would also remove any product floating on the water table or trapped in the capillary fringe. It could remediate the shale and deep overlying material in a relatively short period of time. This would remove the source of groundwater contamination and reduce the length of time required to operate the IRA for migration control in the downgradient plume area. The TPE system could be operated concurrently with the IRA.

The cost of a full-scale TPE system (50 hp) at the FPTA is estimated at \$100,000-\$200,000 capital cost. At OU-1, the IRA includes both water and vapor collection and treatment systems. If the TPE was added on to this system, significant cost savings would result, and a cost in the low end of that range would be likely. Similarly, if TPE is added on to the existing system, operation and maintenance (O&M) cost would be limited primarily to the cost of electric power. The estimated O&M cost for such a system is \$30,000-\$50,000 per year. This cost would likely only be incurred for 1 to 5 years, as opposed to typically 30 years for a pump and treat system.

As suggested by the EPA, the TPE technology may be useful for removal actions and remedial actions at other locations at Ellsworth AFB where site characteristics match the requirements for the technology (Appendix A). The PREECA TPE selection criteria shown in Table 3-2 fit the profile of many sites at Ellsworth AFB. In particular, OU-11 (Basewide Groundwater) contains several sites with fuel hydrocarbon

and/or chlorinated VOCs. Site BG-04 (Hooterville Plume), which has a significant TCE plume extending off base, and OU-9 (Auto Hobby Shop), which contains fuel hydrocarbons in groundwater, may also be good candidate sites.

5.0 REFERENCES

EA Engineering and Science, Inc., 1994a. Draft—Final Remedial Investigation Report, Operable Unit 01 at Ellsworth AFB, South Dakota, September.

EA Engineering and Science, Inc., 1994b. Treatability Studies, Technical Report, Operable Units 1, 2, and 4 Ellsworth AFB, South Dakota, November.

Rust Environmental, Inc., 1995. Final Remedial Design Work Plan, Interim Remedial Actions, Operable Units 1 and 4 Early Actions, JP-4 Plume and FRA, Ellsworth AFB, South Dakota, March.

Rust Environmental, Inc., 1995. Final Design Analysis, Interim Remedial Actions - Operable Unit 1 and 4, Early Actions, JP-4 Plume and Flightline Refueling Area, Ellsworth AFB, South Dakota, March.

U.S. Air Force, 1995. United States Air Force Presumptive Remedy Engineering Evaluation/Cost Analysis (PREECA), Final, 5 May.

Radian Corporation, 1995. Ellsworth AFB Operable Unit 1 2-Phase[™] Vacuum Extraction Pilot-Scale Test Work Plan, Ellsworth AFB, South Dakota, June.

APPENDIX A

EPA Region VIII Letter



UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

REGION VIII

999 18th STREET - SUITE 500 DENVER, COLORADO 80202-2466

Post-It* Fax Note 7671	Date # of pages
TO BILL BUCHANS	From D. PETERIEU
Co./Dept,	Co.
Phone #	Phone # 605-385-2675
Fax # 615-483-9061	Fax # 605-385-6619

Ref: 8HWM-FF

Dell S. Peterson, PE
Installation Restoration Program Manager
28 SPTG/DEVR
Ellsworth AFB, South Dakota 57706-5000

Subject:

Presumptive Remedy Engineering Evaluation/Cost Analysis (PREECA);

2-Phase Pilot Test Evaluation Report.

Dear Mr. Peterson:

The presentation at your offices regarding the PREBCA document and the 2-Phase extraction technology was very informative. I have read both documents and offer some comments for the use of the documents at Ellsworth.

PREECA

- 1. This "plug-in" EE/CA would be a useful tool for accelerating non-time-critical (NTC) removal actions at Ellsworth Air Force Base. Certain sites within and outside of designated operable units (ous) may qualify for a removal action. For sites within OUs, we must evaluate the time savings of performing removal actions given the current stage of the remedial action.
- 2. The technologies listed in the PREECA document may be applicable to certain sites at Ellsworth. As stated during the presentation, we still must evaluate the cost effectiveness of the technologies based on the site-specific characteristics (e.g. capital costs vs. operation and maintenance costs).
- 3. The "plug-in" approach and the PREECA document would be useful for non-CERCLA cleanup actions. Underground Storage Tank (UST) cleanup projects across the Base seem to have similar cleanup characteristics. The PREECA document could be used as the technology evaluation document (i.e. Corrective Measures Study Report), with the State's approval, to make remedy selection decisions.
- 4. The phrase "relative risk" is used throughout the document. Using the word "relative" may be misleading. At a facility with all low risk sites, one of the sites at the facility will have high risk, relative to the other sites, but still be low risk.

APPENDIX B

Well and Drilling Logs

GROUNDWATER SAMPLING RECORD

Page 1 of 1

WELL NUMBER: MW 9301 DATE/TIME: 6-23-95 PROJECT/NUMBER: 612-001	1 1120	7			Sunny Windy 60°F Gary Dylce	
PIELD MEASUREMENTS OVM/OVA READING (ppm): DEPTH TO WATER BELOW TO WATER COLUMN HEIGHT (ft.): CASING DIAMETER (ft.): ACTUAL PURGE VOLUME (ga	0C (ft.):_):_745 2" = 0 al.):	2 .167ft		WELL DEPTH PRODUCT TO 3 WELL VOL DECONTAME	EPTH (ft.): H BELOW TOC (ft.): 43.9 HICKNESS (ft.): LUMES (gal.): 46 INATION METHOD: Alcanor Thed weder	
PURGING MEASUREMENT MI PURGING METHOD:	Bailer:	- PVC	 Stainless Steel	Teflon	•	
SAMPLING METHOD:	Pump: Bailer: Pump:	□ Submersible □ PVC □ Bladder	Bladder Stainless Steel	□ Teflon	□ Polyethylene	

Time 23 Jun 48	Cumulative Volume Purged (gal.)	pН	Conductivity (µmhos/cm)	T (•F)	Comments (Water clarity, odor, well conditions, etc.)
0959	th. 2.5	7	1.00	52°	Gray /Silt present
1015	~7.5	7.5	1.06	56°	Stockety Gray
10 30	~14.501	7.5	1.00	540	Gray Step to allow well to recharge
1000	~ V7.0	7.5	1.00	56.	Gray + Silty
1115	~22.0	7.5	1.00	540	Gray + Silty

SAMPLES

ID#	Matrix	Container	Preservative	Filtered (Y/N)	Analytical Method
MW930101	Water	3×40ml 6A	uci ·	N	8260
					·

COMMENTS / CALCULATIONS 3 Well Volumes (gal)

3 Well Volumes (gal) = 3π (radius in ft)² (water column height in ft) (7.48 gal/ft³water)

SAMPLERS SIGNATURE(S)

	DRILLING LOG										13248		
1. COMPANY	Y MUHE	adian		2. 9	RILLING SL	SUBCONTRACTOR Huntingdim					SHEET		7
3. PROJECT	5. MOJECI ENSWOrth AFB Z-Phase Test						4. LOCATION OU-						
5. NAME OF	5. NAME OF DRILLER Kenneth Diers						ACTURER'S DES	SIGNATIO	N OF DRILL CA	45			
AND SAL	NO TYPES OF MPLING EQUIP	ORILLING MENT				8. HOLE	LOCATION (C)	4	from MI				7
	h I P					9. SURFA	CE ELEVATION	•					1
2 tt'	21/2-in 1	ore arrel				10. DATE	STARTED 4	- ZO -	95	11. DATE COMP	^{μπο} 6	- 20 -95	
12. OVERBU	IRDEN THICK	EZZ				15. DEPTI	H GROUNDWATE		INTEREN	ff		•	1
13. DEPTH	ORHLED INTO	ROCK	· · · · · · · · · · · · · · · · · · ·			16. DEPTI	H TO WATER A	HO ELAP	SED TIME AFTER		ETED		1
1	DEPTH OF HO	45 II				17. OTHE	R WATER LEVEL	MEASU	REMENTS (SPECIF	۲)			-
18. GEOTEC	HNICAL SAMP	us None	DISTURBED	UNO	HSTURBED		19. TOTAL NUM	BER OF	CORE BOXEZ VI	one	***	*	-
	S FOR CHEM	CAL ANALTSES	voc	METALS		OTHER	(SPECITY)	опн	ER (SPECITY)	OTHER (SP	ECEY)	21. TOTAL CORE	-
	Non											RECOVERY Z	
	Tot or Hou	Prezoneta	BACKFILLED	MONITORING	M.T.	OTHER	(\$25(0)	23. Si	HATURE OF INSP	PECTOR			
GRAPHIC	THU WS	ritumena			nn s	CREENING	GEOTECH S	11401 C	Lough				4
roc	DEPTH 5		SCRIPTION OF MATERIALS			q SUF12	OR CORE B		INTERVAL	RECOVERY		REMARKS h	
	1 2 3 4 5 6 7 8 9 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	Olive BIK P (54 2/1) Cla Ver Brown (10 y Yellowish Sandy Gra Sorted, sat 3-indres in silty and cl (aliche (no	Clayery Gravel. overly sorted, loo proved at FTA Hey gravel, Strol J poorly sorted, to Dai orange (10 yre usel, very poorl pround pebble diameter, or ayey (<10://), t remented), li ated and day	sc, surface Ky, dry K odor: 6/6) -ly coassionally coassionally	SA=				5-10 ft	2.4 ft	B 7	= 0 2= 2= 3	muhankaakaalaakaalaakaalaaka >
0 0	10 —	PROJ	ECT (17 CV) -							HOLE NO.	<u></u>		E 10
		I	FC 615-001-3	31-37						1	P-1		

SA = Sample screen with PID in ppm.
BZ = Breathing zone PID recesarement in ppm.

		DRILLING	LOG				HOLE NO.	1324
PROJECT E	Ellswor	th AFB 2-Phase Test INSPE	ctor Gan	Dyke			SHEET SHEETS 3	
GRAPHIC LOG 3	ОЕРТН 5	DESCRIPTION OF MATERIALS c	FIELD SCREENING RESULTS d	GEOTECH SAMPLE OR CORE BOX NO.	SAMPLE INTERVAL I	RECOVERY	REMARKS h	
0	10 =	Asabove Sandy gravel						E
0	11-	3 3			10-15 H.	Z.4 ft.	BZ=3	E
6	=		SA=3				06.2	E
00	12-							
) D	13-							E
000								E
0 0	14-							E
ه ن								E
	15-	Brown to Black, Claver around						- <u>E</u> - 1
- 0-1	15 =	Brown to Block, Clayey grand, poorly sorted, subround pebbles, moist.			15-20 ft.	7.6 St		E
U-0						2.0 11.		E
	17.	Light Olive Gray, clay (weatherd						E
 	=	Pierre State) mottle color with some yellow brown, sticky,						E
	18-	moist. (54 5/2)						E
	Ι., Ξ							F
	19							
	20-							F.
	\Box	Light Olive Gray, fractured					BZ = 3	- E-2
-2-2-	21-	land clay of deal 32 Charles	SA=30		Z0-Z5A	4.0 ft.	125 2	E
	Ι. Ξ	Silty clay, grades in b untract- wred clay at about 22 ft, mottled colony with dark yellow urange (10 4R 6/6) and greenish black (56 21), where un froctored						E
	22	(10 YR 6/6) and Greenish Hack						
	23	(56 2/1), where un froduced	-					E
		it is sticky and cohesive, minor amounts of five sand						F
- ==-	24-	and sitt.						E
==	目							E
==	25	•					<u> </u>	-
===	26—				Cuttings			E
===	20-				04/7			E
=	27—							E
==	\exists	(201)						E
==	28 —	Same to 30 ft				HOLE NO.		<u> </u>
E	zq -	612-001-31-37				P) -	
==	<u> </u>	- ,				•		
==	36							[30

		DRILLING	LOG				ноих но. P-1	134
PROJECT 6	Ellswo	rth Z-Phase test ""	SPECTOR Gary [كإلاف			SHEET SHEETS 3	7
GRAPHIC LOG 0	DEPTH b	DESCRIPTION OF MATERIALS	4 SEZITIZ LIETD SCISEDHING	GEOTECH SAMPLE OR CORE BOX NO.	SAMPLE INTERVAL (RECOVERY	REMARKS	
<u> </u>	30 =	As above. Sitty clay. Fractured		3U-35 ft.	3.9 ft.		BZ=3	==3(
	11-	from 30 to 21.6	SA=2		>	*		
==	12-	·		•				上
===	13-							E
- <u></u> -	' =							E
	14-							
	35 = 25 =							<u>E</u> _3
					Cuttings		BZ = 3	E
<u> </u>	16				only		b c = 3	
	17-							
	18-							E
								Ш
	19-					:		
===	40 =		-					E 4
	, =	Greenish black (56 Z/1), clay to slightly weathered Prerne Shale.		! !	(u Hwgs			E
	21-	Shale.			6/10			E
	22-							E
	23-							E
				:				E
	24							E
///	45 - 25 -	Total Depth=45ft.						<u>E</u> 4
	26-						,	E
	37							E
	27							E
	28 —	PROJECT	<u> </u>	<u> </u>		HOLE NO.		<u> </u>
		612-001-31-37					P-1	

SINGLE COMPLETION WI	Well Number P-
CONSTRUCTION LOG	•
Project Ellsworth Z- Phase Test Location UU-1	Project Number 612-001-31-37
	Datum
Top of Casing Elevation	Ground Surface Elevation
<u> </u>	BORING
Å L	A. Total Depth (ft)
	G B. Boring Diameter (in.)
	Drilling Method 4 SA
	WELL CONSTRUCTION
	C. Casing Length (ft)
	Type Z-INCh PYC
	D. Casing Diameter (ft)
E	H E. Depth to Top of Slotted Interval (ft)
-a-	F. Perforated Casing Length (ft) 70'
	Perforated Interval From 44 to 24
	Perforation Type Solled U.UIO
	Perforation Size O.01
A I I C	G. Surface Grout Interval (ft)
	Grout Material
	H. Backfilled Interval (ft)
	Backfill Material MA
	I. Sealed Interval (ft)
	Seal Material Barbarite Grander
	J. Filter Pack Interval (ft) 45-21-
F-70	J Pack Material U(10)
	K. Bottom Seal Interval (ft)
	Seal Material
	L. Depth to Top of Casing (in)
	M. Protective Casing Diameter (in)
- CAP:	F + 1012 bass salica sd.
	K
- B -	

	DRII	LING	LOG					HOLE NO.	-Z
1. COMPANY MANE RACION SMEET 1 OF 1 SMEETS									
3. PROJECT Ellswith AFT	3 7-Phase	Test	4 1001	4 DOCATION ON-1					
5. NAME OF DRILLER Kenneth			6. HAVE	FACTURER'S DE		OF DRILL	nE		
7. SIZES AND TYPES OF DRILLING AND SAMPLING EQUIPMENT			8. HOLE	LOCATION Z	5 fl.	trom 1	1W9301	٥١	
74"00,3"ID PHAS HSA: 5 H			9. SURF	ACE ELEVATION		.,,,,,			
split con Darrel			10. DAT	E STARTED 6-	70-9	<	11. DATE COMPL	mo6-20-	95
12. OVERBURDEN THICKNESS				TH GROUNDWATE					
13. DEPTH DRILLED INTO ROCK			16. DEP	TH TO WATER A	HO ELAP	SED TIME AFTER	DRILLING COMPLI	ETEO	
14. TOTAL DEPTH OF HOLE 45 ft.			17. OTH	ER WATER LEVE	L WEASUI	REMENTS (SPECIF	Υ)		-
18. GEOTECHNICAL SAMPLES NUM-Q.	DISTURBED	UNO	STURBED	19. TOTAL NUM	BER OF	CORE BOXES			
20. SAMPLES FOR CHEMICAL ANALYSIS	voc	WETALS	OTHE	R (SPECOTY)	отн	ER (SPECIFY)	OTHER (SP		TAL CORE
NONE								RE	COVERY
22. DEPOSITION OF HOLE	BACKFILLED	MONITORING W	NELL OTHE	JR (S2EC37Y)	23. SIC	CHAPTURE OF INSP	ECTOR		· · · · · · · · · · · · · · · · · · ·
Completed as Piezometer			FIELD SCREENIN	с ссотсн		SUPPLY		<u></u>	
	CRIPTION OF MATERIALS		RESULTS 4	OR CORE I		INTERVAL	RECOVERY 9	REMAI h	
Sandy clay, oline Gray, conditions conditions round, odo Brown, clay	(5472) to br (obbles, ador, or frable, party sor 152) sandy ch vertile + colorles meter, potates an inc.	voist tech ing with to r sub-	SA= 720			0-5A	7.5 A.	BZ=3	
7 Olin gray to moderate to	, subrounded, me , fine, ; well sorted, mo	sist toury silty sand	2,					BZ= 7	
PROJ	ECT 61Z-001-	2127				•	HOLE NO.	P-Z_	

SA= Sample screen with PID in ppm. BZ= Brathing zone sample with PID in ppm.

		DRILLING	LOG				HOLE NO. P-Z	1348
PROJECT E	llswo	rth AFB 2-Phase Test INSPE	COOR Gary 1) ke			SHEET OF Z SHEETS 3	
GRAPHIC LOG a	DEPTH b	DESCRIPTION OF MATERIALS c	FIELD SCREENING RESULTS d	GEOTECH SAMPLE OR CORE BOX NO.	SAMPLE INTERVAL I	RECOVERY 9	REMARKS	
000 000 000	10 11	Brown, surely gravel, poorly Sorted, medium to course sand, Sub-rounded, numerous publics	SA=160		いらみ	1.4 ff.	B7=4	
0 0 0 0 0 0	12	and abbles, slight odor						
0000	14	wet at 15 ft.						
00	16-	Roduct-like sheen, stronguder	SA=400		15-20	1.0 ft.		
000	17—						Bz = 6-10	
	19—					•	Weathered Stale contact based on drilling.	
	21	Light olive gray, clay (weathered Preme Stale), whichle, mottled (during, brown & gray & divegray,		SA= 96	ZU-Z5	3.0 H.	BZ=3-6	
	22	Archy.						
	24			·				
記号	26—	(are barnel wet		SA= 184	25-30	Zoft.	B2=1-10	
	27 — — — 28 —	PROJECT						
	z9 -	612-001-31-37				HOLE NO.	P-Z	

PROJECT	E Ilen	DRILLING with 2-Phase Test 1		Dyke			SHEET 2	\dashv
	1130	20170 2 11002 1151"	TIELD SCREDUNG	GEOTTCH SWIPLE	SAMPLE		of 3 sheets 3	-
CELAPHIC LOG	DEPTH t	DESCRIPTION OF MATERIALS	SEZATIZ	OR CORE BOX NO.	INTERVAL	RECOVERY 9	SXRAW36	
	超 =	Olive gray clay. As, above.			No			
		5 m 3, 2 5 5 6 7 m	1		Sample			
	' ' <u>'</u>							
	=							
	12-							
	13-							
	=							
	14-							İ
] =							
	35 = 13-							
	_				NoSando		BZ= 3-7	
	16-							
	=						ł	
	17-							
	=		ļ				i	
-==	18-						į	
===								
	19-							
-=-								
<u>===</u>	<u></u> 28_=				<u> </u>			
111	=	Dack Greenish Black (56 2/1)	SA=10		40-45	5.0 ft.	2-	
	21_	state Fractured along bracking planes, dry, friable.	3/,-(0			0.011	BZ-Z	
111	1 - =	planes, dry, friable.						
111	22							
11/	=							
111	23-	1					l	
11/1		‡						
111]_, =	•						
11/	24-							
<u> </u>	45 =	1			İ		BZ=Z	
	T	Total Depth 45ft.					1	
	26-	·			1			
	=			1				
	27—	3						
	=	†						
	28 -			<u> </u>				
		PROJECT				HOLE HO.)-2_	

SINGLE COMPLETION WELL	
CONSTRUCTION LOG	Well Number P7
Project Elicaruth AFB 2-P. Tot	612-001-31-37
Location OUT	Project Number 6(2-00(-3(-37) Datum Ground Sourfelese
Top of Casing Elevation	Ground Surface Elevation
→ M →	BORING
	A. Total Depth (ft) 45
G	B. Boring Diameter (in.) 7 ⁴ / ₂
	Drilling Method 45A
	. WELL CONSTRUCTION
	C. Casing Length (ft)
	Type
	D. Casing Diameter (ft) 2-No
E H	E. Depth to Top of Slotted Interval (ft) 24.5
-D-	F. Perforated Casing Length (ft) Zo H
l lest	Perforated Interval From 44.5 to 24.5 ft
Marker Leed	Perforation Type Sto Heal
Jb.0 <u>*</u>	Perforation Size O.01"
A C to det	G. Surface Grout Interval (ft)
who will	Grout Material
	H. Backfilled Interval (ft)
	Backfill Material
	I. Sealed Interval (ft) 72.3 -19.2
	Seal Material Bentowite Pellet
	J. Filter Pack Interval (ft) 45-22.3
	Pack Material 10/70 Silica Sch
	K. Bottom Seal Interval (ft)
	Seal Material
	L. Depth to Top of Casing (in)
	M. Protective Casing Diameter (in)
	a bac Sand
↓ K	4 boss Sand 7/3 budget of pellets
— B — →	"13 DAGGET OT PULLETS

Ellsu	Kenneth i	-p. vase Test	2. Di	PILLING SI	UBCONTRACT	-^^					[~] P-3		
DRILLER OFFICE OF	10rth Z-Ph Kenneth I	vase Test			SUBCONTRUCTOR HUNTANGEDON)					SHEET	1 1 shoots 2		
DRILLER O TYPES OF	Kenneth i		3. PROJECT Ellsworth Z-Phase Test					TOUTION ()(Y -)					
) TYPES OF PUNG EQUIP		5. HAME OF DRILLER Kenneth Diers					SIGNATION	OF DRILL (M	F				
>,3"					8. HOLE L	CATION 5	111		1w9301	ما	·		
					9. SURFACI	E ELEVATION	0 11	110111	(40 1301				
5A g	Spilt				10. DATE S	STARTED	6-21	-95	11. DATE COMPL	ETTO 6	-21-95		
DEN THICK	X223H				15. DEPTH	GROUNDWATE							
RILLED INTO	POCK		•		16. DEPTH	TO WATER A	ND ELAP	SED TIME AFTER	DRILLING COMPLE	TED			
	45 ++				17. OTHER	WATER LEVEL	L WEASU	REMENTS (SPECIF	Υ)				
NICAL SAMP	TES A KINNE	DISTURBED	UNO	XSTURBED	15	. TOTAL MUN	BER OF	CORE BOXES					
		voc .	WETALS		OTHER ((SPECIFY)	отн	ER (SPECIFY)	OTHER (SPE	(CTY)	21. TOTAL CORE		
											RECOVERY %		
	=	BACKFILLED	MONITORING	WELL	OTHER (SPECIFY)	23. SiG	WATURE OF MISE	ECTOR				
ed as	Piezometer					1		aug	<u> </u>				
DEPTH 5		c						SAMPLY INTERVAL	RECOVERY		REMARKS h		
	L1 Ulive Gro	ug + Black mod	ttled sw.la.										
1-	(SYRUH)	Morst striky		 				0-2H	2.5 A.	B	7=2		
Ξ	Gray (54- / 6	oards surted with	fine to	SA=	ZO								
2	Sandy a	cy, mothled, de-cy	nda.c.										
_	İ							ı			(
) 	Crassing 9	raired)									l		
4-													
	Mr. L.	. h =	اه. سن										
5—	pendy sorted	, sitty sub anc to	shound,										
П	Park greenist	ngray (564 41	1) to		, ,			5-10 #	1.9 ft				
6-	ligh other bo	# (317 12) wwo	₹ 86.	54=	66			'\-		(3:	z = 2		
, =	Sand 15 mag	down to course	anda										
7=	SUFFER ISLIGH	ashing an emphasi	سلأما										
8=	is well early	monsoilidahed;	sithsand		\						į		
∃	silt, louse	and moist	w·+~\)								
9-								•					
$\begin{bmatrix} 1 \\ 0 \end{bmatrix}$													
10 7	PROJ	tcr / -							HOLE NO				
	PTH OF HE HICAL SAMP FOR CHEM NO HOLD OF HOLD	PTH OF HOLE 45 ft HICAL SAMPLES ALUNA FOR CHEWICAL ANALYSE NOWE HOF HOLE LA CHIVE Grow THE Brown 1 H. Ulive from Sorred Crossobly of 1 Crossobly of	PITH OF HOLE 45 ft HICLI SUMPLES ALUNA FOR CHENICAL MALITSE NOWLE HI OF HOLE BLOCKPILLED HOS FICTOMETET CH. CHILLY Group + Block most Listing of the Block most Listing of the Block most Listing of the Block with Toray (cy poorts street with 2 52) Sandy clay, mothed, dry The Black clayers From Sundy clay, mothed, dry The Black clayers From Sundy clay, mothed, dry The Black clayers The Clive to Black c	PTH OF HOLE 45 ft NCAL SUMPLES AUNA FOR CHENICAL ANALYSIS NOWL H OF HOLE DESTRIPTION OF WATERIALS LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE LATT AND STANLY LICYTE STA	PITH OF HOLE 45 ft NOWLE STUPES AND DISTURBED UNDISTURBED FOR CHEMICAL ANALYSE NOWLE BACKFILLED MONITORING WELL THE AS PICTOMETER DESCRIPTION OF MILERIALS FILLD PORT STOWN CLCAY, SIGHTY Sandy, LISTRATO POORLY STOWN SINCE OF SANDY CLCY, MOTHER JAMES, AND AND TO SUBJECT OF SUBJEC	PITH OF MOLE 45 ft INCLI SAMPLES AND DISTURBED UNDISTURBED FOR CHEMICAL MALTISE NO HOLE BLOCKPILLED NOMITORING WELL OTHER ! AND BROWN CLARY, SINGHTY SANDY, LISTER HAY CORST STREET STREET BY BROWN CLARY, SINGHTY SANDY, LISTER HAY CORST STREET WITH FIRE TO STREET 15Y POORTS STREET WITH FIRE TO POORTS STREET WITH FIRE TO STREET 15Y POORTS STREET WITH FIRE TO STREET 15Y POORTS STREET WITH FIRE TO TO BLOCK LICY OF STREET TO BLOCK STREET WITH SANDY SANDY STREET SOUND STREET TO CONTROL OF STREET TO CONTROL OF STREET TO CONTROL OF STREET TO CONTROL OF STREET STREET SOUND STREET TO CONTROL OF STREET STREET STREET SA = 660 PROJECT PROJ	PTH OF HOLE 45 ft NOTICE SUMPLES ALONA FOR CHEMICAL AMATES NO HOLE FOR CHEMICAL AMATES NO HOLE NO HOLE BACKFILLED NOMITORING WELL OTHER (SPECTY) CO AS PICZOTRETCT DEPTH DESCRIPTION OF MATERIALS FIELD SCREENING COUTCH RESULTS OR CORE FIELD SCREENING COUTCH RESULTS OR CORE FIELD SCREENING OR CORE RESULTS OR CORE FIELD SCREENING OR CORE RESULTS OR CORE FIELD SCREENING OR CORE RESULTS OR CORE FIELD SCREENING OR CORE RESULTS OR CORE FIELD SCREENING OR CORE RESULTS OR CORE RESULTS OR CORE FIELD SCREENING OR CORE RESULTS RESULTS RESULTS OR CORE RESULTS RESULTS OR CORE RESULTS RESULTS RESULTS OR CORE RESULTS RESULTS RESULTS OR CORE RESULT	PTH OF HOLE 45 ft ICLI SUMPLES NOUND DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE THE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE THE DESCRIPTION OF MATERILLE THE DESCRIPTION OF MATERILLE DESCRIPTION OF MATERILLE PORTS SUPPLIED SA = ZO Office Spread to brown sends of Subpread The Description of Materille The Description of Materille THE DESCRIPTION OF MATERILLE THE DESCRIPTION OF MATERILLE SA = ZO Office Spread to brown sends of Subpread The Description of Materille The Description of Mater	THE OF HOLE 45 ft 17. OTHER WATER LEVEL MEASUREMENTS (SPECITY) NOTION OF HOLE SUPPLIES NOTION OF HOLE SUPPLIES NOTION OF HOLE SUPPLIES NOTION OF HITELES NOTION	THE OF MOLE 45 \$\frac{1}{45} \text{ Control of Sections} 17. OTHER WATER LEVEL MEASUREMENTS (SPECITY) 17. OTHER WATER LEVEL MEASUREMENTS (SPECITY) 17. OTHER WATER LEVEL MEASUREMENTS (SPECITY) 17. OTHER WATER LEVEL MEASUREMENTS (SPECITY) 17. OTHER WATER LEVEL MEASUREMENTS 17.	THE OF MALE 45 \$\frac{1}{45}\$\$ \$\frac{1}{15}\$\$ OSTREED UNDSTREED 19. TOTAL MUNBER OF CORE BOXES FOR CREMEN MUSTISS 190 HEALS OF CORE BOXES FOR CREMEN MUSTISS 190 HEALS OF CORE BOXES NOWLE SECRETION OF MITCHES 190 HEALS SECRETION OFFICE (SPECTY) DEPTH OR MORE GENERAL CONTROL MUSTISM OF CORE BOXES ALCOHOLOUS STREED WOMTORNO WILL OFFICE SECRETION OF CORTICH SAMPLE SAMP		

		HOLE NO. P-3	1334					
OJECT (Ellswo	An 2-Phase Test INSM	ECTOR Gary	Dyke			SHEET of Z SHEETS 3	7
LOG d	DEPTH b	DESCRIPTION OF MATERIALS c	FIELD SCREENING RESULTS d	GEOTECH SAMPLE OR CORE BOX HO.	SAMPLE INTERVAL I	RECOVERY	REMARKS	
<u>. :</u> ا	10 =	As above				17.0		
, c	11-	Light Oline group (57 5/2) byellow brown, growel, very poorly scrted. Silty and scendy, have, peoples by 3-mach aromator	SA= Z4-		W-15 H.	1.6 ft.	BZ=Z	E
, ,		order hard such sected.	SK= C4					=
0 0	12-	3-nch wantefor	:					
0 U								E
٠. ا	13-							E
ا د د	=			}				E
′, °,	14-							=
ر د ر	=							E
	15-	BOOKIND, LOOKS sand marterite to						<u>-</u>
		Block (M), loans sand, moderate to well sorted, stained, wet	SA= 390		20-25	4.844.	BZ=4	E
	16-	Light Chica mon (57 812)			H02-51		52 7	E
	=	weathered Prere Shate, mottled						E
	'/=	weathered Preve Shate, mottled coloring from It where gray to dark.	1			<u> </u>		
	=	dire, variably traduced						E
	18-							F
	19							E
	' =							E
	20-							E
	=)			F
	21-				(uttings		1	E
					osly.			F
 	22-							E
	=							F
	23-							E
	=							E
	24-							E
	=					1		E
===	25						·	
==	=							E
===	26-							
===								E
==	27—							
<u> </u>	28	To 30f4.						E
===	<u> </u>	PROJECT	<u> </u>		I	HOLE NO.		
==	Z9 _	612-001-31-37				P	-3	
===								

			HOLE NO. P-3	135				
PROJECT	Nswar	th 2-Phase Test Insm	CTOR Gary D	1ke			SHEET SHEETS 3	
GRAPHIC LOG 3	DEPTH 5	DESCRIPTION OF MATERIALS c	FIELD SCREENING RESULTS 4	GEOTECH SAMPLE OR CORE BOX NO.	SAMPLE INTERVAL I	SECONELLA SECONELLA	REMARKS h	30
	12	As above. Light olive gray clay. Factured from 30 to 32.5	5A=3		30·3 5	5 fl.	BZ=2	
	35	·			35-40	514.		35
	16	Tools On the Control of the Control						الساساسال
	19	Park Olive Gray <u>state</u> . Pierre state, friable, no odor Upper 14. is ust with fractures along broading planes.			40-45	4.5 A.	BZ=Z_	
	21—							
	24— 45— 25—							
	26————————————————————————————————————	Total Depth 45.5 ft.						
	<u> </u>	PROJECT		- <u></u>		HOLE HO.	2-3	

SINGLE COMPLETION WELL	
CONSTRUCTION LOG	Well Number P-3
Project Elisworth Z-Phase	Project Number 612-00 - 31-37
Location CU-\	Datum
Top of Casing Elevation	Ground Surface Elevation
→ M →	
	A. Total Depth (ft) 45 ff
	-1 /
G	
<u> </u>	Drilling Method HSA
	WELL CONSTRUCTION
	C. Casing Length (ft)
	Type PVC
	D. Casing Diameter (ft) 2 ivol
E H	E. Depth to Top of Slotted Interval (ft) 24.5
-a-	F. Perforated Casing Length (ft) Zo ft
	Perforated Interval From 44.5 to 24.5 ft
	Perforation Type 5 (offed
	Perforation Size <u>C.01</u>
A C I	G. Surface Grout Interval (ft)
<u> </u>	Grout Material
	H. Backfilled Interval (ft)
	Backfill Material M4
	I. Sealed Interval (ft) $\frac{21.4 - 19.7}{}$
	Seal Material Granular Beutaute.
	J. Filter Pack Interval (ft) 45.5-21.9
	Pack Material 10/W Silica Sol.
F	K. Bottom Seal Interval (ft)
	Seal Material
	L. Depth to Top of Casing (in)
	M. Protective Casing Diameter (in)
	10%
↓	10/2 Sand 1 Granular Buntavite
	1 Vanular Dintarte
Д Б	

				•	Page	/of(
		CONTAL	INERIZED MAT	ERIALS LOG		
Project	Ellsworth AFE	4B Z	Phose Proj		<u>uu-1</u>	
City E	Ellsworth AFE	State	South	Dakota		
	Well or Boring Number	Material Type	Date Filed First Last	Location Moved to	Final Disposition	Signatures of Movers
Drum Number	D.	Soil	6/20 6/20			
	7 /	2011	6/21/17/			

sand a land a complete of the first of the sand of the sand of the sand of the sand of the sand of the sand of

	Well or Boring Number	Material Type	Date	Filed Last	Location Moved to	Final Disposition	Signatures of Movers
Drum Number	P-	Soil	6/20				
2	P-1	Suil	6/20	6720			
	p.z.	Soil	6/20	6/20			
3 4	P-2	501	6/20	6/20			
5	P-Z_	Soil	6/20	6/20			
6	Vapor Probes	Suil	6/21	6/21			
7	P-3	Soil	6/21	62			
4)	P-3/Vapor Proces	Soil	6/21	6/21			
9	Vapor Perbe	Soil				ļ.	
(D	All	Decon. Water	6/22	6/22			
11	A11	Decan Wichen				 	
12	Vapor Pobe	Suil	6/21	6/22			
		<u> </u>	<u> </u>	<u> </u>			
			-				
				<u> </u>			
			-				
			-		-		
					+		
				+			

WELL DEVELOPMENT LOG

Page __ of __

WELL NUMBER: P-1 DATE/TIME: 6/22/95 / PROJECT/NUMBER: 662-001-31-37	LOCATION: UL-1 Elkworth AFB WEATHER: Musty Sunny EUF REPORTEDBY: Gary Dyla
PIELD MEASUREMENTS DEVELOPMENT STARTED: 1310 DEPTH TO WATER BELOW TOC (ft.): 18.80 WATER COLUMN HEIGHT (ft.): 3 WELL VOLUMES (gal.):	DEVELOPMENT ENDED: 1435 WELL DEPTH BELOW TOC (ft.): 45 ft CASING DIAMETER (ft.): 0.167 DISCHARGE VOLUME (gal.):
DEVELOPMENT METHOD/TOOLS USED: Vi-ted Sarge	Huck, bailer

Time	Cumulative Volume Purged (gal.)	pН	Conductivity (µmhos/cm)	T と(9F)	Comments (Water clarity, etc.)
1355	1/4 gal	6.8		A. 1	Turbid + Silty
1910	Yagal 4 gal	7.5		19.8	Turbidt silfy
1925	9	7.5		12.6	les tarbect
1435	13	7.6		13.6	51, shifty trubial, note near 6+m. W.Le 37 ft, So work av P-Z
() -					W.Le 37 ft, so work av P-Z
					· ·
					·
					·

COMMENTS Use vented surge block to surge well for 5-minutes. Then bail.

DEVELOPERS SIGNATURE(S)



WELL DEVELOPMENT LOG

Page 1 of 1

WELL NUMBER: P-Z DATE/TIME: 6-22-95 / PROJECT/NUMBER: 6(2-001-045-31-37		WEATHER: Unexast Rainy 700 REPORTEDBY: Gray Dyla
DEVELOPMENT STARTED: 1450 DEPTH TO WATER BELOW TOC (ft.): 17.69 WATER COLUMN HEIGHT (ft.):		DEVELOPMENT ENDED: 1640 WELL DEPTH BELOW TOC (ft.): 45.0 CASING DIAMETER (ft.): (Z - i wch.) 0.167- DISCHARGE VOLUME (gal.):
THE STATE OF THE SALE HINE	and bailer	

Time	Cumulative Volume Purged (gal.)	pН	Conductivity (µmhos/cm)	T (9F)C	Comments (Water clarity, etc.)
1450	<i>Y</i> 4	7.8		136	Turbid + silty Turbia + silty Shop to allow recharge. Still very silty Bailed dry. Very silty therbid.
1510	10	7.7		17.9	Turbia tsilty
1513	13	,		-	Stop to allow recharge. Still very silty
1640	18	-	-	_	Bailed dry. Very silly thurbid.
			/		
	·				
					·

COMMENTS Surge well with verted surge blode first.

DEVELOPERS SIGNATURE(S)

Many

WELL NUMBER: P-3	LOCATION: OU-1 5 (15 worth AFIS
WELL HOMBER.	WEATHER: (NHICCOT, Pain 70°
DATE/TIME: 622-95 /	WEATHER OWNED THE
PROJECT/NUMBER: 612-001-31-37	REPORTED BY: GOLY Dyce
	·
2	
FIELD MEASUREMENTS	
DEVELOPMENT STARTED: 1520	DEVELOPMENT ENDED: 1615
DEPTH TO WATER BELOW TOC (ft.): 17.82	WELL DEPTH BELOW TOC (ft.): 45.0
•	CASING DIAMETER (ft.): 0.167
WATER COLUMN HEIGHT (ft.):	
3 WELL VOLUMES (gal.):	DISCHARGE VOLUME (gal.):
DEVELOPMENT METHOD/TOOLS USED: Surge bluck follow	ind by bailer
DEAETOLWENT WELHOD/LOOPS ORFD: 2006 2006 4006	MG

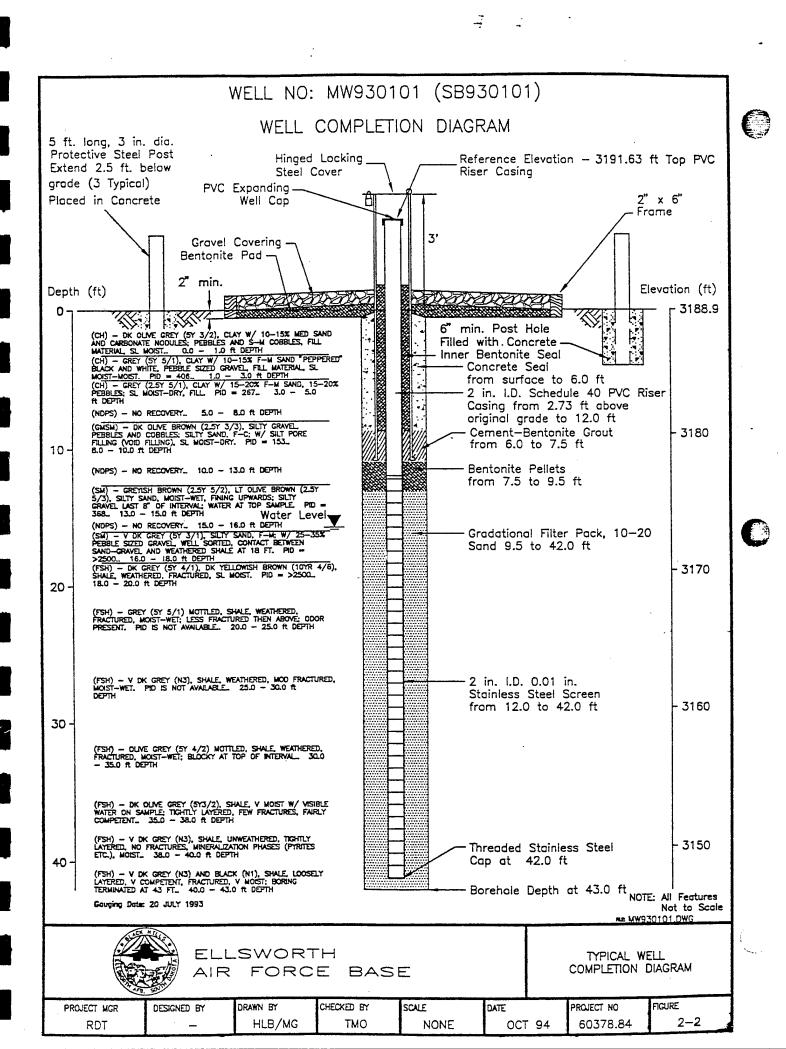
Time	Cumulative Volume Purged (gal.)	pН	Conductivity (µmhos/cm)	T (4FC)	Comments (Water clarity, etc.)
1535	Y4	7.5		16.4	Stryhtly turbech. Turbid + Silty
1555	5	6.9		14.8	Turbid + Silly
1605	(0	6.9		12.6	
1615	12	6.9		12,5	Mostly clear:
			1		
					•
					·
			·		

COMMENTS Surged with vented surge block before developing.



SOIL BORING NO: SB930101 SOIL BORING LITHOLOGY DIAGRAM

Depth (ft)						Ele	evation (ft)
FīII		0.00 ft - 1.00 f (CH) - GREY (5Y PEBBLE SIZED GR ft Depth.	t Depth. 5/1), CLAY W/ AVEL, FILL MATERL	10-15% FMG SAND AL, SL MOIST-MOIS	MED SAND AND CALL MOIST, DIFFICULT "PEPPERED" BLACT PID = 406. 1.0 M SAND, 15-20% th.	K AND WHITE, 00 ft - 3.00	7 3188.
		NO RECOVERY. (GMSM) — DK OLL SILTY SAND, V FIN = 153. 8.00 ft	VE BROWN (2.5Y IE, W/ SILT PORE 10.00 ft Depth	3/3), SILTY GRAVEI FILLING (VOID FILL	L, PEBBLES AND C JNG), SL MOIST—DR	OBBLES AND Y. PID	- 3180
10 Sand & Gravel		NO RECOVERY. (SM) - GREYISH MOIST-WET, FINING	BROWN (2.5Y 5/2 UPWARDS, SILTY) AND LT OLIVE BR	ROWN (2.5Y5/3), SI NTO A SILTY GRAY 0 ft — 15.00 ft De	ILTY SAND, EL LAST 8 FT OF	
Sand & Silt		NO RECOVERY. (SM) — V DK GREPOORLY GRADED; PID = >2500. 1	EY (5Y 3/1), SILT CONTACT BETWEEN 6.00 ft - 18.00	Y SAND, F-M, W/ SAND-GRAVEL AN ft Depth.	25-35% PEBBLE S D WEATHERED SHAL 4/6). SL MOIST. N	IZED GRAVEL, E AT 18 FT.	- 3170
20 -		FRACTURED. PID	= >2500. 18.00) ft — 20.00 ft De	epth. ^ RED, LESS FRACTI E. 20.00 ft – 25		- 3170
50 F. SHALE		V DK GREY (N3). AVAILABLE. 25.00	WEATHERED SHA ft - 30.00 ft D	LE, MOD FRACTURE	D, MOIST-WET, PID	IS NOT	- 3160
		DK OLIVE GREY (5	3.5' SLIGHTLY FRA	ACTURED, MOIST-WE	ON SAMPLE, SHALE of t = 38.00 ft D	00 ft Depth.	
0 -					LY LAYERED, NO F 10.00 ft Depth. E, LOOSELY LAYER F, 40.00 ft - 43.		- 3150
Boring Completion Date 13-	-JUN-93	······································			**************************************		FLE IMED
	ELL	SWORT FORC	H E BAS	E	Coord	dinates E: 11! N: 12	54904.8 24205.4
PROJECT MGR DESIGNE	D BY	DRAWN BY TRB/MG	CHECKED BY HLB	SCALE NONE	DATE APR/94	PROJECT NO 60378.84	FIGURE



APPENDIX C

Field Data Tables

4:16 PM 12/11/95 Piezometer PIL_TEST.XLS

^a The water level is measured from below the top of the casing.

Actual Schedule Actual Sch					Ellsv	vorth A	Vir For	Ellsworth Air Force Base - Two (2) Pha Vapor Probes Field Measurements Data Sheet	- P	o (2) Ph SS ata Shee	ase Pil	Two (2) Phase Pilot Test robes 11s Data Sheet				
National N			Actu	al Sched	ule						Vарн	ir Probe \	/женит			
VFI VF2 VF3 VF22D VF20D VF60D VF60D VF10 VF22D VF20D VF2				T.	me			Shallow	Deep	Shallow	Deep	Shallow		Deen	Deen	Deen
12.32 12.35 12.46 12.38 12.47 12.48 12.52 12.25 12.35 12.46 12.48 12.45 12.46 12.48 12.52 12.25 12.35 12.46 12.48 12.45 12.48 12.52 12.25 12.25 13.05 13.06 12.45 12.48 12.52 12.25 13.05 13.05 13.06 12.45 12.48 12.45 12.48 12.52 12.25 13.05 13.05 13.05 13.05 13.05 13.05 13.05 13.05 13.25 13.0		VP1	VP2	VP3	VP22D	VP50D	VP60D	VPIS	VP1D	VP2S	VP2D	VP3S	VP3D	VP22D	VPS0D	VP60D
12.32 0.00 <t< th=""><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th>MW93</th><th></th><th>š</th><th></th><th></th><th></th><th></th><th></th><th></th></t<>								MW93		š						
12.37 12.35 0.00 0.00 0.00 0.02 0.02 0.02 0.01 0.02 <		12:32	0:00	0:00	0:00	0:00	00:00	0.2	0.15							
12.45 12.46 12.38 12.47 12.48 12.52 0.22 0.13 0.14 b 0.15 0.25 0.16 0.15 0.25 0.16 0.15 0.15 0.25 0.16 0.15 0.25 0.15 0.15 0.15 0.15 0.25 0.24 b 0.17 0.29 0.22 0.24 0.23 0.24	\dashv	12:37	12:35	0:00	0:00	0:00	0:00	0.2	0.25	0.1	0.1					
12.53 13.05 13.06 13.03 12.59 13.02 0.25 0.29 0.17 0.0	\dashv	12:45	12:46	12:38	12:47	12:48	12:52	0.22	0.29	0.13	0.14	q	0.15	0.25	0.16	0.07
13:22 13:23 13:24 13:25 13:25 13:24 13:25 13:25 13:24 13:25 13:25 13:24 13:25 13:27 13:28 8:39 8:37 8:38 8:40 0.62 0.73 0.42 0.42 0.73 0.42 0.42 0.73 0.44 0.42 0.42 0.42 0.42 0.43 0.42 0.43 0.44 0.42 0.43 0.44 0.42 0.43 0.44 0.45 0.42 0.43 0.44 0.45		12:53	13:05	13:00	13:03	12:59	13:02	0.25	0.29	0.16	0.17	þ	0.17	0.29	0.2	0.12
14:50 14:50 15:00 <th< td=""><td>_</td><td>13:22</td><td>13:23</td><td>13:27</td><td>13:24</td><td>13:25</td><td>13:28</td><td>0.31</td><td>0.36</td><td>0.22</td><td>0.24</td><td>9</td><td>0.2</td><td>0.34</td><td>0.24</td><td>0.14</td></th<>	_	13:22	13:23	13:27	13:24	13:25	13:28	0.31	0.36	0.22	0.24	9	0.2	0.34	0.24	0.14
16:46 16:49 16:51 16:48 16:50 16:54 16:55 16:54 16:55 16:54 16:54 16:55 16:54 16:56 16:54 16:56 16:54 16:56 16:54 16:56 16:54 16:56 16:54 16:54 16:56 16:54 16:56 16:54 16:56 16:57 16:45 16:56 16:57 16:45 16:56 16:57 16:45 16:56 16:57 <th< td=""><td>_</td><td>14:50</td><td>14:58</td><td>15:02</td><td>14:59</td><td>15:00</td><td>15:03</td><td>0.38</td><td>0.5</td><td>0.25</td><td>0.26</td><td>P</td><td>0.29</td><td>0.42</td><td>0.3</td><td>0.18</td></th<>	_	14:50	14:58	15:02	14:59	15:00	15:03	0.38	0.5	0.25	0.26	P	0.29	0.42	0.3	0.18
21:02 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 21:04 0.43 0.44 0.42 0.44 0.42 0.44<	\dashv	16:46	16:49	16:51	16:48	16:50	16:52	0.47	0.56	0.27	0.27	q	0.29	0.49	0.31	0.22
8.36 8.38 8.39 8.37 8.38 8.40 0.62 0.73 0.44 0.42 b 0.43 0.64 0.74 13.25 13.27 13.28 13.32 13.31 13.33 0.65 0.75 0.45 b 0.48 0.64 0.57 16.18 16.22 16.26 16.26 0.65 0.79 0.43 0.45 b 0.48 0.66 0.53 16.18 16.20 16.22 16.26 0.65 0.79 0.49 1.08 0.64 0.73 0.73 18.50 15.08 15.11 15.10 15.12 0.68 0.82 0.44 0.46 4.1 0.73 0.	┥	21:02	21:07	21:10	21:04	21:09	21:12	0.53	0.64	0.34	0.37	q	0.38	0.57	0.42	0.26
13:25 13:27 13:28 13:32 13:31 13:33 0.65 0.75 0.45 b 0.49 0.66 0.53 16:18 16:20 16:22 16:22 16:24 16:26 16:26 0.65 0.79 0.43 0.45 b 0.48 0.64 0.53 8:59 9:00 16:22 16:26 16:24 16:26 0.65 0.79 0.43 b 0.49 0.64 0.53 0.73 0.52 15:02 15:08 15:02 20:22 20:20 20:22 20:20 0.69 0.8 0.44 0.44 0.45 0.74 0.71 0.53 15:04 12:44 12:47 12:48 8:51 8:50 8:52 0.69 0.8 0.44 0.44 4.1 0.7 0.7 16:25 16:28 16:30 16:32 16:31 16:34 0.64 0.8 0.44 4.1 0.7 0.7 0.44 4.1 0.7	\dashv	8:36	8:38	8:39	8:37	8:38	8:40	0.62	0.73	0.4	0.42	q	0.43	0.64	0.47	0.27
16:18 16:20 16:20 16:20 0.65 0.79 0.43 0.45 b 0.48 0.64 0.53 8:59 9:00 9:02 9:04 9:03 9:06 0.7 0.81 0.45 0.47 b 0.65 0.73 0.52 15:02 15:08 15:11 15:10 15:12 0.68 0.82 0.45 0.49 0.03 0.73 0.53 20:15 20:16 20:18 20:22 20:20 20:23 0.7 0.83 0.46 0.47 2 0.47 0.7 0.83 0.46 0.41 0.5 0.73 0.51 20:15 20:18 20:20 20:20 20:20 20:23 0.74 0.46 0.41 0.7 0.7 0.84 0.44 0.46 4.1 0.7 0.7 0.42 0.44 0.46 0.41 0.7 0.7 0.7 0.7 0.84 0.44 0.46 4.1 0.7 0.7 0.7	-	13:25	13:27	13:28	13:32	13:31	13:33	0.65	0.76	0.42	0.45	q	0.49	99.0	0.5	0.29
8:59 9:00 9:02 9:04 9:03 9:06 0.7 0.81 0.45 0.47 b 0.5 0.73 0.52 15:02 15:02 15:08 15:11 15:10 15:12 0.68 0.82 0.45 0.49 1.08 0.53 0.73 0.54 0.74 0.74 0.71 0.51 0.53 0.53 0.74 0.71 0.72 0	\dashv	16:18	16:20	16:22	16:26	16:24	16:26	0.65	0.79	0.43	0.45	q	0.48	0.64	0.53	0.32
15:02 15:05 15:08 15:11 15:12 0.68 0.82 0.45 0.49 1.08 0.53 0.73 0.83 20:15 20:16 20:18 20:22 20:20 20:23 0.7 0.83 0.46 0.47 2 0.47 0.71 0.51 8:43 8:46 8:51 8:52 0.69 0.8 0.44 0.46 4.1 0.5 0.7 0.71 0.51 12:44 12:47 12:51 12:52 0.69 0.8 0.44 4.1 0.5 0.7 0.73 0.74 0.7 0.7 0.8 0.44 4.1 0.5 0.7 0.7 0.44 4.1 0.7 0.7 0.4 4.1 0.7 0.7 0.4 4.1 0.7 0.7 0.4 0.4 0.7 0.7 0.7 0.7 0.7 0.4 4.1 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 <t< td=""><td>-</td><td>8:59</td><td>9:00</td><td>9:02</td><td>9:04</td><td>9:03</td><td>90:6</td><td>0.7</td><td>0.81</td><td>0.45</td><td>0.47</td><td>p</td><td>0.5</td><td>0.73</td><td>0.52</td><td>0.35</td></t<>	-	8:59	9:00	9:02	9:04	9:03	90:6	0.7	0.81	0.45	0.47	p	0.5	0.73	0.52	0.35
20:15 20:16 20:18 20:22 20:20 20:23 0.7 0.83 0.46 0.47 2 0.47 0.71 0.51 8:43 8:46 8:48 8:51 8:50 8:52 0.69 0.8 0.44 0.41 0.5 0.71 0.51 12:44 12:47 12:48 12:51 12:50 12:52 0.69 0.75 0.42 0.44 4.1 0.5 0.71 0.51 16:25 16:28 16:30 16:32 16:31 16:34 0.64 0.8 0.44 4.1 0.5 0.71 0.72 0.44 4.1 0.5 0.71 0.49 0.71 0.44 4.1 0.5 0.71 0.44 4.1 0.5 0.71 0.44 4.1 0.5 0.71 0.44 4.1 0.5 0.71 0.71 0.71 0.71 0.71 0.71 0.71 0.71 0.71 0.71 0.72 0.72 0.72 0.72 0.72	\dashv	15:02	15:05	15:08	15:11	15:10	15:12	89.0	0.82	0.45	0.49	1.08	0.53	0.73	0.53	0.32
8:43 8:46 8:48 8:51 8:50 8:52 0.69 0.8 0.44 0.46 4.1 0.5 0.7 0.51 0.5 12:44 12:44 12:48 12:51 12:52 0.66 0.75 0.42 0.44 4.1 0.5 0.71 0.51 16:25 16:36 12:51 12:50 12:52 0.66 0.75 0.45 0.46 3.5 0.48 0.71 0.51 0.71 0.51 16:25 16:36 16:31 16:34 0.69 0.81 0.46 0.45 0.48 0.75 0.48 0.7 0.48 0.7 0.48 0.7 0.48 0.7 0.48 0.7 0.48 0.7 0.49 0.7 0.47 4.8 0.7 0.73 0.7 0.49 0.7 0.45 0.7 0.7 0.7 0.4 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 0.7 <td>\dashv</td> <td>20:15</td> <td>20:16</td> <td>20:18</td> <td>20:22</td> <td>20:20</td> <td>20:23</td> <td>0.7</td> <td>0.83</td> <td>0.46</td> <td>0.47</td> <td>2</td> <td>0.47</td> <td>0.71</td> <td>0.51</td> <td>0.31</td>	\dashv	20:15	20:16	20:18	20:22	20:20	20:23	0.7	0.83	0.46	0.47	2	0.47	0.71	0.51	0.31
12:44 12:47 12:48 12:51 12:50 12:52 0.6 0.75 0.44 4.1 0.5 0.71 0.51 16:25 16:28 16:30 16:32 16:31 16:34 0.64 0.8 0.46 3.5 0.48 0.7 0.49 8:36 8:39 8:41 8:42 8:46 0.69 0.81 0.45 0.47 5 0.73 0.73 0.53 11:38 11:40 11:42 11:43 11:47 0.7 0.84 0.45 0.47 4.8 0.75 0.73 0.53 11:38 11:40 11:41 14:13 0.0 0	\dashv	8:43	8:46	8:48	8:51	8:50	8:52	69.0	0.8	0.44	0.46	4.1	0.5	0.7	0.51	0.32
16:25 16:28 16:30 16:31 16:34 0.64 0.8 0.4 0.46 3.5 0.48 0.7 0.49 8:36 8:39 8:41 8:44 8:42 8:46 0.69 0.81 0.45 0.47 5 0.5 0.73 0.53 11:38 11:40 11:42 11:41 11:43 11:47 0.7 0.84 0.45 0.47 4.8 0.5 0.73 0.51 0.51 14:11 14:12 14:14 11:41 14:13 0.7 0.84 0.45 0.47 4.8 0.52 0.7 0.51 0.7 <t< td=""><td>\dashv</td><td>12:44</td><td>12:47</td><td>12:48</td><td>12:51</td><td>12:50</td><td>12:52</td><td>9.0</td><td>0.75</td><td>0.42</td><td>0.44</td><td>4.1</td><td>0.5</td><td>0.71</td><td>0.51</td><td>0.34</td></t<>	\dashv	12:44	12:47	12:48	12:51	12:50	12:52	9.0	0.75	0.42	0.44	4.1	0.5	0.71	0.51	0.34
8:36 8:39 8:41 8:42 8:46 0.69 0.81 0.45 0.47 5 0.5 0.73 0.53 11:38 11:40 11:42 11:44 11:43 11:47 0.7 0.84 0.45 0.47 4.8 0.52 0.7 0.51 14:11 14:12 14:14 11:43 11:47 0.7 0.84 0.45 0.47 4.8 0.52 0.7 0.51 14:11 14:12 14:14 14:13 0 <td>+</td> <td>16:25</td> <td>16:28</td> <td>16:30</td> <td>16:32</td> <td>16:31</td> <td>16:34</td> <td>0.64</td> <td>8.0</td> <td>0.4</td> <td>0.46</td> <td>3.5</td> <td>0.48</td> <td>0.7</td> <td>0.49</td> <td>0.31</td>	+	16:25	16:28	16:30	16:32	16:31	16:34	0.64	8.0	0.4	0.46	3.5	0.48	0.7	0.49	0.31
11:38 11:40 11:42 11:44 11:43 11:47 0.7 0.84 0.45 0.47 4.8 0.52 0.7 0.51 14:11 14:12 14:12 14:14 14:13 0 0 0 0 0 0 0 0 14:54 14:55 14:57 14:59 14:58 15:00 0 0 0 0 0 0 0 0 0	+	8:36	8:39	8:41	8:44	8:42	8:46	69.0	0.81	0.45	0.47	5	0.5	0.73	0.53	0.33
14:11 14:12 14:14 14:13 0	+	11:38	11:40	11:42	11:44	11:43	11:47	0.7	0.84	0.45	0.47	4.8	0.52	0.7	0.51	0.35
ESVE Test 14:54 14:55 14:57 14:59 14:58 15:00 0 0.05 0 0.79 0	-	14:11	14:11	14:12	14:12	14:14	14:13	0		0	0	0	0	0	0	0
14:54 14:55 14:57 14:59 14:58 15:00 0 0.05 0 0.79 0 0 0 0 16:43 16:44 16:46 16:49 16:49 16:49 0 0 0 1.12 0	ŀ							ESV	E Test							
16:43 16:44 16:46 16:45 16:49 0 0 0 112 0 0 0 8:49 8:49 8:53 8:51 8:51 0 0 0 0 4.1 0	_	14:54	14:55	14:57	14:59	14:58	15:00	0	0.05	0	0	0.79	0	0	0	0
8:49 8:49 8:51 8:51 0 0 0 4.1 0 0 0 11:38 11:38 11:39 11:41 11:41 0 0 0 0 3.85 0 0 0 14:01 14:02 14:04 14:04 14:05 0 0 0 3.5 0 0 0 14:01 14:02 14:04 14:05 0 0 0 3.5 0 0 0	\dashv	16:43	16:44	16:46	16:49	16:47	16:49	0	0	0	0	1.12	0	0	0	0
11:38 11:39 11:40 11:41 11:41 0 0 0 3.85 0 0 0 14:01 14:02 14:03 14:04 14:05 0 0 0 3.5 0 0 0 14:01 14:02 14:03 14:04 14:05 0	\dashv	8:49	8:49	8:53	8:51	8:50	8:51	0	0	0	0	4.1	0	0	0	0
14:01 14:02 14:03 14:04 14:05 0	\dashv	11:38	11:38	11:39	11:41	11:40	11:41	0	0	0	0	3.85	0	0	0	0
14:28	+	14:01	14:02	14:03	14:04	14:04	14:05	0	0	0	0	3.5	0	0	0	0
	30-Jun			14:28								3.6				

^b No reading.

			Ellsworth	sworth Air Force Base	e Base -		Two (2) Phase Pilot Test	ot Test		
			2-PH	ASE Systen	n Operatin	g Conditic	2-PHASE System Operating Conditions Data Sheet	set		
Actual !	Actual Schedule	Syste	System Inlet	Wellhead	head	Seal	Seal Fluid	Exhaus	Exhaust Vapor	
		Temp.	Уасиин	Vacuum	Valve	Temp.	Pressure	Тетр.	Pressure	
Day	Time	(deg F)	(in. Hg)	(in. Hg)	Position	(deg F)	(psi)	(deg F)	(psi)	Comments
					MW930101	11 Test				
25-Jun	12:40	40	26.0	11.0	closed	178	1	120	2	
25-Jun	14:50	40	25.0	7.0	closed	178	1	130	2.5	
25-Jun	16:40	40	25.0	0.9	closed	178	1	133	2.6	
25-Jun	18:30	40	25.0	5.5	closed	178	1	132	2.6	
25-Jun	20:50	40	23.5	4.5	closed	176	1	132	2.7	
26-Jun	8:40	40	24.0	2.8	closed	176	1	132	3	
26-Jun	12:20	40	24.0	2.8	closed	176	1	141	2.5	
26-Jun	16:10	40	24.5	2.5	closed	176	1	145	2.7	
27-Jun	9:10	40	24.0	2.0	closed	175	1	142	2.7	
27-Jun	15:20	40	24.5	1.9	closed	172	1	150	2.8	T.O.S. 17.5"Hg
27-Jun	20:10	40	23.0	1.8	closed	174	1	138	2.8	
28-Jun	9:00	40	21.7	1.0	closed	180	1	121	2.9	T.O.S. 17.5"Hg
28-Jun	13:00	40	22.0	1.0	closed	179	1	124	2.9	
28-Jun	16:40	40	22.6	1.0	closed	177	1	125	2.8	T.O.S. 17.4"Hg
29-Jun	8:50	40	22.5	< 1.0	closed	169	1	125	2.9	T.O.S. 17.4"Hg
29-Jun	11:50	40	22.5	< 1.0	closed	172	1	128	2.9	T.O.S. 17.4"Hg
					ESVE Test	Test				
29-Jun	15:10	40	27.0	19.7	%09	177	1	112	1.7	
29-Jun	16:40	40	27.0	20.0	20%	178	1	112	1.8	T.O.S. 24"Hg
30-Jun	9:10	27	27.5	20.7	20%	178	1	115	1.9	T.O.S. 24"Hg
30-Jun	11:50	31	27.0	20.5	20%	178	1	119	1.8	T.O.S. 24.5"Hg
30-Jun	14:30	31	27.5	20.5	20%	180	1	121	1.8	T.O.S. 24.5"Hg

T.O.S. = Top Of Straw

	F			- Two (2) Ph		t
Actual S	Schedule Time	Cumulative Time (hrs)	Totalizer Readings (gal)	Effluent to Tank Actual Liquid Flow (gal)	Liquid Flow (gpm)	Gas out Stack Vapor Flow Rate (cfm)
			MW930	101 Test		
25-Jun	12:40	0.00	55753.4	<u>-</u>	0.00	21
25-Jun	14:50	2.17	55851.9	98.5	0.75	36
25-Jun	16:40	4.00	55894.3	42.4	0.76	40
25-Jun	18:30	5.83	55923.3	29.0	0.39	41
25-Jun	20:50	8.17	55964.8	41.5	0.26	44
26-Jun	8:40	20.00	56127.7	162.9	0.30	51
26-Jun	12:20	23.67	56180.8	53.1	0.23	51
26-Jun	16:10	27.50	56231.1	50.3	0.24	55
27-Jun	9:10	44.50	56440.4	209.3	0.22	56
27-Jun	15:20	50.67	56504.7	64.3	0.21	57
27-Jun	20:10	55.50	56556.6	51.9	0.17	57
28-Jun	9:00	68.33	56691.4	134.8	0.17	55
28-Jun	13:00	72.33	56734.0	42.6	0.17	57
28-Jun	16:40	76.00	56766.4	32.4	0.15	57
29-Jun	8:50	91.17	56938.7	172.3	0.17	57
29-Jun	11:50	94.17	56954.9	16.2	0.09	57
end	12:32	94.87	56978.7	23.8	0.26	•
			ESVE	Test		
29-Jun	15:10	110.03	56990.2	11.5	0.08	13.5
29-Jun	16:40	111.53	57000.3	10.1	0.11	13.5
30-Jun	9:10	128.03	57091.4	91.1	0.09	13.5
30-Jun	11:50	130.70	57103.6	12.2	0.06	13.5
30-Jun	14:30	133.37	57122.0	18.4	0.07	13.5

Flow Rates PIL_TEST.XLS 4:40 PM 12/11/95

Ellsworth Air Force Base Two (2) Phase Pilot Test Field Measurements Data Sheet Vapor Flow out Stack Pipe Breathing Actual Schedule Effluent Zone Time Day (ppm) (ppm) MW930101 Test 25-Jun 15:58 888 25-Jun 16:01 25-Jun 18:30 956 1 25-Jun 21:03 749 1 26-Jun 8:45 755 12:30 702 26-Jun 26-Jun 16:40 609 27-Jun 9:42 742 <1 27-Jun 15:23 682 <1 27-Jun 16:20 663 <1 28-Jun 9:00 764 <1 28-Jun 16:35 720 <1 29-Jun 8:50 763 <1 29-Jun 11:38 810 <1 29-Jun 11:47 766 <1 29-Jun 13:38 * 782 **ESVE Test** 29-Jun 16:40 139 <1 14:26 <1 29-Jun 342 29-Jun 16:40 139 <1 30-Jun 9:40 34 <1 30-Jun 11:10 29 <1

^a Measurements using Photo Ionizing Detector (PID)

^{*}End of a 4 day test on MW930101.

Ellsworth Air Force Base Two (2) Phase Pilot Test Field Measurements Data Sheet

Actual !	Schedule	Barometric Readings
Day	Time	(MBars)
25-Jun	12:15	908
26-Jun	16:52	905
27-Jun	8:40	904
27-Jun	15:00	906
27-Jun	20:18	906
28-Jun	8:43	908
28-Jun	16:23	910
29-Jun	8:35	913
29-Jun	15:51	913

Barometer PIL_TEST.XLS 9:07 AM 12/12/95

APPENDIX D

Groundwater Sample Analytical Data



ENERGY LABORATORIES, INC.

P.O. BOX 2470 • RAPID CITY, SD 57709 • PHONE (605) 342-1225 610 FARNWOOD STREET • RAPID CITY, SD 57701 • FAX (605) 342-1397

Radian Corporation 8501 Box 201088 Austin, TX 78720-1088 AMENDED REPORT

Ellsworth AFB, SD

Sampled: 06-23/25-95

June 30, 1995 95-38005-007

Submitted: 06-25-95

					•		1
Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed

Water Analysis

MW930101 95-38005 **8260 LONG**

RH:06-26-95 Units <u>µg/L</u>

PQL' 1,1-Dichloroethene < 10 10 Methylene Chloride 10 < 10 trans-1.2-Dichloroethene $< 10^{2}$ 10 < 10² 1,1-Dichloroethane 10 2, 2-Dichloropropane <10 10 39005 10 cis-1, 2-Dichloroethene Bromochloromethane < 10 10 Chloroform < 10 10 1.1.1-Trichloroethane < 10 10 Carbon Tetrachloride <10 10 1,1-Dichloropropene < 10 10 Benzene 8704 10 1,2 Dichloroethane <10 10 Trichloroethene 18 1.0 1, 2-Dichloropropane < 10 10 Dibromomethane < 10 10 Bromodichloromethane < 10 10 Trans-1,3-Dichloropropene < 10 10 Toluene 100 10 cis-1,3-Dichloropropene < 10 10 1,1,2-Trichloroethane < 10 10 Tetrachloroethene <10 10 1,3-Dichloropropane <10 10 Dibromochloromethane <10 10 1, 2-Dibromoethane < 10 10 Chlorobenzene 10 < 10 1,1,1,2-Tetrachloroethane < 10 10 Ethylbenzene 2804 10 M+P Xylenes 8404 10 0-Xylene 140 10 Styrene < 10 10 Bromoform < 10 10 Isopropylbenzene 40 10 Bromobenzene < 10 10 1,1,2,2-Tetrachloroethane < 10 10 1,2.3-Trichloropropane < 10 10 n-Propylbenzene 49 10 2-Chlorotoluene < 10 10 4-Chlorotoluene < 10 10 1,3,5-Trimethylbenzene 120 10 tert-Butylbenzene <10 10 1,2,4-Trimethylbenzene 3104 10 $< 10^{2}$ sec-Butvibenzene 10 1.3-Dichlorobenzene < 10 10 1,4-Dichlorobenzene < 10 10 p-isopropyltoluene 21 10 1,2-Dichlorobenzene <10 10 n-Butylbanzene 14 10 1,2-Dibromo-3-Chloropropane < 10 10 1,2,4-Trichlorobenzene <10 10 Naphthalene 150⁴ 10 Hexachlorobutadiene 10 <10

Site	Depth	Lab No.	Methodology ,	Analysis	Results	Units	Analyzed
MW930101	cont. 9	5-38005	8260 LONG continue	d			RH:06-26-9
						<u>PQL</u>	Units <u>µg/L</u>
				1, 2, 3-Trichlorobenzene	<10	10	
				Acetone	380	200	
				Methyl Ethyl Ketone	<100	100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane	<10	10	
				Vinyl Chloride	<10	10	
				Bromomethane	<10	10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	10	
				Carbon Disulfide	<10	10	
				Vinyl Acetate	<10	10	
				Methyl Isobutyl Ketone	420	100	
				2-Hexanone	< 100	100	
				Acrolein	<100	100	
				Acrylonitrile	<100	100	
				Methyltertiary Butyl Ether	<10	10	;
				lodomethane	<10	10	•
			Surrogate Recoveries				
			•	1, 2-Dichloroethane-d4	66°		% Recovery
				Toluene-d8	93		70 11000 V E1 V
				4-Bromofluorobenzene	106		

 $^{^{1}}$ Sample diluted $\underline{10x}$ at analysis due to the high level of cis-1,2-dichloroethene present. 2 Present but less than the PQL.

 $^{^3}$ The high level of benzene present caused a suppression of the 1,2-dichloroethane-d $_4$. Value derived from a $\underline{100x}$ dilution.

⁵ Value derived from a 1000x dilution.

Site	Depth Lab No.	Methodology	Analysis	Results	Units	Analyzed
ffluent-1	95-38006	8260 LONG				PLI-OC DE
inacine	33-30000	DZOO LONG			PQL	RH:06-25-9 Units <u>µg/L</u>
			1,1-Dichloroethene	<1.0	1.0	24
			Methylene Chloride	<1.0	1.0	
			trans-1, 2-Dichloroethene 1, 1-Dichloroethane	<1.0	1.0	
			2, 2-Dichloropropane	<1.0 <1.0	1.0 1.0	
			cis-1, 2-Dichloroethene	<1.0	1.0	
			Bromochloromethane	<1.0	1.0	
			Chloroform 1,1,1-Trichloroethane	<1.0 <1.0	1.0 1.0	
			Carbon Tetrachloride	<1.0	1.0	
			1,1-Dichloropropene	<1.0	1.0	
			Benzene	<1.0	1.0	
			1, 2-Dichloroethane Trichloroethene	<1.0 <1.0	1.0 1.0	
			1,2-Dichloropropane	<1.0	1.0	
			Dibromomethane	<1.0	1.0	
			Bromodichloromethane	<1.0	1.0	
			Trans-1,3-Dichloropropene Toluene	<1.0	1.0	
			cis-1,3-Dichloropropene	<1.0 <1.0	1.0	:
			1,1,2-Trichloroethane	<1.0	1.0	•
			Tetrachloroethene	< 1.0	1.0	
			1,3-Dichloropropane	<1.0	1.0	
			Dibromochloromethane 1,2-Dibromoethane	<1.0 <1.0	1.0 1.0	
	•		Chlorobenzene	<1.0	1.0	
			1,1,1,2-Tetrachloroethane	<1.0	1.0	
			Ethylbenzene	<1.0	1.0	
			M + P Xylenes O-Xylene	<1.0 <1.0	1.0 1.0	
			Styrene	<1.0	1.0	
			Bromoform	< 1.0	1.0	
			Sopropylbenzene	<1.0	1.0	
			Bromobenzene 1,1,2,2-Tetrachloroethane	<1.0 <1.0	1.0 1.0	
			1, 2, 3-Trichloropropane	< 1.0	1.0	
			n-Propylbenzene	<1.0	1.0	
			2-Chlorotaluene 4-Chlorotaluene	<1.0 <1.0	1.0 1.0	
			1,3,5-Trimethylbenzene	<1.0	1.0	
			tert-Butylbenzene	<1.0	1.0	
			1,2,4-Trimethylbenzene sec-Butylbenzene	<1.0	1.0 1.0	
			1,3-Dichlorobenzene	<1.0 <1.0	1.0	
			1,4-Dichlorobenzene	< 1.0	1.0	
			p-isopropyitoluene	< 1.0	1.0	
			1,2-Dichlorobenzene n-Butylbenzene	<1.0 <1.0	1.0 1.0	
			1, 2-Dibromo-3-Chloropropani		1.0	
			1,2,4-Trichlorobenzene	<1.0	1.0	
			Naphthalene Hexachiorobutadiene	<1.0 <1.0	1.0 1.0	
			1,2,3-Trichlorobenzene	<1.0	1.0	
			Acetone	54	20	
			Methyl Ethyl Ketone	<10	10	
			Dichlorodifluoromethane Chloromethane	<1.0 <1.0	1.0 1.0	
			Vinyl Chloride	<1.0	1.0	
			Bromomethane	<1.0	1.0	
			Chloroethane Trichlorofluoromethane	<1.0 <1.0	1.0 1.0	
			2-Chloroethylvinylether	<1.0	1.0	
			Carbon Disulfide	<1.0	1.0	
			Vinyl Acetate	<1.0	1.0	
			Methyl Isobutyl Ketone 2-Hexanone	<10 '<10	10 10	
			Acrolein	<10	10	
			Acrylonitrile	<10	10	
			Methyltertiary Butyl Ether lodomethane	<1.0 <1.0	1.0 1.0	
			1000mernañ8	< 1.0	1.0	
		Surrogate Recoveries	1 2 60 11 11 11			D. coveni
			1,2-Dichloroethane-d4 Toluene-d8	100 100	%	Recovery

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
			50. 11. 1 5070				
ffluent 1	cont.	95-38006	EPA Method 8270		< 10	µg/L	⁸ 06-29-9
				Acenaphthylene Anthracene	<10 <10		
				Azobenzene	<10		
				Benzidine	< 20		
				Benzo(a)Anthracene	< 10		
				Benzo(b)fluoranthene	< 10		
				Benzo(k)fluoranthene	<10		
				Benzo(g,h,i)perylene Benzo(a)pyrene	<10 <10		
				4-Bromophenyl-phenylether	<10		
				Butylbenzyiphthalate	< 10		
				4-Chloro-3-Methylphenol	< 10		
				bis(-2-Chloroethoxy)Methane	<10		
•				bis(-2-Chloroethyl)Ether bis(2-Chloroisopropyl)ether	<10 <10		
				2-Chloronaphthalene	<10		
				2-Chlorophenol	<10		
				4-Chlorophenol	< 10	;	
				4-Chlorophenyl-phenylether-	< 10	•	
				Chrysene	<10		
				Dibenzo(a,h)anthracene	<10 <10		
				1,2-Dichlorobenzene 1,3-Dichlorobenzene	<10		
				1,4-Dichlorobenzene	<10		
				3,3-Dichlorobenzidine	< 20		
				2,4-Dichlorophenol	< 10		
				Diethylphthalate	< 10		
				Dimethyl Phthalate 2,4-Dimethylphenol	<10 <10		
				Di-n-Butylphthalate	<10		
				4,6-Dinitro-2-methylphenol	< 50		
				2,4-Dinitrophenol	< 50		
				2,4-Dinitrotoluene	< 10		
				2,6-Dinitrotoluene	< 10		
				Di-n-octyl Phthalate	< 10		
				bis(2-ethylhexyl)Phthalate Fluoranthene	2.3JB <10		
				Fluorene	<10		
				Hexachlorobenzene	< 10		
				Hexachlorobutadiene	< 10		
				Hexachlorocyclopentadiene	<10		
				Hexachloroethane	<10		
				Indeno(1,2,3-c,d)pyrene Isophorone	<10 <10		
				1-Methylnaphthalene	<10		
				2-Methylnaphthalene	< 10		
				2-Methylphenol	< 10		
				4-Methylphenol/3-Methylphenol			
				Naphthalene	< 10		
	4			Nitrobenzene	<10		
				2-Nitrophenol 4-Nitrophenol	<10 <50		
				N-Nitrosodimethylamine	<10		
				N-nitroso-Di-n-propylamine	<10		
				N-nitrosodiphenylamine	<10		
				Pentachlorophenol	< 50		
				Phenanthrene	<10		
				Phenol Pyrene	<10 <10		
				Pyridine	< 20		
				1,2,4-Trichlorobenzene	<10		
				2,4,5-Trichlorophenol	< 10		
				2,4,6-Trichtorophenol	<10		
			Surrogate Recovery	•		0/ 5	QC Limits
				2-fluorophenol	30 30	% Recovery	21-100 10-94
				Phenol-d5 Nitrobenzene-d5	30 43		35-114
				2-Fluorobiphenyi	43		43-116
				2,4,6-Tribromophenol	35		10-123

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-1		95-38007	8260 LONG				DU 00 00 0
muem-1		95-36007	8280 LONG			PQL1	RH:06-26-9
				1,1-Dichloroethene	<10	10	Units <u>µg/L</u>
				Methylene Chloride	<10	10	
		•		trans-1, 2-Dichloroethene	<10	10	
				1,1-Dichloroethane 2,2-Dichloropropane	<10 <10	10 10	
				cis-1, 2-Dichloroethene	700³	10	
				Bromochloromethane	< 10	10	
				Chloroform 1,1,1-Trichloroethane	<10 <10	10 10	
				Carbon Tetrachloride	<10	10	
				1,1-Dichloropropene	< 10	10	
				Benzene	220³	10	
				1,2-Dichloroethane Trichloroethene	< 10 16	10 10	
				1,2-Dichloropropane	<10	10	
				Dibromomethane	<10	10	
				Bromodichloromethane Trans-1,3-Dichloropropene	<10 <10	10 10	:
				Toluene	28	10	
				cis-1,3-Dichloropropene	< 10	10	
				1,1,2-Trichloroethane	<10	10	
				Tetrachloroethene 1,3-Dichloropropane	<10 <10	10 10	
				Dibromochloromethane	< 10	10	
				1,2-Dibromoethane	<10	10	
				Chlorobenzene 1,1,1,2-Tetrachloroethane	<10 <10	10 10	
				Ethylbenzene	65	10	
				M+P Xylenes	180	10	
				O-Xylene Styrene	13 <10	10 10	
		•		Bromoform	<10	10	
				Isopropylbenzene	11	10	
				Bromobenzene 1,1,2,2-Tetrachloroethane	<10 <10	10 10	
				1,2,3-Trichloropropane	<10	10	
				n-Propylbenzene	12	10	
				2-Chlorotoluene 4-Chlorotoluene	<10 <10	10 10	
				1,3,5-Trimethylbenzene	24	10	
				tert-Butylbenzene	< 10	10	
				1,2,4-Trimethylbenzene sec-Butylbenzene	130 <10	10 10	•
				1,3-Dichlorobenzene	<10	10	
				1,4-Dichlorobenzene	< 10	10	
				p-Isopropyltoluene	<10	10	
				1,2-Dichlorobenzene n-Butylbenzene	<10 <10	10 10	
				1, 2-Dibromo-3-Chloropropane		10	
				1,2,4-Trichlorobenzene	<10	10	
				Naphthalene Hexachlorobutadiene	30 <10	10 10	
				1,2,3-Trichlorobenzene	<10	10	
				Acetone	< 200	200	
				Methyl Ethyl Ketone Dichlorodifluoromethane	<100 <10	100 10	
				Chloromethane	<10	10	
				Vinyl Chloride	<10	10	
				Bromomethane Chloroethane	<10 <10	10 . 10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	10	
				Carbon Disulfide Vinyl Acetate	<10 <10	10 10	
				Methyl Isobutyl Ketone	<100	100	
				2-Hexanone	< 100	100	
				A crolein	<100	100	
				Acrylonitrile Methyltertiary Butyl Ether	<100 <10	100 10	
				lodomethane	<10	10	

Site	Depth Lab No.	Methodology	Analysis	Results	Units	Analyzed
Influent-1 con	t. 95-38007	8260 LONG Surrogate Recoveries			RH:06	6-26-95
		Juliogate Necovenes	1,2-Dichloroethane-de Toluene-d8 4-Bromofluorobenzen	92	% Re	ecovery

Kurt R. Slentz_

Sample diluted $\underline{10x}$ at analysis due to non-target compound sample matrix interference. The high level of benzene present caused a suppression of the 1,2-Dichloroethane-d₄

³ Value derived from a 100x dilution.

	QUALITY	ASSURANCE DATA	Ą		
Method Blank	8260 LONG				RH:06-25-95
				PQL	Units µg/L
		1,1-Dichloroethene	< 1.0	1.0	5 mm <u>247-</u>
		Methylene Chloride	< 1.0	1.0	
		trans-1,2-Dichloroethene	<1.0	1.0	
		1,1-Dichloroethane	< 1.0	1.0	
		2,2-Dichloropropane	< 1.0	1.0	
•		cis-1,2-Dichloroethene Bromochloromethane	< 1.0 < 1.0	1.0 1.0	
•		Chioroform	< 1.0	1.0	
		1,1,1-Trichloroethane	< 1.0	1,0	
		Carbon Tetrachloride	< 1.0	1.0	
		1,1-Dichloropropene	< 1.0	1.0	
		Benzene	< 1.0	1.0	
		1,2-Dichloroethane	< 1.0	1.0	
		Trichloroethene 1,2-Dichloropropane	< 1.0	1.0	
		Dibromomethane	< 1.0 < 1.0	1.0 1.0	
		Bromodichloromethane	<1.0	1.0	
		Trans-1,3-Dichloropropene	<1.0	1.0	
		Toluene	< 1.0	1.0	
		cis-1,3-Dichloropropene	< 1.0	1.0	
		1,1,2-Trichloroethane	< 1.0	1.0	
		Tetrachloroethene	< 1.0	1.0	
		1,3-Dichloropropane	< 1.0	1.0	
		Dibromochloromethane 1,2-Dibromoethane	< 1.0	1.0	
		Chlorobenzene	< 1.0 < 1.0	1.0 1.0	
		1,1,1,2-Tetrachloroethane	< 1.0	1.0	
		Ethylbenzene	< 1.0	1.0	
		M + P Xylenes	< 1.0	1.0	
		O-Xylene	< 1.0	1,0	•
		Styrene	< 1.0	1.0	
		Bromoform	< 1.0	1.0	
		Isopropylbenzene Bromobenzene	< 1.0 < 1.0	1.0 1.0	
		1,1,2,2-Tetrachloroethane	<1.0	1.0	
		1,2,3-Trichloropropane	<1.0	1.0	
		n-Propylbenzene	< 1.0	1.0	
		2-Chiorotoluene	< 1.0	1.0	
		4-Chlorotoluene	< 1.0	1.0	
		1,3,5-Trimethylbenzene	< 1.0	1.0	
		tert-Butylbenzene 1,2,4-Trimethylbenzene	<1.0	1.0	
		sec-Butylbenzene	< 1.0 < 1.0	1.0 1.0	
		1,3-Dichlorobenzene	< 1.0	1.0	
		1,4-Dichlorobenzene	< 1.0	1.0	
		p-isopropyltoluene	< 1.0	1.0	
		1,2-Dichlorobenzene	< 1.0	1.0	
		n-Butylbenzene	< 1.0	1.0	
		1,2-Dibromo-3-Chloropropane 1,2,4-Trichlorobenzene	<1.0 <1.0	1.0	
		Naphthalene	< 1.0	1.0	
		Hexachlorobutadiene	<1.0	1.0	
		1,2,3-Trichlorobenzene	< 1.0	1.0	
		Acetone	< 20	20	
		Methyl Ethyl Ketone	<10	10	
		Dichlorodifluoromethane	<1.0	1.0	
		Chloromethane	< 1.0	1.0	
		Vinyl Chloride Bromomethane	<1.0	1.0	
	•	Chloroethane	<1.0 <1.0	1.0 1.0	
		Trichlorofluoromethane	<1.0	1.0	
		2-Chloroethylvinylether	< 1.0	1.0	
		Carbon Disulfide	<1.0	1.0	
		Vinyl Acetate	< 1.0	1.0	
		Methyl Isobutyl Ketone	< 10	10	
		2-Hexanone	< 10	10	
		Acrolein	< 10	10	
		Acrylonitrile	<10	10	
		Methyltertiary Butyl Ether lodomethane	<1.0 <1.0	1.0 1.0	
		odometrare	~ 1.0	1.0	
•	Surrogate Recoveries				
	•	1,2-Dichloroethane-d4	100	%	Recovery
		Toluene-d8	100		
		4-Bromofluorobenzene	100		

Method Blank	8260 LONG				RH:06-26-95
				POL	Units µg/L
		1,1-Dichloroethene	< 1.0	1.0	
		Methylene Chloride	< 1.0	1.0	
		trans-1,2-Dichloroethene 1,1-Dichloroethane	< 1.0	1.0 1.0	
		2,2-Dichloropropane	<1.0 <1.0	1.0	
		cis-1,2-Dichloroethene	< 1.0	1.0	
		Bromochloromethane	< 1.0	1.0	
		Chloroform	< 1.0	1.0	
		1,1,1-Trichlorgethane	< 1.0	1.0	
		Carbon Tetrachloride	< 1.0	1.0	
		1,1-Dichloropropene	< 1.0	1.0	
		Benzene	< 1.0	1.0	
		1,2-Dichloroethane	< 1.0	1.0	
		Trichloroethene	< 1.0	1.0	
		1,2-Dichloropropane	< 1.0	1.0	
		Dibromomethane Bromodichloromethane	< 1.0	1.0	
		Trans-1,3-Dichloropropene-	<1.0 <1.0	1.0 1.0	
		Toluene	< 1.0	1.0	
		cis-1,3-Dichloropropene	< 1.0	1.0	
		1,1,2-Trichloroethane	< 1.0	1.0	
		Tetrachloroethene	< 1.0	1.0	
		1,3-Dichloropropane	< 1.0	1.0	
		Dibromochloromethane	< 1.0	1.0	
		1,2-Dibromoethane	< 1.0	1.0	
		Chlorobenzene	< 1.0	1.0	
		1,1,1,2-Tetrachloroethane	< 1.0	1.0	
		Ethylbenzene	< 1.0	1.0	•
		M + P Xylenes O-Xylene	< 1.0 < 1.0	1.0	
		Styrene	< 1.0	1.0	
		Bromoform	<1.0	1.0	
		Isopropylbenzene	<1.0	1.0	
		Bromobenzene	< 1.0	1.0	
		1,1,2,2-Tetrachloroethane	< 1.0	1.0	
		1,2,3-Trichloropropane	< 1.0	1.0	
		n-Propylbenzene	< 1.0	1.0	
		2-Chlorotoluene	< 1.0	1.0	
		4-Chlorotoluene	< 1.0	1.0	
		1,3,5-Trimethylbenzene	< 1.0	1.0	
	•	tert-Butylbenzene 1,2,4-Trimethylbenzene	<1.0 <1.0	1.0 1.0	
		sec-Butylbenzene	< 1.0	1.0	
		1,3-Dichlorobenzene	< 1.0	1.0	
		1,4-Dichlorobenzene	< 1.0	1.0	
		p-isopropyltoluene	< 1.0	1.0	
		1,2-Dichlorobenzene	< 1.0	1.0	
		n-Butylbenzene	< 1.0	1.0	
		1,2-Dibromo-3-Chloropropane	< 1.0	1.0	
		1,2,4-Trichlorobenzene Naphthalene	< 1.0	1.0	
		Hexachlorobutadiene	<1.0 <1.0	1,0 1,0	
		1,2,3-Trichlorobenzene	<1.0	1.0	
		Acetone	<20	20	
		Methyl Ethyl Ketone	<10	10	
		Dichlorodifluoromethane	< 1.0	1.0	
		Chloromethane	< 1.0	1.0	
		Vinyl Chloride	< 1.0	1.0	
		Bromomethane	< 1.0	1.0	
		Chloroethane	<1.0	10	
		Trichlorofluoromethane	<1.0	1.0	
		2-Chloroethylvinylether Carbon Disulfide	<1.0 <1.0	1.0 1.0	
		Vinyl Acetate	<1.0	1.0	
		Methyl Isobutyl Ketone	<10	10	
		2-Hexanone	<10	10	
		Acrolein	<10	10	
		Acrylonitrile	< 10	10	
		Methyltertiary Butyl Ether	< 1.0	1.0	
		fodomethane	< 1.0	1.0	
	0 2				
	Surrogate Recoveries	1.2 Diables at the de	100	.,	Bassess
		1,2-Dichloroethane-d4 Toluene-d8	103	%	Recovery
		4-Bromofluorobenzene	99 102		
		. S. S. S. S. S. S. S. S. S. S. S. S. S.	102		

Trip Blank		QUALITY F	ASSURANCE DATA			
1.1-Dichlorentheme	Trip Blank	8260 LONG			POL	
### 1.20 Colorwortherne			1,1-Dichloroethene	< 1.0		CALI
1.1.01chromethame			Methylene Chloride			
2.2 Clothorospiegore en 1.2 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.0 Clothorospiegore en 1.						
Surrogeneral Surr			• •			
### Remodificamentaria						
Chlorostem						
Carbon Tarapabande				< 1.0	1.0	
1.1-Delivacepegene Sentene 1.0 1.0 Benerate 1.0 1.0 1.2-Diffishersethame 1.0 1.0 1.2-Diffishersethame 1.0 1.0 Bernode Chick operations 1.0 1.0 Trans-1.3-Dicktorepresene 1.0 1.0 Trans-1.3-Dicktorepresene 1.0 1.0 Trans-1.3-Dicktorepresene 1.0 1.0 Trans-1.3-Dicktorepresene 1.0 1.0 Trans-1.3-Dicktorepresene 1.0 1.0 Told 1.1.2-Transfersethame 1.0 1.0 I.1.2-Transfersetham 1.0 1.0 I.1.2-Transfersetham 1.0 1.0 I.1.2-Transfersetham 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktorepresente 1.0 1.0 Dicktore 1.1.1.2-Transfersethame 1.0 1.0 Dicktorepresente 1.0			1,1,1-Trichloroethane			
Berneme						
1,2 Dictionentance			• •			
TriciProcentione 1,2 Gickhorepropage 1,0 1,0 1,0 1,0 1,0 1,0 1,0 1,0 1,0 1,0						
1,2-Dickhonopropara						
Bernodichizenerelhane					1.0	
Trans-1,3-Dickhoropropersor Toleane 1,3-Dickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,2-Trickhoropropersor 1,1,1,2-Trickhoropropersor 1,1,1,2-Trickhoropropersor 1,1,1,2-Trickhoropropersor 1,1,1,2-Trickhoropropersor 1,1,1,1,1,1,1,1,1,1,1,1,1,1,1,1,1,1,1,			Dibromomethane	< 1.0		
Televier	•					
1,3-Dichieropropose		•				
1,1,2-7 insheresthane						
Terrachiorectheme						
1,3-Dichloropopane						
1,2 Disvoncethare					1.0	
Chlorobentrane	,		Dibromochloromethane	< 1.0	1.0	
1,1,1,2-Text-chroresthane			1,2-Dibromoethane			
Ethylbenzene (1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0 1.0						
M + P Xylenes						
O.Xylene						
Sivrene			•			
Bromelorm	•					
Brombonzene				< 1.0	1.0	
1.1.2.2-Tetrachtorocthane			isopropylbenzene	<1.0		
1,2,3-Trichloropropane	•	•				
n-Propylenzane		•				•
2-Chlorotoluene	•					
4-Chlorotolutene						
tert-Bury/benzene						
1,2,4-Timethylbenzene		•	1,3,5-Trimethylbenzene			
See-Buty/Ibenzene 1.0 1.0						
1,3-Dichlorobenzene						
1,4-Dichlorobenzene						
p-isopropyltokuene						
n-Butylbersene					1.0	
1,2-Dibromo-3-Chloropropane			1,2-Dichlorobenzene			
1,2,4-Trichlorobenzene						
Naphthalene		•				
Hexachlorobutadiene						
1,2,3-Trichlorobenzene						
Methyl Ethyl Ketone		·				
Dichlorodifluoromethane			Acetone	98		
Chloromethane						
Vinyl Chloride						
Bromomethane						
Chloroethane						
2-Chloroethylvinylether <1.0 1.0					1.0	
Carbon Disulfide			Trichlorofluoromethane	< 1.0		
Vinyl Acetate			_ , ,			
Methyl Isobutyl Ketone						
2-Hexanone <10 10 Acrotein <10 10 Acrotein <10 10 Acrylonitrile <10 10 Methyltertiary Butyl Ether <1.0 1.0 lodomethane <1.0 1.0 Surrogate Recoveries 1,2-Dichloroethane-d4 95 % Recovery Toluene-d8 100						
Acrolein < 10 10 Acrylonitrile < 10 10 Methyltertiary Butyl Ether < 1.0 1.0 lodomethane < 1.0 1.0 Surrogate Recoveries 1,2-Dichloroethane-d4 95 % Recovery Tolune-d8 100						
Acrylonitrile						
Methyltertiary Butyl Ether					10	
Surrogate Recoveries 1,2-Dichloroethane-d4 95 Recovery Toluene-d8 100						
1,2-Dichloroethane-d4 95 % Recovery Toluene-d8 100			Iodomethane	<1.0	1.0	
1,2-Dichloroethane-d4 95 % Recovery Toluene-d8 100		Surranta Pagavarian				
Toluene-d8 100		Surrogate necoveries	1,2-Dichloroethane-d4	95		% Recovery
4-Bromofluorobenzene 102						
			4-Bromofluorobenzene	102		

Site Depth Lab No. Methodology Analysis Results Units

Analyzed

	QUALITY A	ASSURANCE DATA				
Laboratory Reagent Report	EPA Method 8270				B O	6-28-95
Laboratory Reagent Report	Er A Wethod 3270	Acenaphthene	<10	unit	- 0	5-28-95
		Acenaphthylene	<10	<i>µ</i> g/L		
		Anthracene	<10			
		Azobenzene	<10			
		Benzidine	< 20			
		Benzo(a)Anthracene	<10			
		Benzo(b)fluoranthene	<10			
		Benzo(k)fluoranthene	<10			
		Benzo(g,h,i)perylene	<10			
		Benzo(a)pyrene	<10			
		4-Bromophenyl-phenylether	<10			
		Butylbenzylphthalate	<10			
		4-Chloro-3-Methylphenol	<10			
		bis(-2-Chloroethoxy)Methane	<10			
		bis(-2-Chloroethyl)Ether	<10			
		bis(2-Chloroisopropyl)ether	<10			
		2-Chloronaphthalene 2-Chlorophenol	<10 <10			
		4-Chlorophenol	< 10			
		4-Chlorophenyl-phenylether	<10			
		Chrysene	<10			
		Dibenzo(a,h)anthracene	<10			
		1,2-Dichlorobenzene	< 10			
		1,3-Dichlorobenzene	<10			
		1,4-Dichlorobenzene	< 10			
		3,3-Dichlorobenzidine	< 20			
		2,4-Dichlorophenol	< 10			
		Diethylphthalate	2.9J			
		Dimethyl Phthalate	<10			
		2,4-Dimethylphenol	< 10			
		Di-n-Butylphthalate	<10			
		4,6-Dinitro-2-methylphenol	< 50			
		2,4-Dinitrophenol 2,4-Dinitrotoluene	<50 <10			
		2,6-Dinitrotoluene	<10			
		Di-n-octyl Phthalate	<10			
		bis(2-ethylhexyl)Phthalate	1.7J			
		Fluoranthene	<10			
		Fluorene	<10			
		Hexachlorobenzene	< 10			
		Hexachlorobutadiene	<10			
	•	Hexachlorocyclopentadiene	<10			
		Hexachloroethane	< 10			
		Indeno(1,2,3-c,d)pyrene	<10			
		Isophorone	<10			
		1-Methylnaphthalene	<10			
		2-Methylphenol	<10			
		2-Methylphenol 4-Methylphenol/3-Methylphenol	<10 <10			
		Naphthalene	<10			
		Nitrobenzene	<10			
		2-Nitrophenol	<10			
		4-Nitrophenol	< 50			
		N-Nitrosodimethylamine	<10			
		N-nitroso-Di-n-propylamine	< 10			
		N-nitrosodiphenylamine	<10			
		Pentachlorophenol	< 50			
		Phenanthrene	<10			
		Phenol	<10			
		Pyrene Distriction	< 10			
		Pyridine	< 20			
		1,2,4-Trichlorobenzene	<10			
		2,4,5-Trichlorophenol	<10			
		2,4,6-Trichlorophenol	<10			
	Surrogate Recovery				QC Limits	
	•	2-fluorophenol	45	% Recovery	21-100	
		Phenol-d5	41	•	10-94	
		Nitrobenzene-d5	60		35-114	
		2-Fluorobiphenyl	57		43-116	
		2,4,6-Tribromophenol	61		10-123	
		Terphenyl-d14	74		33-141	

2,4,6-Tribromophenol Terphenyl-d14

61 74

Blank Spike	EPA Method 8270	•			QC Limits	8 06-28-95
		Acenaphthene	40*	% Recovery	46-118	
		4-Chloro-3-Methylphenol	52		23-97	
		2-Chlorophenol	42		27.123	
		1,4-Dichlorobenzene	30*		36-97	
	•	2,4-Dinitrotoluene	48		24-96	
		4-Nitrophenol	59		10-80	
		N-nitroso-Di-n-propylamine	41		41-116	
		Pentachlorophenol	45		9-103	
		Phenol	40		12-110	
		Pyrene	45		26-127	
		1,2,4-Trichlorobenzene	31 •		39-98	
	Surrogate Recovery				QC Limits	
		2-fluorophenol	39	% Recovery	21-100	
		Phenol-d5	39	•	10-94	
		Nitrobenzene-d5	53		35-114	
		2-Fluorobiphenyl	52		43-116	
		2,4,6-Tribromophenol	66		10-123	
		Terphenyl-d14	74		33-141	

^{*} Value outside QC limits.

⁶ Analysis performed at Energy Laboratories, Billings, Montana.



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Constant of James Machin Radian Corporation 3201 C. Street, Suite 405 Anchorage, AK 99503

EAFB

Project #612-001-31-37

Sampled: 06-25/26/27-95 Submitted: 06-27-95

July 17, 1995

95-38061-67

Site Depth Lab No. Methodology Analysis Results Units Analyze	ŧ
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Water Analysis

Influent-2

95-38061 8260 LONG

0-95

			. RH:06-30
1.1 Dieblessesbare	<10	<u>PQL</u> '	
1,1-Dichloroethene	<10	10	µg/L
Methylene Chloride trans-1,2-Dichloroethene	<10	10	
	<10	10	
1,1-Dichloroethane	<10	10	
2,2-Dichloropropane cis-1,2-Dichloroethene	790 ³	10	
		· -	
Bromochloromethane	<10	10	
Chloroform	<10	10	
1,1,1-Trichloroethane	<10	10 10	
Carbon Tetrachloride	<10	· -	
1,1-Dichloropropene	<10	10	
Benzene	270 3	10	
1,2-Dichloroethane	< 10	10	
Trichloroethene	19	10	
1, 2-Dichloropropane	<10	10	
Dibromomethane	<10	10	
Bromodichloromethane	< 10	10	
Trans-1,3-Dichloropropene	< 10	10	
Toluene	53	10	
cis-1,3-Dichloropropene	< 10	10	
1,1,2-Trichloroethane	< 10	10	
Tetrachloroethene	<10	10	
1,3-Dichloropropane	<10	10	
Dibromochloromethane	<10	10	
1,2-Dibromoethane	< 10	10	
Chlorobenzene	< 10	10	
1,1,1,2-Tetrachloroethane	< 10	10	
Ethylbenzene	110	10	
M + P Xylenes	260	10	
O-Xylene	18	10	
Styrene	<10	10	
Bromoform	<10	10	
Isopropylbenzene	15	10	
Bromobenzene	< 10	.10	
1,1,2,2-Tetrachloroethane	<10	10	
1, 2, 3-Trichloropropane	< 10	10	
n-Propyibenzene	17	10	
2-Chlorotokiene	<10	10	
4-Chlorotoluene	< 10	10	
1,3,5-Trimethylbenzene	31	10	
tert-Butylbenzene	<10	10	
1,2,4-Trimethylbenzene	160	10	
sec-Butylbenzene	<10	10	
1,3-Dichlorobenzene	<10	10	
1,4-Dichlorobenzene	<10	10	
p-isopropyitoluene	<10 4	10	
1,2-Dichlorobenzene	<10	10	
n-Butylbenzene	<10	10	
1,2-Dibromo-3-Chloropropane	<10	10	
1,2,4-Trichlorobenzene	<10	10	
Naphthalene	39	10	
Hexachlorobutadiene	<10	10	
	· -	-	

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Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-2 co	ont.	95-38061	8260 LONG				RH:06-30-9
						<u>PQL</u> 1	
				1,2,3-Trichlorobenzene	<10	10	μg/L
				Acetone	240	200	
				Methyl Ethyl Ketone	<100⁴	100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane	<10	10	
				Vinyl Chloride	<10	10	
				Bromomethane	< 10	10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	10	
				Carbon Disulfide	<10	10	
				Vinyl Acetate	<10	10	
				Methyl Isobutyl Ketone	< 100	100	
				2-Hexanone	<100	100	•
				Acrolein	< 100	100	
				Acrylonitrile	<100	100	
				Methyltertiary Butyl Ether	<10	10	
				lodomethane	<10	10	
		s	Surrogate Recoveries				
				1,2-Dichloroethane-d4	63 2	% Recovery	
				Toluene-d8	95		
				4-Bromofluorobenzene	98		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

² The high level of benzene present caused a suppression of the 1,2-dichloroethene-d4

³ Value derived from a 100X dilution.

⁴ Present but less than the PQL.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-3		95-38062	8260 LONG			PQL 1	PU-07 05 6
		00 00002	0200 20110	1,1-Dichloroethene	<10	10	RH:07-05-9
				Methylene Chloride	<10	10	μg/L
				trans-1,2-Dichloroethene	<10	10	
				1,1-Dichloroethane	<10	10	
				2,2-Dichloropropane	< 10	10	
				cis-1,2-Dichloroethene	1000 3	10	
				Bromochloromethane	< 10	10	
				Chloroform	<10	10	
				1,1,1-Trichloroethane Carbon Tetrachloride	<10 <10	10 10	
				1,1-Dichloropropene	<10	10	
				Benzene	360 '	10	
				1,2-Dichloroethane	<10	10	
	•			Trichloroethene	26	10	
				1,2-Dichloropropane	<10	10	
				Dibromomethane	<10	10	
				Bromodichloromethane	<10	10	
				Trans-1,3-Dichloropropene	< 10	10	
				Toluene	95	10 10	
				cis-1,3-Dichloropropene 1,1,2-Trichloroethane	<10 <10	10	
				Tetrachloroethene	<10	10	
				1,3-Dichloropropane	<10	10	•
				Dibromochloromethane	<10	10	
				1,2-Dibromoethane	< 10	10	
				Chlorobenzene	<10	10	
				1,1,1,2-Tetrachloroethane	<10	10	
				Ethylbenzene	130	10	
				M + P Xylenes	320 1	10	
				O-Xylene	42	10	
				Styrene Bromoform	<10 <10	10 10	
				Isopropylbenzene	21	10	
				Bromobenzene	<10	10	
				1,1,2,2-Tetrachloroethane	<10	10	
				1,2,3-Trichloropropane	< 10	10	
				n-Propyibenzene	28	10	
				2-Chlorotoluene	<10	10	
				4-Chiorotoluene	< 10	10	
				1,3,5-Trimethylbenzene	86	10	
				tert-Butylbenzene 1,2,4-Trimethylbenzene	< 10 180 ³	10 10	
				sec-Butylbenzene	<10 4	10	
				1,3-Dichlorobenzene	<10	10	
				1,4-Dichlorobenzene	< 10	10	
				p-Isopropyltoluene	18	10	
				1,2-Dichlorobenzene	<10	10	
				n-Butylbenzene	16	10	
				1,2-Dibromo-3-Chloropropane		10	
				1,2,4-Trichlorobenzene	< 10	10	
				Naphthalene	62	10	
				Hexachlorobutadiene 1,2,3-Trichlorobenzene	<10 <10	10 10	
				1,2,3-1richlorobenzene Acetone	330	200	
				Methyl Ethyl Ketone	120	100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane	<10	10	
				Vinyl Chloride	<10	10	
				Bromomethane	< 10	10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	. 10	
				Carbon Disulfide Vinyl Acetate	<10 <10	10 10	
				Methyl Isobutyl Ketone	<100	100	
				2-Hexanone	<100	100	
				Acrolein	< 100	100	
`				Acrylonitrile	<100	100	
				Methyltertiary Butyl Ether	<10	10	
				lodomethane	< 10	10	
		:	Surrogate Recoveries				
				1,2-Dichloroethane-d4	69 2	% Recovery	
				Toluene-d8	108		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

² The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

³ Value derived from a 100X dilution.

⁴ Present but less than the PQL.

Site [Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
luent-4		95-38063	8260 LONG	1 1 Diablass of an		Pal '	RH:07-05-9
				1,1-Dichloroethene Methylene Chloride	<10 <10	10 10	µg/L
		trans-1,2-Dichloroethene	<10	10			
		1,1-Dichloroethane	< 10	10			
				2,2-Dichloropropane cis-1,2-Dichloroethene	<10	10	
				Bromochloromethane	910 ' <10	10 10	
				Chloroform	<10	10	
				1,1,1-Trichloroethane	<10	10	
				Carbon Tetrachloride 1,1-Dichloropropene	<10 <10	10 10	
				Benzene	330 '	10	
				1,2-Dichloroethane	<10	10	
				Trichloroethene	31	10	
				1,2-Dichloropropane Dibromomethane	<10 <10	10 10	
				Bromodichloromethane	<10	10	
				Trans-1,3-Dichloropropene	<10	10	
				Toluene	130	10	
				cis-1,3-Dichloropropene 1,1,2-Trichloroethane	< 10	10	
				Tetrachloroethene	<10 <10	10 10	
				1,3-Dichloropropane	<10	10	
				Dibromochloromethane	<10	10	
				1,2-Dibromoethane Chlorobenzene	<10 <10	10 10	
				1,1,1,2-Tetrachloroethane	<10	10	
				Ethylbenzene	150	10	
				M + P Xylenes	390 '	10	
				O-Xylene Styrene	50 <10	10 10	
				Bromoform	<10	10	
				Isopropylbenzene	23	10	
				Bromobenzene 1,1,2,2-Tetrachloroethane	< 10	10	
				1,2,3-Trichloropropane	< 10 < 10	10 10	
				n-Propylbenzene	31	10	
				2-Chlorotoluene	< 10	10	
				4-Chlorotoluene 1,3,5-Trimethylbenzene	<10 84	10 10	
				tert-Butylbenzene	<10	10	
				1,2,4-Trimethylbenzene	220 3	10	
				sec-Butylbenzene	<10 4	10	
				1,3-Dichlorobenzene 1,4-Dichlorobenzene	<10 <10	10 10	
				p-Isopropyltoluene	17	10	
				1,2-Dichlorobenzene	< 10	10	
				n-Butylbenzene	15	10	
				1,2-Dibromo-3-Chloropropane 1,2,4-Trichlorobenzene	<10 <10	10 10	
				Naphthalene	52	10	
				Hexachlorobutadiene	< 10	10	
				1,2,3-Trichlorobenzene Acetone	< 10	10	
				Methyl Ethyl Ketone	400 130	200 100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane	< 10	10	
				Vinyl Chloride Bromomethane	<10 <10	10 10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	10	
				Carbon Disulfide Vinyl Acetate	<10 <10	10 10	
				Methyl Isobutyl Ketone	<100	100	
				2-Hexanone	< 100	100	
				Acrolein	< 100	100	
				Acrylonitrile Methyltertiary Butyl Ether	<100 <10	100 10	
				lodomethane	<10	10	
		Su	urrogate Recoveries				
				1,2-Dichloroethane-d4	62 '	% Recovery	
				Toluene-d8	102		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.
 Value derived from a 100X dilution.

⁴ Present but less than the PQL.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-5		95-38064	8260 LONG			POL 1	RH:07-05-9
				1,1-Dichloroethene	<10	10	μg/L
				Methylene Chloride trans-1,2-Dichloroethene	< 10 < 10	10 10	
				1,1-Dichloroethane	<10	10	
				2,2-Dichloropropane	<10	10	
				cis-1,2-Dichloroethene	790 '	10	
				Bromochloromethane	< 10	10	
				Chloroform	<10	10	
				1,1,1-Trichloroethane Carbon Tetrachloride	<10 <10	10 10	
				1,1-Dichloropropene	<10	10	
				Benzene	270 3	10	
				1,2-Dichloroethane	<10	10	
				Trichloroethene	29	10	
				1,2-Dichloropropane	<10	10	
				Dibromomethane	<10	10	
				Bromodichloromethane Trans-1,3-Dichloropropene	<10 <10	10 10	
				Toluene	130	10	
				cis-1,3-Dichloropropene	<10	10	
				1,1,2-Trichloroethane	< 10	10	
				Tetrachloroethene	<10	. 10	:
				1,3-Dichloropropane	<10	10	
				Dibromochloromethane 1,2-Dibromoethane	<10 <10	10 10	
				Chlorobenzene	<10	10	
				1,1,1,2-Tetrachloroethane	<10	10	
				Ethylbenzene	130	10	
				M + P Xylenes	380	10	
				0-Xylene Styrene	40 <10	10 10	
				Bromoform	<10	10	
				Isopropylbenzene	14	10	
				Bromobenzene	< 10	10	
				1,1,2,2-Tetrachloroethane	< 10	10	
				1,2,3-Trichloropropane	<10	10 10	
				n-Propylbenzene 2-Chlorotoluene	< 10 < 10	10	
				4-Chlorotoluene	<10	10	
				1,3,5-Trimethylbenzene	48	10	
			•	tert-Butylbenzene	<10	10	
				1,2,4-Trimethylbenzene sec-Butylbenzene	30 ,	10	
	,			1,3-Dichlorobenzene	<10 <10	10 10	
				1,4-Dichlorobenzene	<10	10	
				p-Isopropyltokiene	14	10	
				1,2-Dichlorobenzene	< 10	10	
•				n-Butylbenzene	10	10	
				1,2-Dibromo-3-Chloropropane 1,2,4-Trichlorobenzene	<10 <10	10 10	
				Naphthalene	64	10	
				Hexachlorobutadiene	< 10	10	
				1,2,3-Trichlorobenzene	< 10	10	
				Acetone	380	200	
				Methyl Ethyl Ketone Dichlorodifluoromethane	130 <10	100 10	
				Chloromethane	<10	10	
				Vinyl Chloride	<10	10	
				Bromomethane	<10	10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether Carbon Disulfide	<10 <10	10 10	
				Vinyl Acetate	<10	10	•
				Methyl Isobutyl Ketone	<100	100	
				2-Hexanone	<100	100	
				Acrol ei n	<100	100	
				Acrylonitrile	<100	100	
				Methyltertiary Butyl Ether	<10	10 10	
			Surrogate Recoveries	lodomethane	<10	10	
				1,2-Dichloroethane-d4	72 ²	% Recovery	
				Toluene-d8	105		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

² The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

³ Value derived from a 100X dilution.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
<i>(</i>)		25 2225					
nfluent-6 95-38065 8260 LONG		95-38065	8260 LONG	1.1 Diahlan da		POL '	RH:07-05-
	1,1-Dichloroethene	<10	10	μg/L			
				Methylene Chloride trans-1,2-Dichloroethene	<10 <10	10 10	
				1,1-Dichloroethane	< 10	10	
				2,2-Dichloropropane	<10	10	
				cis-1,2-Dichloroethene	810 '		
				Bromochloromethane	< 10	10	
				Chioroform	<10	10	
				1,1,1-Trichloroethane	<10	10	
				Carbon Tetrachloride 1,1-Dichloropropene	<10 <10	10 10	
				Benzene	260 '	10	
				1,2-Dichloroethane	<10	10	
				Trichloroethene	28	10	
				1,2-Dichloropropane	<10	10	
				Dibromomethane	< 10	10	
				Bromodichloromethane	< 10	10	
				Trans-1,3-Dichloropropene	< 10	10	
				Toluene cis-1,3-Dichloropropene	14 0 <10	10 10	
				1,1,2-Trichloroethane	<10	10	
				Tetrachloroethene	<10	10	
				1,3-Dichloropropane	<10	10	:
				Dibromochloromethane	< 10	10	
				1,2-Dibromoethane	< 10	10	
				Chlorobenzene	< 10	10	
				1,1,1,2-Tetrachioroethane	< 10	10	
		Ethylbenzene M + P Xylenes	130	10			
		0-Xylene	400 44	10 10			
	Styrene	< 10	10				
		Bromoform	< 10	10			
				Isopropyibenzene	18	10	
				Bromobenzene	< 10	10	
				1,1,2,2-Tetrachloroethane	< 10	10	
				1,2,3-Trichloropropane n-Propylbenzene	< 10	10	
				2-Chlorotoluene	17 <10	10 10	
				4-Chlorotoluene	<10	10	
				1,3,5-Trimethylbenzene	42	10	
				tert-Butylbenzene	< 10	10	
				1,2,4-Trimethylbenzene	15 '	10	
				sec-Butylbenzene	< 10	10,	
				1,3-Dichlorobenzene 1,4-Dichlorobenzene	<10 <10	10 10	
				p-IsoprapyItoluene	11	10	
				1,2-Dichlorobenzene	<10	10	
				n-Butylbenzene	<10 *	10	
				1,2-Dibromo-3-Chloropropane	< 10	10	
				1,2,4-Trichlorobenzene	< 10	10	
				Naphthalene Hexaphorobutadiana	57	10	
				Hexachlorobutadiene 1,2,3-Trichlorobenzene	< 10 < 10	10 10	
				Acetone	490	200	
				Methyl Ethyl Ketone	160	100	
				Dichlorodifluoromethane	< 10	10	
				Chloromethane	< 10	10	
				Vinyl Chloride	< 10	10	
				Bromomethane Chloroethane	< 10	10	
				Chloroethane Trichlorofluoromethane	<10 <10	10 10	
				2-Chloroethylvinylether	< 10	10	
				Carbon Disulfide	<10	10	
				Vinyl Acetate	<10	10	
				Methyl Isobutyl Ketone	< 100	100	
				2-Hexanone	< 100	100	
				Acrolein	< 100	100	
				Acrylonitrile Methyltertian, Surul Ether	<100	100	
				Methyltertiary Butyl Ether lodomethane	<10 <10	10 10	
		s	urrogate Recoveries	10 00 med an ic	< 10	10	
		-	•	1,2-Dichloroethane-d4	72 ²	% Recovery	
				Toluene-d8	98	•	
				4-Bromofluorobenzene	106		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

² The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

³ Value derived from a 100X dilution.

⁴ Present but less than the PQL.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-6D		95-38066	8260 LONG			PQL 1	RH:07-07-9
				1,1-Dichloroethene	<10	10	µg/L
				Methylene Chloride trans-1,2-Dichloroethene	<10 <10	10 10	
				1.1-Dichloroethane	<10	10	
				2,2-Dichloropropane	<10	10	
				cis-1,2-Dichloroethene	760 3	10	
				Bromochloromethane Chloroform	<10	10	
				1,1,1-Trichloroethane	<10 <10	10 10	
				Carbon Tetrachloride	<10	10	
				1,1-Dichloropropene	<10	10	
				Benzene	260 '	10	
				1,2-Dichloroethane Trichloroethene	<10	10	
				1,2-Dichloropropane	27 <10	10 10	
				Dibromomethane	<10	10	
				Bromodichloromethane	< 10	10	
				Trans-1,3-Dichloropropene	< 10	10	
				Toluene	130	10	
				cis-1,3-Dichloropropene 1,1,2-Trichloroethane	<10 <10	10 10	
				Tetrachloroethene	<10	10	
				1,3-Dichloropropane	<10	10	
				Dibromochloromethane	<10	10	
				1,2-Dibromoethane	<10	10	
				Chlorobenzene	<10	10	
				1,1,1,2-Tetrachloroethane Ethylbenzene	<10 120	10 10	
				M + P Xylenes	350	10	
				O-Xylene	42	10	
				Styrene	< 10	10	
				Bromoform	<10	10	
				Isopropylbenzene Bromobenzene	16	10	
				1,1,2,2-Tetrachloroethane	<10 <10	10 10	
				1,2,3-Trichloropropane	<10	10	
				n-Propylbenzene	16	10	
				2-Chlorotoluene	<10	10	
				4-Chlorotoluene 1,3,5-Trimethylbenzene	<10 41	10 10	
				tert-Butylbenzene	<10	10	
				1,2,4-Trimethylbenzene	200	10	
				sec-Butylbenzene	< 10	10	
				1,3-Dichlorobenzene	<10	10	
				1,4-Dichlorobenzene	< 10 11	10 10	
				p-Isopropyltoluene 1,2-Dichlorobenzene	<10	10	
				n-Butylbenzene	<10 4	10	
				1,2-Dibromo-3-Chloropropane	. <10	10	
				1,2,4-Trichlorobenzene	< 10	10	
				Naphthalene Hexachlorobutadiene	55	10	
				Hexachlorobutadiene 1,2,3-Trichlorobenzene	< 10 < 10	10 10	
				Acetone	450	200	
				Methyl Ethyl Ketone	150	100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane	<10	10	
				Vinyl Chloride Bromomethane	<10 <10	10 10	
				Chloroethane	<10	10	
		•		Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	10	
		Carbon Disulfide Vinyl Acetate	<10	10			
		Methyl Isobutyl Ketone	<10 <100	10 100			
				2-Hexanone	<100	100	
				Acrolein	<100	100	
				Acrylonitrile	< 100	100	
				Methyltertiary Butyl Ether	<10	10	
		•	Surrogate Recoveries	lodomethane	<10	10	
		•		1,2-Dichloroethane-d4	73 ²	% Recovery	
				Toluene-d8	105	•	
				4-Bromofluorobenzene	104		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

² The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

³ Value derived from a 100X dilution.

⁴ Present but less than the PQL.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
ıfluent-7		95-38067	8260 LONG			POL 1	RH:07-06-9
		00 00007	0200 20110	1,1-Dichloroethene	<10	10	μg/L
				Methylene Chloride	<10	10	r3/-
				trans-1,2-Dichloroethene	<10	10	
				1,1-Dichloroethane	<10	10	
				2,2-Dichloropropane cis-1,2-Dichloroethene	<10 8 10 '	10 10	
				Bromochloromethane	<10	10	
				Chloroform	<10	10	
				1,1,1-Trichloroethane	<10	10	
				Carbon Tetrachloride 1,1-Dichloropropene	<10 <10	10 10	
				Benzene	290 '	10	
				1,2-Dichloroethane	<10	10	
				Trichloroethene	32	10	
				1,2-Dichloropropane	<10	10	
				Dibromomethane Bromodichloromethane	<10 <10	10 10	
				Trans-1,3-Dichloropropene	<10	10	
				Toluene	180	10	
				cis-1,3-Dichloropropene	<10	10	
				1,1,2-Trichloroethane	<10	10	•
				Tetrachlorgethene	<10	10	,
				Dibromochloromethane	<10 <10	10 10	
		•		1,2-Dibromoethane	<10	10	
				Chlorobenzene	<10	10	
				1,1,1,2-Tetrachloroethane	<10	10	
				Ethylbenzene	130	10	
				M + P Xylenes O-Xylene	440 ³ 82	10 10	
				Styrene	<10	10	
				Bromoform	<10	10	
				Isopropylbenzene	15	10	
				Bromobenzene	<10	10	
				1,1,2,2-Tetrachloroethane 1,2,3-Trichloropropane	<10 <10	10 10	
				n-Propylbenzene	21	10	
				2-Chlorotoluene	< 10	10	
				4-Chlorotoluene	<10	10	
				1,3,5-Trimethylbenzene tert-Butylbenzene	54 <10	10 10	
				1,2,4-Trimethylbenzene	240 '	10	
				sec-Butylbenzene	<10	10	
				1,3-Dichlorobenzene	<10	10	
				1,4-Dichlorobenzene	<10	10	
				p-isopropyltoluene 1,2-Dichlorobenzene	14 <10	10 10	
				n-Butylbenzene	12	10	
				1,2-Dibromo-3-Chloropropane	<10	10	
				1,2,4-Trichlorobenzene	<10	10	
				Naphthalene Hexachlorobutadiene	86	10 10	
				1,2,3-Trichlorobenzene	<10 <10	10	
				Acetone	740	200	
				Methyl Ethyl Ketone	230	100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane Vinyl Chloride	<10 <10	10 10	
				Bromomethane	<10	10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	• 10	
				2-Chloroethylvinylether	<10	10	
				Carbon Disulfide Vinyl Acetate	<10 <10	10 10	
				Methyl isobutyl Ketone	<100	100	
				2-Hexanone	<100	100	
				Acrolein	<100	100	
				Acrylonitrile Methylterriany Butyl Ether	<100	100 10	
				Methyltertiary Butyl Ether lodomethane	<10 <10	10	
		5	Surrogate Recoveries	· · · · · · · · · · · · · · · · · · ·		. •	
			-	1,2-Dichloroethane-d4	68 2	% Recovery	
				Toluene-d8	99		

Sample diluted 10X at analysis due to the high level of target compounds present.
 The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.
 Value derived from a 100X dilution.
 Present but less than the PQL.

Trip Blank						
Mathylesc Chloride 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 2.1 - Zelphinrestheme 1.0 1.0 3.1 - Zelphinrestheme 1.0 1.0 3.1 - Zelphinrestheme 1.0 1.0 4.1 - Zelphinrestheme 1.0 1.0 5.1 - Zelphinrestheme 1.0 1.0 6.1 - Zelphinrestheme 1.0 1.0 7.1 - Zelphinrestheme 1.0 1.0 7.1 - Zelphinrestheme 1.0 1.0 7.1 - Zelphinrestheme 1.0 1.0 7.2 - Zelphinrestheme 1.0 1.0 7.3 - Zelphinrestheme 1.0 1.0 7.3 - Zelphinrestheme 1.0 1.0 7.4 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 7.5 - Zelphinrestheme 1.0 1.0 8 - Zelphinrestheme 1.0 1.0 8 - Zelphinrestheme 1.0 1.0 9 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinrestheme 1.0 1.0 1.1 - Zelphinre	Trip Blank	8260 LONG			PQL	RH:06-30-95
			1,1-Dichloroethene	< 1.0	1.0	µg/L
1,1-0-Globrosethane			•			
2,2-Dichloroprepare < 1.0 1.0 en-1,2-Dichloroprepare < 1.0 1.0 Berescolonomethere < 1.0 1.0 Classofter < 1.0 1.0 Classofter < 1.0 1.0 Classofter < 1.0 1.0 Classofter < 1.0 1.0 Classofter < 1.0 1.0 Classofter < 1.0 1.0 Classofter < 1.0 1.0 Li - Li - Li - Li - Li - Li - Li - Li -						
Commonship			-			
Bammodelhomenthane						
Chlorostem						
Carbon Tetra ablonide						
1,1-Disleropropone			1,1,1-Trichloroethane	<1.0	1.0	
Benzene				< 1.0	1.0	
1,2 Diselsoreshane						
TideNoreshare						
1,2 Dichloropropage Dichloromorbitance 1,0 1,0 Bernodichibromethane 1,0 1,0 Taran-1,3 Dichloropropage 1,0 1,0 Tolunam 1,1,2 Trichloropropage 1,0 1,0 Tolunam 1,1,2 Trichloropropage 1,0 1,0 Tolunam 1,1,2 Trichloropropage 1,0 1,0 Teta-shirosethane 1,0 1,0 1,3 Dichloropropage 1,0 1,0 1,3 Dichloropropage 1,0 1,0 1,2 Dichromorbitane 1,0 1,0 1,2 Dichromorbitane 1,0 1,0 1,1,2 Tricatedirorethane 1,0 1,0 1,1,2 Tricatedirorethane 1,0 1,0 1,1,2 Tricatedirorethane 1,0 1,0 1,1,2 Tricatedirorethane 1,0 1,0 1,0 Dichromorbitane 1,0 1,0 1,1,2 Tricatedirorethane 1,0 1,0 1,0 Dichromorbitane 1,0 1,0 1,1,2 Tricatedirorethane 1,0 1,0 1,2 Tricatedirorethane 1,0 1,0 1,3 Trimethylacateae 1,0 1,0 1,4 Trindshiphacateae 1,0 1,0 1,5 Dichloroberateae 1,0 1,0 1,6 Dichloroberateae 1,0 1,0 1,7 Dichloroberateae 1,0 1,0 1,8 Dichloroberateae 1,0 1,0 1,9 Dichloroberateae 1,0 1,0 1,0 Dichloroberateae 1,0 1,0 1,1,1,2 Trichloroberateae 1,0 1,0 1,1,2 Dichloroberateae 1,0 1,0 1,2 Dichloroberateae 1,0 1,0 1,2 Dichloroberateae 1,0 1,0 1,						
Dibromomethene						
Bennodichloremethene			The state of the s			
Tolune			Bromodichloromethane			
Total						
1.1.2-Trichlorosthane						•
Tetrachioratheme						
1.3-Dichloropepane						•
Dibromochloromethane						
1,2 Dibromothane			. ,			
Chlorebenzene						
Ethythoenzene				-		4
M + P Xylenes			1,1,1,2-Tetrachloroethane	<1.0	1.0	
C-Xylene			·			
Styrene						
Bromeform			•			
Isoprosylbenzene			•			
Brombenzene						
1,2,3-Trichloropropane						
n-Propylbenzene			1,1,2,2-Tetrachloroethane	< 1.0	1.0	
2-Chlorotoluene			1,2,3-Trichloropropane			
4-Chlorotoluene						
1,3,5-Trimethylbenzene						
tert-Butlythenzene						
1,2,4-Trimethylbenzene						
Sec-Butylbenzene C1.0						
1,4-Dichlorobenzene				< 1.0	1.0	
P-isopropyltoluene				< 1.0	1.0	
1,2-Dichlorobenzene			-			
n-Butylbenzene < 1.0 1.0 1,2-Dibromo-3-Chloropropane < 1.0 1.0 1,2-Erichlorobenzene < 1.0 1.0 Naphthalene < 1.0 1.0 Naphthalene < 1.0 1.0 1,0 1,0 1,0 1,0 1,0 1,0 1,0 1,0 1,0 1,						
1,2-Dibromo-3-Chloropropane <1.0						
1,2,4-Trichlorobenzene			•			
Hexachlorobutadiene						
1,2,3-Trichlorobenzene			Naphthalene	< 1.0	1.0	
Acetone						
Methyl Ethyl Ketone						
Dichlorodifluoromethane						
Chloromethane						
Vinyl Chloride						
Chloroethane						
Trichlorofluoromethane				< 1.0	1.0	
2-Chloroethylvinylether						
Carbon Disulfide						
Vinyl Acetate						
Methyl Isobutyl Ketone < 10						
2-Hexanone						
Acrylonitrile						
Methyltertiary Butyl Ether < 1.0 1.0 lodomethane < 1.0 1.0 Surrogate Recoveries 1,2-Dichloroethane-d4 90 % Recovery Toluene-d8 103						
lodomethane <1.0 1.0 Surrogate Recoveries 1,2-Dichloroethane-d4 90 % Recovery Toluene-d8 103			•			
Surrogate Recoveries 1,2-Dichloroethane-d4 90 % Recovery Toluene-d8 103						
1,2-Dichloroethane-d4 90 % Recovery Toluene-d8 103		Surrogate Recoveries	lodomethane	<1.0	1.0	
Toluene-d8 103		Santagate necovenes	1,2-Dichloroethane-d4	90	% Recovery	
4-Bromofluorobenzene 104						
			4-Bromofluorobenzene	104		

QUALITY ASSURANCE DATA

QUALITY ASSURANCE DATA								
Method Blank	8260 LONG			POL	RH:06-30-95			
mothed Blank		1,1-Dichloroethene	<1.0	1.0	μg/L			
	•	Methylene Chloride	< 1.0	1.0	pyrc			
	•	trans-1,2-Dichloroethene	< 1.0	1.0				
		1,1-Dichloroethane	< 1.0	1.0				
		2,2-Dichloropropane	< 1.0	1.0				
		cis-1,2-Dichloroethene	<1.0	1.0				
		Bromochloromethane	<1.0	1.0				
		Chloroform 1,1,1-Trichloroethane	<1.0 <1.0	1.0 1.0				
		Carbon Tetrachloride	<1.0	1.0				
		1,1-Dichloropropene	<1.0	1.0				
		Benzene	< 1.0	1.0				
		1,2-Dichloroethane	< 1.0	1.0				
		Trichloroethene	< 1.0	1.0				
		1,2-Dichloropropane	< 1.0	1.0				
		Dibromomethane	<1.0	1.0				
		Bromodichloromethane	< 1.0	1.0				
		Trans-1,3-Dichloropropene	<1.0	1.0	t			
		Toluene	<1.0	1.0				
		cis-1,3-Dichloropropene	<1.0	1.0				
		1,1,2-Trichloroethane Tetrachloroethene	<1.0 <1.0	1.0 1.0				
		1,3-Dichloropropane	<1.0	1.0				
		Dibromochloromethane	< 1.0	1.0				
		1,2-Dibromoethane	<1.0	1.0				
		Chlorobenzene	< 1.0	1.0				
		1,1,1,2-Tetrachloroethane	< 1.0	1.0				
		Ethylbenzene	< 1.0	1.0				
		M+P Xylenes	< 1.0	1.0				
		O-Xylene	< 1.0	1.0				
		Styrene Bromoform	<1.0 <1.0	1.0 1.0				
		Isopropylbenzene	< 1.0	1.0				
		Bromobenzene	<1.0	1.0				
		1,1,2,2-Tetrachloroethane	<1.0	1.0				
		1,2,3-Trichloropropane	< 1.0	1.0				
		n-Propylbenzene	< 1.0	1.0				
		2-Chlorotoluene	< 1.0	1.0				
		4-Chlorotoluene	< 1.0	1.0				
		1,3,5-Trimethylbenzene	<1.0 <1.0	1.0 1.0				
		tert-Butylbenzene 1,2,4-Trimethylbenzene	<1.0	1.0				
		sec-Butylbenzene	< 1.0	1.0				
		1,3-Dichlorobenzene	< 1.0	1.0				
		1,4-Dichlorobenzene	< 1.0	1.0				
		p-isopropyltoluene	< 1.0	1.0				
		1,2-Dichlorobenzene	< 1.0	1.0				
		n-Butylbenzene	< 1.0	1.0				
		1,2-Dibromo-3-Chloropropane	< 1.0	1.0				
		1,2,4-Trichlorobenzene Naphthalene	<1.0 <1.0	1.0 1.0				
		Hexachlorobutadiene	< 1.0	1.0				
		1,2,3-Trichlorobenzene	< 1.0	1.0				
		Acetone	< 20	20				
		Methyl Ethyl Ketone	< 10	10				
		Dichlorodifluoromethane	<1.0	1.0				
		Chloromethane	< 1.0	1.0				
		Vinyl Chloride	< 1.0	1.0				
		Bromomethane Chloroethane	< 1.0 < 1.0	1.0 1.0				
		Trichlorofluoromethane	< 1.0	1.0				
		2-Chloroethylvinylether	<1.0	1.0				
		Carbon Disulfide	< 1.0	1.0				
		Viny! Acetate	< 1.0	1.0				
		Methyl Isobutyl Ketone	< 10	10				
		2-Hexanone	< 10	10				
		Acrolein	< 10	10				
		Acrylonitrile	< 10	10				
		Methyltertiary Butyl Ether lodomethane	<1.0 <1.0	1.0 1.0				
	Surrogate Recoveri		×1.0	1.0				
		1.2 Niekland Abara 14	100 0					

1,2-Dichloroethane-d4

Toluene-d8
4-Bromofluorobenzene

100

106 98 % Recovery

QUALITY ASSURANCE DATA

	QUALIT	Y ASSURANCE DATA	1		
Method Blank	8260 LONG			PQL	RH:07-05-95
		1,1-Dichloroethene	< 1.0	1.0	μg/L
		Methylene Chloride	<1.0	1.0	73
		trans-1,2-Dichloroethene	< 1.0	1.0	
		1,1-Dichloroethane	<1.0	1.0	
		2,2-Dichloropropane	< 1.0	1.0	
		cis-1,2-Dichloroethene	< 1.0	1.0	•
		Bromochioromethane	<1.0	1.0	
		Chloroform 1,1,1-Trichloroethane	<1.0	1.0	
		Carbon Tetrachloride	<1.0 <1.0	1.0	
		1,1-Dichloropropene	<1.0	1.0 1.0	
		Benzene	<1.0	1.0	
		1,2-Dichloroethane	<1.0	1.0	
		Trichloroethene	< 1.0	1.0	
		1,2-Dichloropropane	< 1.0	1.0	
		Dibromomethane	< 1.0	1.0	
		Bromodichloromethane	< 1.0	1.0	
		Trans-1,3-Dichloropropene	< 1.0	1.0	÷
		Taluene	< 1.0	1.0	
		cis-1,3-Dichloropropene	<1.0	1.0	
		1,1,2-Trichloroethane Tetrachloroethene	<1.0	1.0	
		1,3-Dichloropropane	<1.0 <1.0	1.0	
		Dibromochloromethane	<1.0	1.0 1.0	
		1,2-Dibromoethane	< 1.0	1.0	
		Chiorobenzene	<1.0	1.0	
		1,1,1,2-Tetrachloroethane	< 1.0	1.0	
		Ethylbenzene	< 1.0	1.0	
		M+P Xylenes	< 1.0	1.0	
		O-Xylene	<1.0	1.0	
	•	Styrene	< 1.0	1.0	
		Bromoform	< 1.0	1.0	
		isopropylbenzene	< 1.0	1.0	
		Bromobenzene 1,1,2,2-Tetrachloroethane	<1.0 <1.0	1.0 1.0	
	•	1,2,3-Trichloropropane	< 1.0	1.0	
		n-Propylbenzene	< 1.0	1.0	
		2-Chlorotoluene	< 1.0	1.0	
		4-Chlorotoluene	< 1.0	1.0	
		1,3,5-Trimethylbenzene	< 1.0	1.0	
		tert-Butylbenzene	< 1.0	1.0	
		1,2,4-Trimethylbenzene	< 1.0	1.0	
		sec-Butylbenzene	<1.0	1.0	
		1,3-Dichlorobenzene 1,4-Dichlorobenzene	<1.0	1.0	
		p-isopropyltoluene	<1.0 <1.0	1.0 1.0	
		1,2-Dichlorobenzene	< 1.0	1.0	
		n-Butylbenzene	<1.0	1.0	
		1,2-Dibromo-3-Chloropropane	<1.0	1.0	
		1,2,4-Trichlorobenzene	< 1.0	1.0	
		Naphthalene	< 1.0	1.0	
		Hexachlorobutadiene	< 1.0	1.0	
		1,2,3-Trichlorobenzene	< 1.0	1.0	
		Acetone	< 20	20	
		Methyl Ethyl Ketone	<10	, 10	
		Dichlorodifluoromethane Chloromethane	<1.0 <1.0	1.0	
	•	Vinyl Chloride	<1.0	1.0 1.0	
		Bromomethane	<1.0	1.0	
		Chloroethane	<1.0	1.0	
		Trichlorofluoromethane	< 1.0	1.0	
		2-Chloroethylvinylether	< 1.0	1.0	
		Carbon Disulfide	< 1.0	1.0	
		· Vinyl Acetate	< 1.0	1.0	
		Methyl Isobutyl Ketone	< 10	10	
		2-Hexanone	<10	10	
		Acrolein	<10	10	
		Acrylonitrile	<10	10	
		Methyltertiary Butyl Ether lodomethane	<1.0	1.0	
	Surrogate Recoveries	louomethane	<1.0	1.0	
		1,2-Dichloroethane-d4	89	% Recovery	
		Toluene-d8	98		
		4-Bramofluorabenzene	100		

Methodology Site Depth Lab No. Analysis Results Units Analyzed

QUALITY ASSURANCE DATA									
Method Blank	8260 LONG			201	-				
Wethod Dialik	8200 LONG	1,1-Dichloroethene	<1.0	<u>PQL</u> 1.0	RH:07-06-95				
		Methylene Chloride	<1.0	1.0	µg/L				
		trans-1,2-Dichloroethene	<1.0	1.0					
		1,1-Dichloroethane	< 1.0	1.0					
		2,2-Dichloropropane	<1.0	1.0					
		cis-1,2-Dichloroethene	<1.0	1.0					
		Bromochloromethane	<1.0	1.0					
		Chloroform	<1.0	1.0					
		1,1,1-Trichloroethane	<1.0	1.0					
		Carbon Tetrachloride	<1.0	1.0					
		1,1-Dichloropropene	<1.0	1.0					
		Benzene 1,2-Dichloroethane	<1.0	1.0					
		Trichloroethene	<1.0 <1.0	1.0 1.0					
		1,2-Dichloropropane	<1.0	1.0					
		Dibromomethane	<1.0	1.0					
		Bromodichloromethane	<1.0	1.0					
		Trans-1,3-Dichloropropene	<1.0	1.0					
		Toluene	<1.0	1.0	•				
		cis-1,3-Dichloropropene	<1.0	1.0					
		1,1,2-Trichloroethane	<1.0	1.0					
		Tetrachloroethene	<1.0	1.0					
		1,3-Dichloropropane	<1.0	1.0					
		Dibromochloromethane	<1.0	1.0					
		1,2-Dibromoethane	<1.0	1.0					
		Chlorobenzene	<1.0	1.0					
		1,1,1,2-Tetrachloroethane	<1.0	1.0					
		Ethylbenzene M + P Xylenes	<1.0 <1.0	1.0 1.0					
		O-Xylene	<1.0	1.0					
		Styrene	<1.0	1.0					
		Bromoform	<1.0	1.0					
		Isopropylbenzene	<1.0	1.0					
		Bromobenzene	<1.0	1.0					
		1,1,2,2-Tetrachloroethane	<1.0	1.0					
		1,2,3-Trichloropropane	<1.0	1.0					
		n-Propylbenzene	<1.0	1.0					
		2-Chlorataluene	<1.0	1.0					
		4-Chlorotoluene	<1.0	1.0					
		1,3,5-Trimethylbenzene tert-Butylbenzene	<1.0 <1.0	1.0 1.0					
		1,2,4-Trimethylbenzene	<1.0	1.0					
		sec-Butylbenzene	<1.0	1.0					
		1,3-Dichlorobenzene	<1.0	1.0					
		1,4-Dichlorobenzene	< 1.0	1.0					
		p-lsopropyltoluene	<1.0	1.0					
		1,2-Dichlorobenzene	<1.0	1.0					
		n-Butylbenzene	<1.0	1.0					
		1,2-Dibromo-3-Chloropropane	<1.0	1.0					
		1,2,4-Trichlorobenzene	<1.0	1.0					
		Naphthalene	<1.0	1.0					
		Hexachlorobutadiene 1,2,3-Trichlorobenzene	<1.0 <1.0	1.0 1.0					
		Acetone	< 20	20					
		Methyl Ethyl Ketone	<10	10					
		Dichlorodifluoromethane	<1.0	1.0					
		Chloromethane	<1.0	1.0					
		Viny! Chloride	< 1.0	1.0					
		Bromomethane	<1.0	1.0					
		Chloroethane	<1.0	1.0					
		Trichlorofluoromethane	<1.0	1.0					
		2-Chloroethylvinylether	<1.0	1.0					
		Carbon Disulfide	<1.0	1.0					
		Vinyl Acetate	<1.0	1.0					
		Methyl isobutyl Ketone	< 10	10					
		2-Hexanone Acrolein	< 10 < 10	10 10					
		Acrylonitrile	< 10	10					
		Methyltertiary Butyl Ether	<1.0	1.0					
		lodomethane	<1.0	1.0					
	Surrogate Recoveries								
	-	1,2-Dichloroethane-d4	98	% Recovery					
		Toluene-d8	99						
		4-Bromofluorobenzene	100						

Toluene-d8
4-Bromofluorobenzene

100

QUALITY ASSURANCE DATA

ENERGY LABORATORIES, INC. RAPID CITY, SD

VOLATILE ORGANIC COMPOUNDS QUALITY ASSURANCE REPORT FORM

SAMPLE LOT_95-38061	
SAMPLE MATRIX Water	
EXTRACTION DATE -	
ANALYST <u>R.H.</u>	

MATRIX SPIKE / MATRIX SPIKE DUPLICATE DATA

Compound	Spike Added (µg)/L	Sample (µg)	Matrix Spike (μg)	Matrix Spike % Rec	Matrix Spike Duplicate (µg)	Matrix Spike Duplicate % Rec	% Difference (<u>Difference)</u> Average	QC Limits
1,1-Dichloroethene	1000	< 100	1160	116	1130	113	2.6	60-140%
Benzene	1000	270	1430	116	1400	113	2.6	60-140%
Trichloroethene	1000	<100	1260	126	1170	117	7.4	60-140%
Toluene	1000	< 100	1230	123	1240	124	0.81	60-140%
Chlorobenzene	1000	<100	1220	122	1220	122	0	60-140%



ENERGY LABORATORIES, INC.

Not being graft in 1

P.O. BOX 2470 • RAPID CITY, SD 57709 • PHONE (605) 342-1225 610 FARNWOOD STREET • RAPID CITY, SD 57701 • FAX (605) 342-1397

James Machin Radian Corporation 3201 C. Street, Suite 405 Anchorage, AK 99503

EAFB

Project #612-001-31-37

Sampled: 06-27/28/29-95 Submitted: 06-29-95

July 18, 1995

95-38191-95

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed

Water Analysis

Influent-8 95-38191 8260 LONG 95

			; RH:07-07-9
1,1-Dichloroethene	<10	<u>PQL</u> 1	
	<10		µg/L
Methylene Chloride trans-1,2-Dichloroethene	< 10	10 10	
1,1-Dichloroethane			
•	<10	10	
2,2-Dichloropropane	<10	10	
cis-1,2-Dichloroethene	700 ³	10	
Bromochloromethane	<10	10	
Chloroform	< 10	10	
1, 1, 1-Trichloroethane	<10	10	
Carbon Tetrachloride	<10	10	
1,1-Dichloropropene	<10	10	
Benzene	240 ³	10	
1,2-Dichloroethane	<10	10	
Trichloroethene	33	10	
1,2-Dichloropropane	< 10	10	
Dibromomethane	< 10	10	
Bromodichloromethane	<10	10	
Trans-1,3-Dichloropropene	<10	10	
Toluene	200	10	
cis-1,3-Dichloropropene	<10	10	•
1,1,2-Trichloreethane	< 10	10	
Tetrachloroethene	<10	10	
1,3-Dichloropropane	<10	10	
Dibromochloromethane	<10	10	
1, 2-Dibromoethane	<10	10	
Chlorobenzene	< 10	10	
1,1,1,2-Tetrachloroethane	<10	10	
Ethylbenzene	130	10	
M + P Xylenes	430 ³	10	
O-Xylene	80	10	
Styrene	<10	10	
Bromoform	<10	10	
Isopropylbenzene	15	10	
Bromobenzene	<10	10	0
1,1,2,2-Tetrachloroethane	<10	10	
1,2,3-Trichloropropane	<10	10	
n-Propylbenzene	21	10	
2-Chloratoluene	<10	10	
4-Chlorotoluene			
	<10	10	
1,3,5-Trimethylbenzene	54	10	
tert-Butylbenzene	<10	10	
1,2,4-Trimethylbenzene	320	10	
sec-Butylbenzene	<10	10	
1,3-Dichlorobenzene	<10	10	
1,4-Dichlorobenzene	<10	10	
p-Isopropyttoluene	16	10	
1, 2-Dichlorobenzene	<10	10	
n-Butylbenzene	13	10	
1,2-Dibromo-3-Chioropropane	<10	10	
1,2,4-Trichlorobenzene	<10	10	
Naphthalene	97	10	
Hexachlorobutadiene	<10	10	

Site	Depth	Lab No.	Methodology	Analysis	Results	Units		Analyzed
nfluent-8 c	ont.	95-38191	8260 LONG					RH:07-07-99
						POL '		
				1,2,3-Trichlorobenzene	<10	10		μg/L
				Acetone	730	200		 -
				Methyl Ethyl Ketone	220	100		
				Dichlorodifluoromethane	<10	10		
				Chloromethane	<10	10		
				Vinyl Chloride	<10	10		
				Bromomethane	<10	10		
				Chloroethane	<10	10		•
				Trichlorofluoromethane	<10	10		
				2-Chloroethylvinylether	<10	10		
				Carbon Disulfide	<10	10		
				Vinyl Acetate	<10	10		
				Methyl Isobutyl Ketone	<100	100		
				2-Hexanone	-100	100	÷	
				Acrolein	<100	100		
				Acrylonitrile	<100	100		
				Methyltertiary Butyl Ether	<10	10		
				Iodomethane	<10	10		
		\$	Surrogate Recoveries					
				1,2-Dichloroethane-d4	67 2	% Recovery		
				Toluene-d8	101			
				4-Bromofluorobenzene	99			

Sample diluted 10X at analysis due to the high level of target compounds present.

The high level of benzene present caused a suppression of the 1,2-dichloroethene-d4 Value derived from a 100X dilution.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
fluent-9		95-38192	8260 LONG			POL 1	RH:07-07-
				1,1-Dichloroethene	<10	10	µg/L
				Methylene Chloride trans-1,2-Dichloroethene	<10 <10	10 10	
				1,1-Dichloroethane	<10	10	
				2,2-Dichloropropane	<10	10	
				cis-1,2-Dichloroethene	860 3	10	
				Bromochloromethane	<10	10	
				Chloroform 1,1,1-Trichloroethane	<10 <10	10 10	
				Carbon Tetrachloride	<10	10	
				1,1-Dichloropropene	<10	10	
				Benzene	320 '	10	
				1,2-Dichloroethane	<10	10	
				Trichloroethene 1,2-Dichloropropane	44	10 10	
				Dibromomethane	<10 <10	10	
				Bromodichloromethane	<10	10	
				Trans-1,3-Dichloropropene	<10	10	
				Toluene	230 3	10	
				cis-1,3-Dichloropropene	<10	10	
				1,1,2-Trichloroethane Tetrachloroethene	<10	10 10	
				1 2 Disklassassassas	<10 <10		*
				Dibromochloromethane	<10	10	•
				1,2-Dibromoethane	< 10	10	
				Chlorobenzene	< 10	10	
				1,1,1,2-Tetrachloroethane	< 10	10	
				Ethylbenzene M + P Xylenes	160 900 ³	10 10	
				O-Xylene	110	10	
				Styrene	<10	10	
				Bromoform	< 10	10	
				Isopropylbenzene	20	10	
				Bromobenzene	< 10	10	
				1,1,2,2-Tetrachloroethane 1,2,3-Trichloropropane	<10 <10	10 10	
				n-Propylbenzene	28	10	
				2-Chlorotoluene	< 10	10	
				4-Chlorotoluene	<10	10	
				1,3,5-Trimethylbenzene	90	10	
				tert-Butylbenzene 1,2,4-Trimethylbenzene	< 10 1100 ³	10 10	
				sec-Butylbenzene	<10	10	
				1,3-Dichlorobenzene	< 10	10	
				1,4-Dichlorabenzene	< 10	10	
				p-Isopropyltoluene	19	10	
				1,2-Dichlorobenzene n-Butylbenzene	· <10	10	
				1,2-Dibromo-3-Chloropropane	16 <10	10 10	
				1,2,4-Trichlorobenzene	< 10	10	
				Naphthalene	130	10	
				Hexachlorobutadiene	< 10	10	
				1,2,3-Trichlorobenzene	< 10	10	
				Acetone Methyl Ethyl Ketone	1200 360	200 100	
				Dichlorodifluoromethane	< 10	10	
				Chloromethane	<10	10	
				Vinyl Chloride	< 10	10	
				Bromomethane	<10	10	
				Chloroethane	<10	10	
				Trichlorofluoromethane 2-Chloroethylvinylether	<10 <10	10 10	
		Carbon Disulfide	<10	10			
		Vinyl Acetate	< 10	10			
		Methyl Isobutyl Ketone	<100	100			
		2-Hexanone	< 100	100			
		Acrolein Acrylonitrile	<100 <100	100 100			
				Methyltertiary Butyl Ether	<100	100	
				lodomethane	<10	10	
		•	Surrogate Recoveries				
		•		1,2-Dichloroethane-d4	72 ²	% Recovery	
				Toluene-d8	96		
				4-Bromofluorobenzene	99		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

² The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

³ Value derived from a 100X dilution.

Site [Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
fluent-11		95-38194	8260 LONG	1.1 Diables sebase	-10	PQL '	RH:07-07-9
				1,1-Dichloroethene Methylene Chloride	<10 <10	10 10	₽g/L
				trans-1,2-Dichloroethene	<10	10	
				1,1-Dichloroethane	<10	10	
				2,2-Dichloropropane	< 10	10	
				cis-1,2-Dichloroethene	630 3	10	
				Bromochloromethane Chloroform	<10 <10	10 10	
				1,1,1-Trichloroethane	<10	10	
				Carbon Tetrachloride	<10	10	
				1,1-Dichloropropene	< 10	10	
				Benzene 1.2 Bi Ali	230 '	10	
				1,2-Dichloroethane Trichloroethene	< 10 39	10 10	
				1,2-Dichloropropane	<10	10	
		Dibromomethane	<10	10			
				Bromodichloromethane	<10	10	
		Trans-1,3-Dichloropropene	<10	10			
		Taluene cis-1,3-Dichloropropene	180 ⁾ <10	10 10			
				1,1,2-Trichloroethane	<10	10	
				Tetrachloroethene	<10	10	
				1,3-Dichloropropane	< 10	10	;
				Dibromochloromethane	<10	10	
		1,2-Dibromoethane	<10	10			
				Chlorobenzene 1,1,1,2-Tetrachloroethane	<10 <10	10 10	
		Ethylbenzene	130	10			
		M + P Xylenes	420 3	10			
				O-Xylene	120	10	
				Styrene Bromoform	< 10 < 10	10 10	
				Isopropylbenzene	17	10	
				Bromobenzene	<10	10	
				1,1,2,2-Tetrachloroethane		10	
				1,2,3-Trichloropropane	< 10	10	
				n-Propylbenzene 2-Chlorotoluene	23 <10	10 10	
				4-Chlorotoluene	<10	10	
				1,3,5-Trimethylbenzene	82	10	
				tert-Butylbenzene	<10	10	
				1,2,4-Trimethylbenzene sec-Butylbenzene	240 ¹ <10	10 10	
				1,3-Dichlorobenzene	<10	10	
				1,4-Dichlorobenzene	< 10	10	
				p-Isopropyltoluene	18	10	
				1,2-Dichlorobenzene	< 10	10	
				n-Butylbenzene 1,2-Dibromo-3-Chloropropane	16 • <10	10 10	
				1,2,4-Trichlorobenzene	<10	10	
				Naphthalene	110	10	
				Hexachlorobutadiene	< 10	10	
				1,2,3-Trichlorobenzene	< 10	10	
				Acetone Methyl Ethyl Ketone	1300 350	200 100	
				Dichlorodifluoromethane	< 10	10	
				Chloromethane	< 10	10	
				Vinyl Chloride	<10	10	
				Bromomethane Chloroethane	<10 <10	10 10	
	Trichlorofluoromethane	<10	10				
	2-Chloroethylvinylether	< 10	10				
	Carbon Disulfide	<10	10				
	Vinyl Acetate Mathyl leobutyl Katone	< 10	10 100				
				Methyl Isobutyl Ketone 2-Hexanone	<100 <100	100	
				Acrolein	<100	100	
				Acrylonitrile	< 100	100	
				Methyltertiary Butyl Ether	<10	10	
		e	urrogate Recoveries	lodomethane	<10	10	
		5	anogate necovenes	1,2-Dichloroethane-d4	68 4	% Recovery	
				Toluene-d8	96		

Sample diluted 10X at analysis due to the high level of target compounds present.

The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

Value derived from a 100X dilution.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
W930101	-Post Test	95-38195	8260 LONG			PQL 1	RH:07-05
				1,1-Dichloroethene	. <10	10	μg/L
			Methylene Chloride	<10	10	Pg. C	
				trans-1,2-Dichloroethene	<10	10	
				1,1-Dichloroethane	<10	10	
				2,2-Dichloropropane	<10	10	
				cis-1,2-Dichloroethene	1900 '	10	
				Bromochloromethane Chloroform	<10	10	
				1,1,3-Trichloroethane	<10	10	
				Carbon Tetrachloride	<10 <10	10 10	
				1,1-Dichloropropene	<10	10	
				Benzene	470 3	10	
				1,2-Dichloroethane	< 10	10	
				Trichloroethene	120	10	
				1,2-Dichloropropane	< 10	10	
				Dibromomethane	< 10	10	
				Bromodichloromethane	< 10	10	
				Trans-1,3-Dichloropropene	< 10	10	
				Toluene	220	10	
				cis-1,3-Dichloropropene	<10	10	
				1,1,2-Trichloroethane	< 10	10 10	
				Tetrachloroethene 1,3-Dichloropropane	20 <10	10	}
				Dibromochloromethane	<10	10	•
				1,2-Dibromoethane	<10	10	
				Chlorobenzene	< 10	10	
				1,1,1,2-Tetrachloroethane	< 10	10	
				Ethylbenzene	56	10	
				M + P Xylenes	1000 3	10	
				O-Xγlene	160 3	10	
				Styrene	< 10	10	
				Bromoform	< 10	10	
				Isopropylbenzene	14	10	
				Bromobenzene	< 10 < 10	10 10	
				1,1,2,2-Tetrachloroethane 1,2,3-Trichloropropane	< 10	10	
				n-Propylbenzene	<10	10	
				2-Chlorotoluene	<10	10	
				4-Chlorotoluene	< 10	10	
				1,3,5-Trimethylbenzene	210	10	
				tert-Butylbenzene	<10	10	
				1,2,4-Trimethylbenzene	480 3	10	
				sec-Butylbenzene	<10	10	
				1,3-Dichlorobenzene	< 10	10	
				1,4-Dichlorobenzene	< 10	10	
				p-Isopropyltoluene	37	10	
				1,2-Dichlorobenzene n-Butylbenzene	< 10	10	
				1,2-Dibromo-3-Chloropropane	17 e <10	10 10	
				1,2,4-Trichlorobenzene	<10	10	
				Naphthalene	140	10	
				Hexachlorobutadiene	<10	10	
				1,2,3-Trichlorobenzene	<10	10	
				Acetone	1300	200	
				Methyl Ethyl Ketone	350	100	
				Dichlorodifluoromethane	<10	10	
				Chloromethane	<10	10	
			,	Vinyl Chloride Bromomethane	<10 <10	10 10	
				Chloroethane	<10	10	
				Trichlorofluoromethane	<10	10	
				2-Chloroethylvinylether	<10	10	
				Carbon Disulfide	<10	10	
				Vinyl Acetate	<10	10	
				Methyl Isobutyl Ketone	<100	100	
				2-Hexanone	< 100	100	
				Acrolein	< 100	100	
				Acrylonitrile	<100	100	
		*		Methyltertiary Butyl Ether	<10	10	
				lodomethane	< 10	10	
			Surrogate Recoveries	1.2 0:44	n. '	o/ D	
				1,2-Dichloroethane-d4	61 2	% Recovery	
				Toluene-d8	103		

¹ Sample diluted 10X at analysis due to the high level of target compounds present.

Laboratory Manager Kurt R. Slentz_

The high level of benzene present caused a suppression of the 1,2-dichloroethane-d4.

Value derived from a 100X dilution.

	QUALITY	ASSURANCE DATA	١.		
Trip Blank	8260 LONG			801	
mp blank	0200 E0163	3 1 Disklassides	-110	PQL	RH:07-07-95
		1,1-Dichloroethene Methylene Chloride	<1.0	1.0	µg/L
		trans-1,2-Dichloroethene	<1.0 <1.0	1.0 1.0	
		1,1-Dichloroethane	< 1.0	1.0	
		2,2-Dichloropropane	<1.0	1.0	
		cis-1,2-Dichloroethene	<1.0	1.0	
		Bromochloromethane	<1.0	1.0	
		Chloroform	< 1.0	1.0	
		1,1,1-Trichloroethane	< 1.0	1.0	
		Carbon Tetrachloride	< 1.0	1.0	
		1,1-Dichloropropene	< 1.0	1.0	
		Benzene	< 1.0	1.0	
		1,2-Dichloroethane	< 1.0	1.0	
		Trichloroethene	< 1.0	1.0	
		1,2-Dichloropropane	< 1.0	1.0	
		Dibromomethane	< 1.0	1.0	
		Bromodichloromethane	< 1.0	1.0	
		Trans-1,3-Dichloropropene	<1.0	1,0	ž.
		Toluene	<1.0	1.0	
		cis-1,3-Dichloropropene	< 1.0	1.0	
		1,1,2-Trichloroethane	< 1.0	1.0	
		Tetrachloroethene 1,3-Dichloropropane	<1.0	1.0	
		Dibromochloromethane	<1.0 <1.0	1.0 1.0	•
		1,2-Dibromoethane	<1.0	1.0	
		Chlorobenzene	< 1.0	1.0	
		1,1,1,2-Tetrachloroethane	< 1.0	1.0	
		Ethylbenzene	< 1.0	1.0	
		M + P Xylenes	< 1.0	1.0	
		O-Xylene	< 1.0	1.0	
		Styrene	< 1.0	1.0	
		Bromoform	< 1.0	1.0	
		Isopropyibenzene	< 1.0	1.0	
		Bromobenzene	< 1.0	1.0	
		1,1,2,2-Tetrachloroethane	< 1.0	1.0	
		1,2,3-Trichloropropane	< 1.0	1.0	
		n-Propylbenzene	< 1.0	1.0	
		2-Chlorotoluene 4-Chlorotoluene	<1.0 <1.0	1.0	
		1,3,5-Trimethylbenzene	< 1.0	1.0 1.0	
		tert-Butylbenzene	< 1.0	1.0	
		1,2,4-Trimethylbenzene	<1.0	1.0	
		sec-Butylbenzene	< 1.0	1.0	
		1,3-Dichlorobenzene	<1.0	1.0	
		1,4-Dichlorobenzene	< 1.0	1.0	
		p-Isopropyltoluene	< 1.0	1.0	
		1,2-Dichlorobenzene	< 1.0	1.0	
		n-Butylbenzene	< 1.0	1.0	
		1,2-Dibromo-3-Chloropropane	< 1.0	1.0	
		1,2,4-Trichlorobenzene	< 1.0	1.0	
		Naphthalene	< 1.0	1.0	
		Hexachlorobutadiene	< 1.0	1.0	
		1,2,3-Trichlorobenzene	< 1.0	1.0	
		Acetone Methyl Ethyl Ketone	< 20	20	
		Dichlorodifluoromethane	<10 <1.0	10 1.0	
		Chloromethane	< 1.0	1.0	
		Vinyl Chloride	< 1.0	1.0	
		Bromomethane	<1.0	1.0	
		Chloroethane	< 1.0	1.0	
		Trichlorofluoromethane	< 1.0	1.0	
		2-Chloroethylvinylether	< 1.0	1.0	
		Carbon Disulfide	< 1.0	1.0	
		Vinyl Acetate	< 1.0	1.0	
		Methyl Isobutyl Ketone	<10	10	
		2-Hexanone	< 10	10	
		Acrolein	< 10	10	
		Acrylonitrile	< 10	10	
		Methyltertiary Butyl Ether	< 1.0	1.0	
	¢ n	lodomethane	< 1.0	1.0	
	Surrogate Recoveries	1,2-Dichloroethane-d4	101	O D	
		Toluene-d8	104 94	% Recovery	•
		4-Bromofluorobenzene	93		

QUALITY ASSURANCE DATA

	QUALITY	ASSURANCE DATA			
Method Blank	8260 LONG			PQL	PU.07.07.07
	3233 2233	1,1-Dichloroethene	<1.0	1.0	RH:07-07-95
		Methylene Chloride	<1.0	1.0	μg/L
		trans-1,2-Dichloroethene	<1.0	1.0	
		1,1-Dichloroethane	< 1.0	1.0	
		2,2-Dichloropropane	< 1.0	1.0	
	•	cis-1,2-Dichloroethene	< 1.0	1.0	
		Bromochloromethane	< 1.0	1.0	
		Chloroform	< 1.0	1.0	•
		1,1,1-Trichloroethane	< 1.0	1.0	
		Carbon Tetrachloride	<1.0	1.0	
		1,1-Dichloropropene	<1.0	1.0	
		Benzene 1,2-Dichloroethane	< 1.0	1.0	
		Trichloroethene	<1.0	1.0	
		1,2-Dichloropropane	<1.0 <1.0	1.0 1.0	
		Dibromomethane	< 1.0	1.0	
		Bromodichloromethane	< 1.0	1.0	
		Trans-1,3-Dichloropropene	<1.0	1.0	
		Toluene ± ±	< 1.0	1.0	<i>?</i>
		cis-1,3-Dichloropropene	< 1.0	1.0	
		1,1,2-Trichloroethane	< 1.0	1.0	
		Tetrachloroethene	< 1.0	1.0	
		1,3-Dichloropropane	< 1.0	1.0	
		Dibromochloromethane	< 1.0	1.0	
		1,2-Dibromoethane	< 1.0	1.0	
		Chlorobenzene	< 1.0	1.0	
		1,1,1,2-Tetrachloroethane	< 1.0	1.0	
		Ethylbenzene	< 1.0	1.0	
		M+P Xylenes	<1.0	1.0	
		0-Xylene Styrene	< 1.0	1.0	
	•	Bromoform	<1.0 <1.0	1.0 1.0	
		Isopropylbenzene	<1.0	1.0	
		Bromobenzene	<1.0	1.0	
		1,1,2,2-Tetrachloroethane	<1.0	1.0	
		1, 2, 3-Trichloropropane	<1.0	1.0	
		n-Propylbenzene	< 1.0	1.0	
		2-Chlorotoluene	< 1.0	1.0	
		4-Chlorotoluene	< 1.0	1.0	
		1,3,5-Trimethylbenzene	< 1.0	1.0	
		tert-Butylbenzene	< 1.0	1.0	
		1,2,4-Trimethylbenzene	< 1.0	1.0	
		sec-Butylbenzene	<1.0	1.0	
		1,3-Dichlorobenzene 1,4-Dichlorobenzene	<1.0	1.0	
		p-isopropyitoluene	<1.0 <1.0	1.0 1.0	
		1,2-Dichlorobenzene	<1.0	1.0	
		n-Butylbenzene	< 1.0	1.0	
		1,2-Dibromo-3-Chloropropane	< 1.0	1.0	
		1,2,4-Trichlorobenzene	< 1.0	1.0	
		Naphthalene	< 1.0	1.0	
		Hexachlorobutadiene	< 1.0	1.0	
		1,2,3-Trichlorobenzene	<1.0	1.0	
		Acetone	< 20	20	
		Methyl Ethyl Ketone	< 10	10	
		Dichlorodifluoromethane	<1.0	1.0	
		Chloromethane	<1.0	1.0	
		Vinyl Chloride Bromomethane	< 1.0	1.0	
		Chloroethane .	<1.0	1.0	
		Trichlorofluoromethane	<1.0 <1.0	1.0 1.0	
	ı	2-Chloroethylvinylether	< 1.0	1.0	
		Carbon Disulfide	<1.0	1.0	
		Vinyl Acetate	< 1.0	1.0	
		Methyl Isobutyl Ketone	<10	10	
		2-Hexanone	<10	10	
i		Acrolein	<10	10	
		Acrylonitrile	< 10	10	
		Methyltertiary Butyl Ether	< 1.0	1.0	
		lodomethane	< 1.0	1.0	
	Surrogate Recoveries				
		1,2-Dichloroethane-d4	96	% Recovery	
		Toluene-d8	106		

Toluene-d8

4-Bromofluorobenzene

106

102

Analysis Results Units Analyzed Site Depth Lab No. Methodology

QUALITY ASSURANCE DATA

				201	
Method Blank	8260 LONG	1.1.00-6-1	-10	<u>PQL</u> 1.0	RH:07-10-95
		1,1-Dichloroethene Methylene Chloride	<1.0 <1.0	1.0	μg/L
		trans-1,2-Dichloroethene	<1.0	1.0	
		1,1-Dichloroethane	<1.0	1,0	
		2,2-Dichloropropane	<1.0	1.0	
		cis-1,2-Dichloroethene	<1.0	1.0	
		Bromochloromethane	<1.0	1.0	
		Chloroform 1,1,1-Trichloroethane	<1.0 <1.0	1.0 1.0	
		Carbon Tetrachloride	<1.0	1.0	
		1,1-Dichloropropene	<1.0	1.0	
		Benzene	<1.0	1.0	•
		1,2-Dichloroethane	<1.0	1.0	
		Trichloroethene	< 1.0	1.0	
		1,2-Dichloropropane	<1.0	1.0	
		Dibromomethane	<1.0	1.0	
		Bromodichloromethane Trans-1,3-Dichloropropene	<1.0 <1.0	1.0 1.0	}
		Toluene	<1.0	1.0	
		cis-1,3-Dichloropropene	<1.0	1.0	
		1,1,2-Trichloroethane	< 1.0	1.0	
		Tetrachloroethene	<1.0	1.0	
		1,3-Dichloropropane	<1.0	1.0	
		Dibromochloromethane	<1.0	1.0	
		1,2-Dibromoethane	<1.0	1.0	
		Chlorobenzene	<1.0 <1.0	1.0 1.0	
		1,1,1,2-Tetrachloroethane Ethylbenzene	<1.0	1.0	
		M+P Xylenes	<1.0	1.0	
	,	O-Xylene	<1.0	1.0	
		Styrene	< 1.0	1.0	
		Bromoform	<1.0	1.0	
		Isopropylbenzene	<1.0	1.0	
		Bromobenzene 1,1,2,2-Tetrachloroethane	<1.0 <1.0	1.0 1.0	
		1,2,3-Trichloropropane	<1.0	1.0	
		n-Propylbenzene	<1.0	1.0	
		2-Chlorotoluene	<1.0	1.0	
:		4-Chlorotoluene	<1.0	1.0	
		1,3,5-Trimethylbenzene	<1.0	1.0	
		tert-Butylbenzene	<1.0 <1.0	1.0 1.0	
		1,2,4-Trimethylbenzene sec-Butylbenzene	<1.0	1.0	
		1,3-Dichlorobenzene	<1.0	1.0	
		1,4-Dichlorobenzene	< 1.0	1.0	
		p-isopropyltoluene	< 1.0	1.0	
		1,2-Dichlorobenzene	<1.0	1.0	
		n-Butylbenzene	<1.0 <1.0	1.0 1.0	
		1,2-Dibromo-3-Chloropropane 1,2,4-Trichlorobenzene	<1.0	1.0	
		Naphthalene	<1.0	1.0	
		Hexachlorobutadiene	<1.0	1.0	
		1,2,3-Trichlorobenzene	<1.0	1.0	
		Acetone	< 20	20	
		Methyl Ethyl Ketone Dichlorodifluoromethane	<10 <1.0	10 1.0	
		Chloromethane	<1.0	1.0	
		Vinyl Chloride	<1.0	1.0	
		Bromomethane	<1.0	1.0	
		Chloroethane	<1.0	1.0	
*		Trichlorofluoromethane	<1.0	1.0	
		2-Chloroethylvinylether	<1.0	1.0	
		Carbon Disulfide Vinyl Acetate	<1.0 <1.0	1.0 1.0	
		Methyl Isobutyl Ketone	<10	10	
		2-Hexanone	<10	10	
		Acrolein	<10	10	
		Acrylonitrile	<10	10	
		Methyltertiary Butyl Ether	<1.0	1.0	
	Surrogate Recoveries	lodomethane	<1.0	1.0	
	Surrogate necoveries	1,2-Dichloroethane-d4	89	% Recovery	
		Toluene-d8	97		
		4-Bromofluorobenzene	103		

QUALITY ASSURANCE DATA

ENERGY LABORATORIES, INC. RAPID CITY, SD

VOLATILE ORGANIC COMPOUNDS QUALITY ASSURANCE REPORT FORM

SAMPLE LOT SAMPLE MATRIX EXTRACTION DATE ANALYST

95-38191

Water

RH

MATRIX SPIKE / MATRIX SPIKE DUPLICATE DATA

Compound	Spike Added (µg)/L	Sample (µg)	Matrix Spike (μg)	Matrix Spike % Rec	Matrix Spike Duplicate (µg)	Matrix Spike Duplicate % Rec	% Difference (Difference) Average	QC Limits
1,1-Dichloroethene	1000	< 100	990	99	1030	103	4.0	60-140%
Benzene	1000	240	1290	105	1230	99	5.9	60-140%
Trichloroethene	1000	<100	1080	108	1060	106	1.9	60-140%
Toluene	1000	160	1320	116	1220	106	9.0	60-140%
Chlorobenzene	1000	<100	1130	113	1140	114	0.88	60-140%



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James Machin Radian Corporation 3201 C. Street, Suite 405 Anchorage, AK 99503

REVISED REPORT

EAFB

Project #612-001-31-37

Sampled: 06-29/30-95

July 21, 1995 95-38229-31

Submitted: 06-29-95

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed

Water Analysis

Influent-12 95-38229 **8260 LONG**

		PQL	RH:07-13-95
1,1-Dichloroethene	< 1.0	1.0	
Methylene Chloride	1.5	1.0	
trans-1,2-Dichloroethene	< 1.0	1.0	
1,1-Dichloroethane	<1.0	1,0	
2, 2-Dichloropropane	<1.0	1.0	
cis-1,2-Dichloroethene	63²	1.0	
Bromochioromethane	<1.0	1.0	
Chloroform	<1.0	1.0	
1,1,1-Trichloroethane	<1.0	1.0	
Carbon Tetrachloride	<1.0	1.0	
1,1-Dichloropropene	<1.0	1.0	
Benzene	7.5	1.0	
1.2-Dichloroethane	< 1.0	1.0	
Trichloroethene	5.0		
1, 2-Dichloropropane	< 1.0	1.0	
Dibromomethane		1.0	
Bromodichloromethane	< 1.0	1.0	
	<1.0	1.0	
Trans-1,3-Dichloropropene Toluene	< 1.0	1.0	
	11	1.0	
cis-1,3-Dichloropropene 1,1,2-Trichloroethane	< 1.0	1.0	
	< 1.0	1.0	
Tetrachloroethene	<1.03	1.0	
1,3-Dichloropropane	<1.0	1.0	
Dibromochloromethane	<1.0	1.0	
1,2-Dibromoethane	< 1.0	1.0	
Chlorobenzene	<1.0	1.0	
1,1,1,2-Tetrachioroethane	<1.0	1.0	
Ethylbenzene	11	1.0	
M + P Xylenes	46²	1.0	
0-Xylene	93	1.0	
Styrene	<1.0	1.0	
Bromoform	<1.0	1.0	
Isopropylbenzene	2.1	1.0	
Bromobenzene	< 1.0	1.0	
1,1,2,2-Tetrachioroethane	< 1.0	1.0	
1,2,3-Trichloropropane	<1.0	1.0	
n-Propylbenzene	2.6	1.0	
2-Chlorotoluene	<1.0	1.0	
4-Chlorotoluene	<1.0	1.0	
1,3,5-Trimethylbenzene	13	1.0	
tert-Butylbenzene	<1.0	1.0	
1,2,4-Trimethylbenzene	37²	1.0	
sec-Butylbenzene	1.1	1.0	
1,3-Dichlorobenzene	<1.0	1.0	
1,4-Dichlorobenzene	<1.0	1.0	
p-lsopropyltoluene	3.2	1.0	
1,2-Dichlorobenzene	<1.0	1.0	
n-Butylbenzene	2.2	1.0	
1,2-Dibromo-3-Chloropropane	<1.0	1.0	•
1,2,4-Trichlorobenzene	<1.0	1.0	
	•		
Naphthalene	17	1,0	

Site	Depth Lab No.	Methodology	Analysis	Results	Units .	Analyzed
Influent-12 95-38	95-38229	8260 LONG continu	ed		<u> POL</u>	RH:07-13-9! Units <u>µg/L</u>
			1, 2, 3-Trichlorobenzene	<1.0	1.0	
			Acetone	57	20	
			Methyl Ethyl Ketone	78	10	
			Dichlorodifluoromethane	<1.0	1.0	
			Chloromethane	< 1.0	1.0	
			Vinyl Chloride	< 1.0	1.0	
			Bromomethane	<1.0	1.0	
			Chloroethane	<1.0	1.0	
			Trichlorofluoromethane	<1.0	1.0	
			2-Chloroethylvinylether	<1.0	1.0	
			Carbon Disulfide	<1.0	1.0	
			Vinyl Acetate	<1.0	1.0	
			Methyl Isobutyl Ketone	<10	10	
			2-Hexanone	<10	10	
			Acrolein	<10	10	
			Acrylonitrile	<10	10	
			Methyltertiary Butyl Ether	<1.0	1.0	:
			lodomethane	<1.0	1.0	
		Surrogate Recoveries				
			1,2-Dichloroethane-d4	781	%	Recovery
			Toluene-d8	99		
			4-Bromofluorobenzene	110		

¹ Non-target compound sample matrix interference caused a supression of the 1,2-Dichloroethane-d4.

Value derived from a 10x dilution.
 Present but less than the Practical Quantitation Limit.

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-12D)		95-38230	8260 LONG			RH:07-13-9
						POL	Units µg/L
				1,1-Dichloroethene	<1.0	1.0	
				Methylene Chloride	1.8	1.0	
				trans-1, 2-Dichloroethene	<1.0	1.0	
				1,1-Dichloroethane	<1.0	1.0	
				2,2-Dichloropropane	<1.0	1.0	
				cis-1,2-Dichloroethene Bromochloromethane	63'	1.0	
				Chloroform	<1.0 <1.0	1.0 1.0	
			1	1,1,1-Trichloroethane	<1.0	1.0	
			'	Carbon Tetrachloride	<1.0	1.0	
				1,1-Dichloropropene	<1.0	1.0	
				Benzene	8.2	1.0	
				1,2-Dichloroethane	<1.0	1.0	
		•		Trichloroethene	6.0	1.0	
				1,2-Dichloropropane	<1.0	1.0	
				Dibromomethane	<1.0	1.0	
				Bromodichloromethane	<1.0	1.0	
				Trans-1,3-Dichloropropene	<1.0	1.0	
				Toluene	12	1.0	
				cis-1,3-Dichloropropene	<1.0	1.0	•
				1,1,2-Trichloroethane	<1.0	1.0	
				Tetrachloroethene	1.0	1.0	
				1,3-Dichloropropane Dibromochloromethane	<1.0 <1.0	1.0 1.0	
				1, 2-Dibromoethane	<1.0	1.0	
				Chlorobenzene	<1.0	1.0	
				1,1,1,2-Tetrachloroethane	<1.0	1.0	
				Ethylbenzene	13	1.0	
				M+P Xylenes	511	1.0	
				O-Xylene	10	1.0	
				Styrene	<1.0	1.0	
				Bromoform	<1.0	1.0	
				Isopropylbenzene	2.5	1.0	
				Bromobenzene	< 1.0	1.0	
				1,1,2,2-Tetrachloroethane	<1.0	1.0	
				1,2,3-Trichloropropane n-Propylbenzene	<1.0 3 .1	1.0 1.0	
				2-Chlorotoluene	<1.0	1.0	
				4-Chlorotoluene	<1.0	1.0	
				1,3,5-Trimethylbenzene	15	1.0	
			,	tert-Butylbenzene	<1.0	1.0	
				1, 2, 4-Trimethylbenzene	411	1.0	
				sec-Butylbenzene	1.2	1.0	
				1,3-Dichlorobenzene	<1.0	1.0	
				1,4-Dichlorobenzene	<1.0	1.0	
				p-isopropyitoluene	3.7	1.0	
				1,2-Dichlorobenzene	<1.0	1.0	
				n-Butylbenzene	2.5	1.0	
				1,2-Dibromo-3-Chloropropane		1.0	
				1,2,4-Trichlorobenzene Naphthalene	<1.0 19	1.0 1.0	

Site	Depth Lab No.	Methodology	Analysis	Results	Units	Analyzed
Influent-12D	95-38230	8260 LONG continue	ed		POL	RH:07-13-9! Units <i>µg/</i> L
						Omes Bare
			1,2,3-Trichtorobenzene	<1.0	1.0	
			Acetone	37	20	
			Methyl Ethyl Ketone	80	10	
			Dichlorodifluoromethane	<1.0	1.0	
			Chioromethane	<1.0	1.0	
			Vinyl Chloride	<1.0	1.0	
			Bromomethane	<1.0	1.0	
			Chloroethane	<1.0	1.0	
			Trichlorofluoromethane	< 1.0	1.0	
			2-Chloroethylvinylether	<1.0	1.0	
			Carbon Disulfide	< 1.0	1.0	·
			Viny! Acetate	< 1.0	1.0	
			Methyl Isobutyl Ketone	< 10	10	
			2-Hexanone	<10	10	
			Acrolein	<10	10	
			Acrylonitrile	<10	10	
			Methyltertiary Butyl Ether:	<1.0	1.0	±
			Iodomethane	<1.0	1.0	
		Surrogate Recoveries				
			1,2-Dichloroethane-d4	81	%	Recovery
			Toluene-d8	102		
			4-Bromofluorobenzene	113		

¹ Value derived from a <u>10x</u> dilution.

Site ————	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
nfluent-13			95-38231	8260 LONG			RH:07-13-9
						PQL	Units µg/L
			•	1,1-Dichloroethene	<1.0	1.0	
		•		Methylene Chloride	3.3	1.0	
				trans-1, 2-Dichloroethene	<1.0	1.0	
				1,1-Dichloroethane	<1.0	1.0	
				2, 2-Dichloropropane	<1.0	1.0	
				cis-1,2-Dichloroethene	301	1.0	
				Bromochloromethane	<1.0	1.0	
				Chloroform	<1.0	1.0	
				1,1,1-Trichloroethane Carbon Tetrachloride	<1.0	1.0 1.0	
				1,1-Dichloropropene	<1.0 <1.0	1.0	•
				Benzene	1.8	1.0	
				1.2-Dichloroethane	<1.0	1.0	
				Trichloroethene	6.1	1.0	
				1,2-Dichloropropane	<1.0	1.0	
				Dibromomethane	<1.0	1.0	
				Bromodichloromethane	<1.0	1.0	
				Trans-1,3-Dichloropropene	<1.0	1.0	
				Toluene	4.1	1.0	
				cis-1,3-Dichloropropene	<1.0	1.0	:
				1,1,2-Trichloroethane	<1.0	1.0	
				Tetrachloroethene	<1.0²	1.0	
				1,3-Dichloropropane	<1.0	1.0	
				Dibromochloromethane 1, 2-Dibromoethane	<1.0 <1.0	1.0 1.0	
				Chlorobenzene	<1.0	1.0	
				1,1,1,2-Tetrachloroethane	<1.0	1.0	
				Ethylbenzene	4.0	1.0	
				M + P Xylenes	24	1.0	
				O-Xylene	3.3	1.0	
				Styrene	<1.0	1.0	
				Bromoform	<1.0	1.0	
				Isopropylbenzene	1.0	1.0	
				Bromobenzene	<1.0	1.0	
				1,1,2,2-Tetrachloroethane	<1.0	1.0	
				1,2,3-Trichloropropane	<1.0	1.0	
				n-Propylbenzene	1.3	1.0	
				2-Chlorotoluene	<1.0	1.0	
				4-Chlorotoluene 1,3,5-Trimethylbenzene	<1.0 4.6	1.0 1.0	
				tert-Butylbenzene	4.6 <1.0	1.0	
				1,2,4-Trimethylbenzene	16'	1.0	
	•			sec-Butylbenzene	1.2	1.0	
				1,3-Dichlorobenzene	<1.0	1.0	
				1,4-Dichlorobenzene	<1.0	1.0	
				p-Isopropyltoluene	1.5	1.0	
				1,2-Dichlorobenzene	<1.0	1.0	
				n-Butylbenzene	1.2	1.0	
				1.2-Dibromo-3-Chloropropan		1.0	
				1,2,4-Trichlorobenzene	<1.0	1.0	
				Naphthalene	11	1.0	
				Hexachlorobutadiene	<1.0	1.0	

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
Influent-13	95-38231 8260 LONG continued				<u>PQL</u>	RH:07-13-9 Units <i>ug/</i> L	
							-143
				1,2,3-Trichlorobenzene	<1.0	1.0	
				Acetone	< 20	20	
				Methyl Ethyl Ketone	17	10	
				Dichlorodifluoromethane	<1.0	1.0	
				Chloromethane	<1.0	1.0	
				Vinyl Chloride	<1.0	1.0	
				Bromomethane	<1.0	1.0	
				Chloroethane	<1.0	1.0	
				Trichlorofluoromethane	<1.0	1.0	
				2-Chloroethylvinylether	<1.0	1.0	
				Carbon Disulfide	<1.0	1.0	
				Vinyl Acetate	< 1.0	1.0	
				Methyl Isobutyl Ketone	<10	10	
				2-Hexanone	<10	10	
				Acrolein	<10	10	
				Acrylonitrile	<10	10	
				Methyltertiary Butyl Ether-	<1.0	1.0	:
				lodomethane	< 1.0	1.0	
			Surrogate Recoveries				
			•	1,2-Dichloroethane-d4	83	9	6 Recovery
				Toluene-d8	102		,
				4-Bromofluorobenzene	102		

Kurt R. Slentz

Laboratory Manager

 $^{^{1}}$ Value derived from a $\underline{10x}$ dilution. 2 Present but less than the Practical Quantitation Limit.

Site Depth Lab No. Methodology Analysis Results Units Analyzed

QUALITY ASSURANCE DATA

Trip Blank	8260 LONG			RH:07-13-95
THE BIBLIK	8260 LONG		001	
	1,1-Dichloroethene	< 1.0	<u>PQL</u> 1.0	Units µg/L
	Methylene Chloride	<1.0	1.0	
	trans-1, 2-Dichloroethene	<1.0	1.0	
	1,1-Dichloroethane	<1.0	1.0	
	2,2-Dichloropropane	<1.0	1.0	
	cis-1,2-Dichloroethene	<1.0	1.0	
	Bromochloromethane ::	<1.0	1.0	
	Chloroform	<1.0	1.0	
	1,1,1-Trichloroethane	<1.0	1.0	
	Carbon Tetrachloride	<1.0	1.0	
	1,1-Dichloropropene	<1.0	1.0	
	Benzene	<1.0	1.0	
	1,2-Dichloroethane	<1.0	1.0	
	Trichloroethene	<1.0	1.0	
	1,2-Dichloropropane	<1.0	1.0	
	Dibromomethane	<1.0	1.0	
	Bromodichloromethane	<1.0	1.0	
	Trans-1,3-Dichloropropene	< 1.0	1.0	
	Toluene	<1.0	1.0 1.0	
	cis-1,3-Dichloropropene	<1.0		
	1,1,2-Trichloroetnane	<1.0	1.0 1.0	
	Tetrachloroethene	<1.0		
	1,3-Dichtoropropane	<1.0	1.0	
	Dibromochloromethane	< 1.0	1.0	
	1,2-Dibromoethane	<1.0	1.0	
	Chlorobenzene	< 1.0	1.0	
	1,1,1,2-Tetrachloroethane	<1.0	1.0	
	Ethylbenzene	<1.0	1.0	
	M +P Xylenes	<1.0	1.0	
	O-Xylene	<1.0	1.0	
	Styrene	<1.0	1.0	
	Bromoform	<1.0	1.0	
	Isopropyibenzene	<1.0	1.0	
	Bromobenzene	< 1.0	1.0	
	1,1,2,2-Tetrachloroethane	<1.0	1.0	
	1,2,3-Trichloropropane	<1.0	1.0	
	n-Propylbenzene	<1.0	1.0	
	2-Chlorotoluene	<1.0	1.0	
	4-Chlorotoluene	<1.0	1.0	
	1,3,5-Trimethylbenzene	<1.0	1.0	
	tert-Butylbenzene	<1.0	1.0	
	1,2,4-Trimethylbenzene	<1.0	1.0	
	sec-Butylbenzene	<1.0	1.0	
	1,3-Dichlorobenzene	<1.0	. 1.0	
	1,4-Dichlorobenzene	<1.0	1.0	
	p-Isopropyltoluene	<1.0	1.0	
	1,2-Dichlorobenzene	<1.0		
	n-Butylbenzene	<1.0	1.0	
	1,2-Dibromo-3-Chloropropane	<1.0	1.0	•
	1,2,4-Trichlorobenzene	<1.0	1.0	
	Naphthalene	<1.0	1.0	
	Hexachlorobutadiene	<1.0	1.0	

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
Frip Blank			8260 LONG conti	nued		<u>POL</u>	RH:07-13-9 Units <i>µg/</i> L
				1, 2, 3-Trichlorobenzene	<1.0	1.0	
				Acetone	< 20	20	
				Methyl Ethyl Ketone	<1.0	10	
				Dichlorodifluoromethane	<1.0	1.0	
				Chloromethane	<1.0	1.0	
				Vinyl Chloride	<1.0	1.0	
				Bromomethane	<1.0	1.0	
				Chloroethane	<1.0	1.0	
				Trichlorofluoromethane	<1.0	1.0	
				2-Chloroethylvinylether	<1.0	1.0	
				Carbon Disulfide	<1.0	1.0	
				Vinyl Acetate	< 1.0	1.0	
				Methyl Isobutyl Ketone	<10	10	
				2-Hexanone	<10	10	
				Acrolein	<10	10	
				Acrylonitrile	<10	10	
				Methyltertiary Butyl Ether-	<1.0	1.0	•
				Iodomethane	<1.0	1.0	
			Surrogate Recoveries				
				1,2-Dichloroethane-d4	82	%	Recovery
				Toluene-d8	109		
				4-Bromofluorobenzene	113		

ENERGY LABORATORIES, INC. RAPID CITY, SD

VOLATILE ORGANIC COMPOUNDS QUALITY ASSURANCE REPORT FORM

SAMPLE LOT SAMPLE MATRIX **EXTRACTION DATE** ANALYST

95-38231 Water

RH

MATRIX SPIKE / MATRIX SPIKE DUPLICATE DATA

Compound	Spike Added (µg)/L	Sample (µg)	Matrix Spike (µg)	Matrix Spike % Rec	Matrix Spike Duplicate (µg)	Matrix Spike Duplicate % Rec	% Difference (Difference) Average	QC Limits
1,1-Dichloroethene	10.0	<1.0	9.4	94	9.7	97	3.1	60-140%
Benzene	10.0	1.8	11.7	99	11.8	100	1.0	60-140%
Trichloroethene	10.0	6.1	15.6	95	15.6	95	0	60-140%
Toluene	10.0	4.1	14.0	99	13.1	90	9.5	60-140%
Chlorobenzene	10.0	<1.0	11.2	112	11.1	111	0.90	60-140%



ENERGY LABORATORIES, INC.

P.O. BOX 2470 • RAPID CITY, SD 57709 • PHONE (605) 342-1225 610 FARNWOOD STREET • RAPID CITY, SD 57701 • FAX (605) 342-1397

James Machin Radian Corporation 3201 C Street, Suite 405 Anchorage, AK 99503

Addendum to:

November 17, 1995 95-38229-31, 95-38191-95 95-38061-67, 95-38005-07

Site	Depth	Lab No.	Methodology	Analysis	Results	Units	Analyzed
Water Analysis							
MW930101	•	95-38005	1 EPA 8260 Mod	² TPM	<u>µg/L</u> 45,400	<u>PQL</u> 200	RH:06-26-95
Effluent-1		95-38006	EPA 8260 Mod	TPM	65	20	RH:06-25-95
Influent-1		95-38007	EPA 8260 Mod	TPM	41,300	200	RH:06-26-95
Influent-2		95-38061	EPA 8260 Mod	TPM	19,300	200	RH:06-30-95
Influent-3		95-38062	EPA 8260 Mod	TPM	112,100	200	RH:07-05-95
Influent-4		95-38063	EPA 8260 Mod	TPM	42,600	200	RH:07-05-95
Influent-5		95-38064	EPA 8260 Mod	TPM	24,300	230	RH:07-05-95
influent-6		95-38065	EPA 8260 Mod	TPM	28,100	200	RH:07-05-95
Influent-6D		95-38066	EPA 8260 Mod	TPM	31,600	200	RH:07-07-95
Influent-7		95-38067	EPA 8260 Mod	TPM	80,500	200	RH:07-06-95
Influent-8		95-38191	EPA 8260 Mod	TPM	56,700	200	RH:07-07-95
Influent-9		95-38192	EPA 8260 iviod	TPM	41,600	200	RH:07-07-95
Influent-11		95-38194	EPA 8260 Mod	TPM	65,600	200	RH:07-07-95
MW930101-Post	t Test	95-38195	EPA 8260 Mod	TPM	41,500	200	RH:07-05-95
Influent-12		95-38229	EPA 8260 Mod	TPM .	3,180	100	RH:07-13-95
Influent-120		95-38230	EPA 8260 Mod	TPM	2,990	100	RH:07-13-95
Influent-13		95-38231	EPA 8260 Mod	TPM	1,190	100	RH:07-13-95

¹ These data were generated using a modification of EPA Method 8260. The values included in this report represent the Total Purgeable Material that eluded in the approximate range of C_5 - C_{12} . The results for TPM were derived by total integration of the chromatogram from C_5 - C_{12} and comparing the resultant area to a mean response factor for three internal and three surrogate standards. As such, these data should be considered estimated values.

Kurt R. Slentz

Laboratory Manager

² TPM - Total Purgeable Material.

S voc dy 6-29-95 Received for laboratory by (signature): Received by (signature). PLEASE PRINT OR TYPE ALL INFORMATION EXCEPT SIGNATURES Mud 40c (8260) Comments, Special Instructions, etc. ROPINTAT 9-38-95 9021 156/52/ Time Time Date Date CHAIN OF CUSTODY RECORD Relinquished (signature) Relinquished (signature) belseupely sixVienA 7.8 Samole Type: A W S V U O
AirWate Solis/solids Yegetation Unne Other 4 pM 4 number of containers Received by: (signature) Received by: (signature) Sampler's signature Ellsworth AFB 2-Phase Test 605-342-1225 605-342-1397 SAMPLE LD. 90t) 5652-9 Time Effluent-1* voice Trip Blanks In fluent-1 Reporte: Radian Corp. moles to: Radian Corp. Date MW130101 610 Farnwood Street Project Name / Address Relinquished (signature) James Machin Contact Name & Phone composite grab sample P.O. Box 2470 610 Rapid City, SD 57709 G-15/19/1620 027 2-7 TIME 0211 82-9 P.O. # DATE

ENERGY LABORATORIES, INC.

Includes que tone truibk Received for laboratory by (signature): Received by (signature): PLEASE PRINT OR TYPE ALL INFORMATION EXCEPT SIGNATURES Comments, Special Instructions, etc. Marce - | Sh:11 | Sh/Lep) Time Date CHAIN OF CUSTODY RECORD Refinquished (signature) Relinquished (signature) Delseuper sieviens ۲ r ٤ ` v ۲ Sample Type: A W & Wormple Dinne Quner Airer |S-\S ` ۶-Z J ۲ number of containers Received by: (signature) Received by: (signature) Ellowth AFB 2 Amge Tost 605-342-1225 605-342-1397 Triv Blanks Influent-2 Ty flueut-CD Influent-3 IN fluent - F Influent 5 Influent -4 In fluent-6 voice 610 Farmwood Street Project Name / Address EMERGY LABORATORIES, INC. Invoice to: James Machin Contact Name & Phone P.O. Box 2470 611 Rapid City, SD 57709 1215 1834 1/27/95/1000 0760 3/12/9 7/3 TIME 156/27/2 DATE

Hoklsauple Influent-10. Object 10/29/951350 marcie Sonay Received by (signature): PLEASE PRINT OR TYPE ALL INFORMATION EXCEPT SIGNATURES Comments, Special Instructions, etc. Time Date CHAIN OF CUSTODY RECORD Relinquished (signature) Relinquished (signature) Descended sixylent 6928 Z .v > Sample Type: A WV S V O O O Vine Other A vier Water Soils/soils Yegetation Unine Other 3-6 ₹ 1 17 = number of containers Received by: (signature) Received by: (signature) Project Name I Address
Ellyworth AFB 2. More Test 605-342-1225 605-342-1397 In fluent-10# SAMPLE LD. MW930101-POS+TRST 0-5 81 45-52-9 Time Trip Hanks Iuf(4ent-11 In fluent - 9 In Flyeut-8 voice fax Date Contact Name & Phone 1562-610 Farmwood Street eldmss dsrp Report Invoice to: Relinquished (signature) P.O. Box 2470 61C Rapid City, SD 57709 × × بح × composite 1.28-84 0900 02915422: 16/0 e541 -1995 1135 TIME P.O. # DATE ~ >

ENERGY LABORATORIES, INC.

Received for laboratory by (signature): Received by (signature): Comments, Special Instructions, etc. 50:21 Time Time 130/8 Date Date Relinquished (signature) Relinquished (signature) Delseuper & Sevient X X × Sample Type: A W S U O O O Unie Qther ∆ire Qther Qther Qiller Quher 3-8 3-W 3. X number of containers Received by: (signature) Received by: (signature) SUITE 405 99503 ELLSWORTH AFFS 2-PHASE Sampler's signature F SEE SECTION STOLLS THAT CHIN 605-342-1225 605-342-1397 AKTime Time 1/3 YINFLUENT - 12D WFLUENT - 12 ANCHORAGE fax Date 562 0375 Date Tio Slaw V INFWENT Project Name / Address invoice to: Relinquished (signature) Relinquished (signature) Contact Name & Phone JAMES MINCHIN P.O. Box 2470 61 Rapid City, SD 57709 1630 6/29 1630 <u>R</u> TIME 6/29 6/30 P.O. # DATE

PLEASE PRINT OR TYPE ALL INFORMATION EXCEPT SIGNATURES

CHAIN OF CUSTODY RECORD

voice

610 Farnwood Street

EMERGY LABORATORIES, INC.

APPENDIX E

Vapor Sample Analytical Data

MICROSEPS

University of Pittsburgh Applied Research Center 220 William Pitt Way, Pittsburgh, PA 15238 (412) 826-5245 FAX (412) 826-3433

July 7, 1995

Mr. James Machin Radian Corporation 3201 C Street Suite 405 Anchorage, AK 99503

Dear Mr. Machin:

Attached is the final data listing for the samples we received on July 3, 1995, your project #612-001-31-37.

Please give me call if you have questions or I can be of further assistance. Thank you for using MICROSEEPS.

Sincerely,

David J. Masdea

Vand J. Marie

DJM/lsp

Attachment:

RAD36-952487



ANALYSIS OF VOLATILE ORGANICS IN GAS SAMPLES

Gas samples are received and secured in accordance with Microseeps documented sample receipt procedures. Analyses are performed using Microseeps Analytical Method AM4.02. Analytical method AM4.02 is a modification of USEPA Method 3810 (Headspace) and 8000 (Gas Chromatography). Modifications implemented are to accommodate the gas phase sample type only. All applicable quality control procedures are followed including continuing calibration check standards and laboratory blanks. Microseeps Analytical Method AM4.02 will be supplied upon request.

1 OF 3

---- RADIAN CORP. ----

RAD36-952487

---- PROJECT NO. 612-001-31-37 ----

----- PROJECT LOC: ELLSWORTH AFB(2ND PHASE TEST) -----

---- CONCENTRATIONS IN PPMV ----

			BROMO						•						
			METHANE/	FLUORO	1,1		TRANS-1,2	1,1		1,1,1	CARBON		1,2		1,2
SAMPLE	CHLORO	VINYL	CHLORO	CHLORO TRICHLORO	DICHLORO P	METHYLENE	DICHLORO	DICHLORO		TRICHLORO	TETRA		DICHLORO	TRICHLORO	DICHLORO
NAME	METHANE CHLORIDE	CHLORIDE	ETHANE*	METHANE	ETHYLENE	CHLORIDE	ETHYLENE	ETHANE C	ETHANE CHLOROFORM	ETHANE	CHLORIDE	BENZENE	ETHANE	ETHYLENE	PROPANE
V-1	-	⊽	⊽	<.005	0.54	₽	0.5	0.69	0.006	<.005	<.005	37.99	0.05	0.633	.01
٧-2	₹	⊽	⊽	<.005	0.57	▽	0.5	0.74	9000	<.005	<.005	62.56	0.11	1.813	0
۷-3	⊽	⊽	⊽	<.005	0.59	⊽	0.5	0.77	0.005	<.005	<.005	67.39	0.12	2.419	01
V-4	_	⊽	⊽	<.005	0.56	⊽	0.5	0.77	<.005	<.005	<.005	66.64	0.12	2.569	01
۷-5	⊽	⊽	⊽	<.005	0.58	⊽	9.0	0.77	<.005	<.005	<.005	72.64	0.12	2.953	<.01
y- 9	⊽	⊽	⊽	<.005	27.0	⊽	9.0	0.80	<.005	900.0	<.005	67.63	0.11	3.477	<.01
۷-7	⊽	⊽	⊽	<.005	0.43	⊽	9.0	0.79	<.005	900.0	<.005	61.42	0.12	3.370	01
N-8	⊽	₹	₹	<.005		₩	9.0	0.79	<.005	0.006	<.005	58.98	0.12	3,475	01
V-8D	⊽	⊽	⊽	<.005		∇	9.0	0.78	<.005	0.006	<.005	53.78	0.12	3.195	. .01
6-7	⊽	⊽	⊽	<.005		⊽	9.0	0.78	<.005	200.0	<.005	50.46	0.12	3.694	. .01
V-10	⊽	₹	⊽	<.005	0.23	⊽	0.5	0.71	<.005	9000	<.005	40.84	90.0	3.058	01
V-11	⊽	⊽	₩	<.005		⊽	0.5	0.75	<.005	0.007	<.005	49.85	0.12	4.064	^.01
V-13	⊽	⊽	⊽	<.005	0.18	⊽	0.5	0.72	<.005	0.007	<.005	37.71	0.07	3.558	01
V-14	⊽	⊽	⊽	<.005	0.22	⊽	0.5	5.0°	<.005	0.008	<.005	39.93	90.0	3.590	. .01
V-15	⊽	⊽	⊽	<.005	*.01	⊽	0.2	0.29	<.005	<.005	<.005	0.26	0.02	0.012	90.0
V-15D	⊽	⊽	₩	<.005	~.01	⊽	0.2	0.28	<.005	<.005	<.005	0.25	0.01	0.010	90.0
V-16	⊽	⊽	⊽	<.005	. .01	⊽	0.1	0.36	<.005	<.005	<.005	0.25	0.02	0.006	0.05
MDLs	-	-	-	0.005	0.01	-	0.01	0.01	0.005	0.005	0.005	0.07	0.01	0.005	0.01

^{*} COMPOUNDS ELUTE TOGETHER ON ECD: VALUES REPRESENT EITHER OR A COMBINATION OF BOTH.

RAD36-952487

---- PROJECT NO. 612-001-31-37 ----

----- PROJECT LOC: ELLSWORTH AFB(2ND PHASE TEST) -----

---- CONCENTRATIONS IN PPMV ----

1,3 1,4 1,2 DICHLORO DICHLORO BENZENE BENZENE	<.07 <.07		<.07 <.07	•	·	•	·	·	·	·	·	·	•	·	Ĭ	·		
1,3 ICHLORO DICH BENZENE BEN	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	
1,1,2,2 TETRA CHLORO D	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	
BROMO	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	
ETHYL Benzene	1.26	5.34	7.02	7.24	7.28	8.54	8.34	8.05	7.55	8.26	7.25	8.66	7.27	4.34	0.21	0.19	~.07	
CHLORO BENZENE	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	
CHLORO Dibromo Methane	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	
TETRA CHLORO ETHYLENE	0.038	0.145	0.216	0.233	0.256	0.343	0.339	0.351	0.328	0.417	0.371	0.522	0.520	0.359	0.010	0.009	<.005	
NS-1,3 1,1,2 CHLORO TRICHLORO PYLENE ETHANE E	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	
TRANS-1,3 DICHLORO T PROPYLENE	01	01	*.01	. .01	. .01	~.01	*.01	*.01	01	01	*.01	*.01	01	^.01	*.01	01	~.01	
	<07	<.07	<.07	<.07	<.07	*.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	
CIS-1,3 DICHLORO PROPYLENE TOLUENE	<.01	. .01	*.01	. .01	*.01	. .01	*. 01	*.01	. .01	·.01	·.01	*.01	~.01	·.01	. .01	. .01	~.01	
BROMO DICHLORO METHANE P	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	
SAMPLE	۷-1	V-2	1-3	1-4	7-5	9-/	1-7	8-/	7-80	6-1	7-10	/-11	7-13	1-14	/-15	7-150	v-16	

ANALYST INITIALS /////

MICROSEEPS 07-Jul-95

---- RADIAN CORP. ----

RAD36-952487

3 OF 3

---- PROJECT NO. 612-001-31-37 ----

---- PROJECT LOC: ELLSWORTH AFB(2ND PHASE TEST) ----

---- CONCENTRATIONS IN PPMV ----

ADDITIONAL ANALYSIS

•										
SAMPLE			M&P	ò	TOTAL	FILE	DATE	TIME	DATE	DATE
NAME	ACETONE	MIBK	XYLENE	XYLENE	c5-c10	NAME	SAMPLED	SAMPLED	RECEIVED	ANALYZED
V-1	\$		1.13	0.38	3628.59	W53 279	06/25/95	1156	07/03/95	07/04/95
۷-2	\$	⊽	9.83	3.48	6320.99	W53 280	06/25/95	1322	07/03/95	07/04/95
V-3	?	⊽	19.43	6.16	6932.84	W53 281	06/25/95	1602	07/03/95	07/04/95
7- A	\$	⊽	21.84	6.77	6784.85	W53 282	06/25/95	1828	07/03/95	07/04/95
۷-5	?	⊽	22.49	6.70	7193.84	W53 283	06/25/95	2054	07/03/95	07/04/95
9- A	\$	⊽	28.29	8.72	6750.07	W53 284	06/26/95	006	07/03/95	07/04/95
V-7	8	⊽	28.01	8.90	6254.79	W53 285	06/26/95	1215	07/03/95	07/04/95
V-8	?	⊽	27.54	8.73	80.6209	W53 286	06/26/95	1638	07/03/95	07/04/95
V-80	\$	~	25.80	8.38	5585.52	W53 287	36/26/95	1639	07/03/95	07/04/95
6-7	\$	⊽	29.22	9.45	5386.28	W53 288	06/27/95	876	07/03/95	07/04/95
V-10	\$	⊽	26.54	8.82	4430.23	W53 292	06/27/95	1614	07/03/95	07/04/95
V-11	\$	⊽	31.86	10.18	5354.66	W53 293	06/28/95	206	07/03/95	07/04/95
V-13	\$	⊽	29.52	9.98	4267.64	W53 294	06/29/95	1130	07/03/95	07/04/95
V-14	\$	⊽	16.49	4.68	3935.96	W53 295	06/29/95	1338	07/03/95	26/50/20
V-15	\$	⊽	1.40	0.82	52.01	W53 296	06/59/95	1645	07/03/95	07/05/95
V-150	\$	⊽	1.29	0.82	48.78	W53 297	06/29/95	1645	07/03/95	07/05/95
V-16	%	⊽	0.08	<.07	8.91	W53 298	96/30/92	1115	07/03/95	07/05/95
MDLS	8		20.	.07	.07					

ANALYST INITIALS WILL

RAD36-952487

---- RADIAN CORP. ----

----- PROJECT NO. 612-001-31-37 -----

---- PROJECT LOC: ELLSWORTH AFB(2ND PHASE TEST) ------ CONCENTRATIONS IN PPMV -----

CONTINUING CALIBRATION CHECK

STANDARDS: "624"(LEVEL 2), "624"(LEVEL 1), "VC-996", "8240D", "L" R4

REFERENCE: W53A/B268, W53A/B269, W53A272, W53A277, W53A252

COMPOUND	KNOWN	RESULT	PERCENT DIFFERENCE
CHLOROMETHANE	20.8	22.6	8.03
VINYL CHLORIDE	996.0	975.5	2.10
BROMOMETHANE/CHLOROETHANE*	2.7	2.6	3.36
FLUOROTRICHLOROMETHANE	0.765	0.744	2.82
1,1 DICHLOROETHYLENE	1.09	1.07	1.12
METHYLENE CHLORIDE	1.24	1.20	3.2 5
TRANS-1,2 DICHLOROETHYLENE	1.09	1.07	1.40
1,1 DICHLOROETHANE	1.06	1.05	0.85
CHLOROFORM	0.881	0.865	1.85
1,1,1 TRICHLOROETHANE	0.788	0.775	1.68
CARBON TETRACHLORIDE	0.684	0.671	1.94
BENZENE & 1,2-DCA**	2.41	2.38	1.35
1,2 DICHLOROETHANE	1.06	1.04	1.82
TRICHLOROETHYLENE	0.800	0.784	2.04
1,2 DICHLOROPROPANE	0.93	0.91	2.20
BROMODICHLOROMETHANE	0.642	0.627	2.39
CIS-1,3 DICHLOROPROPYLENE	0.95	0.92	3.16
TOLUENE	1.14	1.04	9.40
TRANS-1,3 DICHLOROPROPYLENE	0.95	0.91	3. 83
1,1,2 TRICHLOROETHANE	0.788	0.780	1.03
TETRACHLOROETHYLENE	0.634	0.629	0.79
CHLORODIBROMOMETHANE	0.505	0.496	1.81
CHLOROBENZENE	0.93	0.88	6.74
ETHYL BENZENE	0.99	0.95	4.54
BROMOFORM	0.416	0.406	2.46
1,1,2,2 TETRACHLOROETHANE	0.626	0.620	0.97
1,3 DICHLOROBENZENE	0.72	0.68	5.30
1,4 DICHLOROBENZENE	0.72	0.72	0.56
1,2 DICHLOROBENZENE	0.72	0.71	0.85
ACETONE	33.5	29.0	15.56
MIBK	21.0	19.0	10.31
M&P XYLENE	1.71	1.90	10.00
O-XYLENE	0.86	0.87	0.58

^{*} COMPOUNDS ELUTE TOGETHER ON ECD - VALUES REPRESENT EITHER OR A COMBINATION OF BOTH.

ANALYST INITIALS

^{**} COMPOUNDS ELUTE TOGETHER ON FID - VALUE REPRESENTS A COMBINATION OF BOTH.

RAD36-952487

---- RADIAN CORP. ----

---- PROJECT NO. 612-001-31-37 ----

LOWER

---- PROJECT LOC: ELLSWORTH AFB(2ND PHASE TEST) ---CONCENTRATIONS IN PPMV ----

LABORATORY BLANK RESULTS

N2 IN VIAL

REFERENCE: W53A/B276

		DETECTION
COMPOUND	BLANK	LIMIT
CHLOROMETHANE	ND	1.0
VINYL CHLORIDE	ND	1.0
BROMOMETHANE/CHLOROETHANE*	ND	1.0
FLUOROTRICHLOROMETHANE	ND	0.005
1,1 DICHLOROETHYLENE	ND	0.01
METHYLENE CHLORIDE	ND	1.00
TRANS-1,2 DICHLOROETHYLENE	ND	0.1
1,1 DICHLOROETHANE	ND	0.01
CHLOROFORM	ND	0.005
1,1,1 TRICHLOROETHANE	ND	0.005
CARBON TETRACHLORIDE	ND	0.005
BENZENE	ND	0.07
1,2 DICHLOROETHANE	ND	0.01
TRICHLOROETHYLENE	ND	0.005
1,2 DICHLOROPROPANE	ND	0.01
BROMODICHLOROMETHANE	ND	0.005
CIS-1,3 DICHLOROPROPYLENE	ND	0.01
TOLUENE	ND	0.07
TRANS-1,3 DICHLOROPROPYLENE	ND	0.01
1,1,2 TRICHLOROETHANE	ND	0.005
TETRACHLOROETHYLENE	ND	0.005
CHLOROD I BROMOMETHANE	ND	0.005
CHLOROBENZENE	ND	0.07
ETHYL BENZENE	ND	0.07
BROMOFORM	ND	0.005
1,1,2,2 TETRACHLOROETHANE	ND	0.005
1,3 DICHLOROBENZENE	ND	0.07
1,4 DICHLOROBENZENE	ND	0.07
1,2 DICHLOROBENZENE	ND	0.07
ACETONE	ND	2.0
MIBK	ND	1.0
M&P XYLENE	ND	0.07
O-XYLENE	ND	0.07

^{*} COMPOUNDS ELUTE TOGETHER ON ECD - VALUES REPRESENT EITHER OR A COMBINATION OF BOTH.

ANALYST INITIALS ACC



University of Pittsburgh Applied Research Center 220 William Pitt Way, Pittsburgh, PA 15238 (412) 826-5245 FAX (412) 826-3433

November 20, 1995

Mr. James Machin Radian Corporation 3201 C Street Suite 405 Anchorage, AK 99503

Dear Mr. Machin:

Attached is the additional data listing for the samples we received on July 3, 1995, your project #612-001-31-37.

Please give me call if you have questions or I can be of further assistance. Thank you for using MICROSEEPS.

Sincerely,

David J. Masdea

Zuid T. Wonden

DJM/lsp

Attachment:

RAD36-952487

LAB MANAGER INITIALS DOT

1 OF 3

---- PROJECT NO. 612-001-31-37 -------- RADIAN CORP. ----

RAD36-952487

----- PROJECT LOC: ELLSWORTH AFB(2ND PHASE TEST) ----

---- CONCENTRATIONS IN PPMV ----

			BROMO												
			METHANE/	FLUORO	1,1		TRANS-1,2	1,1		1,1,1	CARBON		1,2		1,2
SAMPLE	CHLORO	VINYL	CHLORO	CHLORO TRICHLORO	DICHLORO	METHYLENE	DICHLORO	DICHLORO		TRICHLORO	TETRA		DICHLORO	TRICHLORO	DICHLORO
NAME	METHANE CHLORIDE	HLORIDE	ETHANE*	METHANE	ETHYLENE	CHLORIDE	ETHYLENE	ETHANE (ETHANE CHLOROFORM	ETHANE	CHLORIDE	BENZENE	ETHANE	ETHYLENE	PROPANE
V-1	-	⊽	₹	<.005	0.54		0.5		0.006	<.005	<.005	37.99	0.05	0.633	٠.01
۷-2	⊽	∇	⊽	<.005	0.57	⊽	0.5	0.74	0.006	<.005	<.005	62.56	0.11	1.813	<.01
٧-3	⊽	⊽	₹	<.005	0.59		0.5		0.005	<.005	<.005	67.39	0.12	2.419	01
۸-4	-	⊽	⊽	<.005	0.56	⊽	0.5		<.005	<.005	<.005	49.99	0.12	2.569	<.01
٧-5	₹	⊽	₹	<.005	0.58	⊽	9.0		<.005	<.005	<.005	72.64	0.12	2.953	<.01
۸-6	₹	⊽	⊽	<.005	0.47	⊽	9.0		<.005	0.006	<.005	67.63	0.11	3.477	<.01
٧-7	7	∇	⊽	<.005	0.43	⊽	9.0		<.005	0.006	<.005	61.42	0.12	3.370	01
N-8	7	~	⊽	<.005	07.0	₩	9.0		<.005	900.0	<.005	58.98	0.12	3.475	01
V-80	₹	₹	⊽	<.005	0.36	⊽	9.0		<.005	0.006	<.005	53.78	0.12	3.195	<.01
6-7	₹	⊽	⊽	<.005	0.31	⊽	9.0		<.005	0.007	<.005	20.46	0.12	3.694	<.01
V-10	۲	₩	⊽	<.005	0.23	⊽	0.5		<.005	0.006	<.005	40.84	90.0	3.058	<.01
V-11	⊽	⊽	⊽	<.005	0.27	▽	0.5		<.005	0.007	<.005	49.85	0.12	4.064	<.01
V-13	2	⊽	⊽	<.005		⊽	0.5		<.005	0.007	<.005	37.71	0.07	3.558	<.01
V-14	2	⊽	₹	<.005		⊽	0.5		<.005	_	<.005	39.93	90.0	3.590	<.01
V-15	2	⊽	₹	<.005		⊽	0.2		<.005	·	<.005	0.26	0.02	0.012	90.0
V-15D	2	⊽	⊽	<.005	01	⊽	0.2		<.005	•	<.005	0.25	0.01	0.010	90.0
V-16	₽	₽	▽	<.005	<.01	₹	0.1		<.005	<.005	<.005	0.25	0.02	0.006	0.05
MDLs	-	-	-	0.005	0.01	-	0.1	0.01	0.005	0.005	0.005	0.07	0.01	0.005	0.01

^{*} COMPOUNDS ELUTE TOGETHER ON ECD: VALUES REPRESENT EITHER OR A COMBINATION OF BOTH.

---- RADIAN CORP. ----

2 OF 3

---- PROJECT NO. 612-001-31-37 ----

----- PROJECT LOC: ELLSWORTH AFB(ZND PHASE TEST) -----

1,2)ICHLORO	BENZENE	~.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	0.07
1,4 1,2 DICHLORO DICHLORO	BENZENE	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	0.07
1,3 DICHLORO D	BENZENE	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	0.07
1,1,2,2 TETRA CHLORO D		<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	0.005
BROMO	FORM	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<,005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	0.005
ETHYL	BENZENE	1.26	1.26	5.34	7.02	7.24	7.28	8.54	8.34	8.05	7.55	8.26	7.25	8.66	7.27	4.34	0.21	0.19	<.07	0.07
CHLORO	;	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	<.07	0.07
CHLORO DIBROMO	METHANE	<.005	<,005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	0.005
TETRA	ETHYLENE	0.038	0.038	0.145	0.216	0.233	0.256	0.343	0.339	0.351	0.328	0.417	0.371	0.522	0.520	0.359	0.010	0.009	<.005	0.005
1,1,2 TRICHLORO		<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<,005	< 002	<.005	<,005	<.005	<.005	<.005	<.005	0.005
TRANS-1,3 DICHLORO T		<.01	01	*.01	*. 01	. .01	<.01	01	. .01	*.01	~.01	*.01	*.01	*. 01	. .01	. .01	01	. .01	. .01	0.01
-	TOLUENE F	<.07	<.07	<.07	*.07	<.07	<.07	<.07	<.07	~.07	~.07	<.07	<.07	<.07	**.07	<.07	<.07	<.07	~.07	0.07
CIS-1,3 DICHLORO	PROPYLENE	. .01	~.01	. .01	. .01	. .01	. .01	01	01	·.01	. .01	~.01	01	. .01	^.01	. .01	~.01	. .01	. .01	0.01
BROMO DICHLORO		<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	<.005	< 002	0.005
SAMPLE	NAME	V-1	V-1	V-2	٧-3	7- 4	٧-5	V-6	۷-7	۷-8	V-8D	6-7	v-10	V-11	V-13	V-14	V-15	V-15D	V-16	MDLs

ANALYST INITIALS

---- RADIAN CORP. ----

RAD36-952487

---- PROJECT NO. 612-001-31-37 ----

----- PROJECT LOC: ELLSWORTH AFB(ZND PHASE TEST) -----

---- CONCENTRATIONS IN PPMV ----

ADDITIONAL ANALYSIS

1			9	•	CIS-1,2	14707			1		
SAMPLE	ACETONE	MIBK	XYLENE	XYLENE	ETHYLENE	101AL C5-C10	NAME	SAMPLED	SAMPLED	DATE RECEIVED	ANALYZED
	\$	⊽	1.13	0.38	61.9	3628.59	W53 279	06/25/95	1156	07/03/95	07/04/95
	\$	⊽	9.83	3.48	71.0	6320.99	W53 280	06/25/95	1322	07/03/95	07/04/95
	\$	⊽	19.43	6.16	74.0	6932.84	W53 281	06/25/95	1602	07/03/95	07/04/95
	\$	₹	21.84	6.77	6.47	6784.85	W53 282	06/25/95	1828	07/03/95	07/04/95
	\$	⊽	22.49	6.70	81.0	7193.84	W53 283	06/25/95	2054	07/03/95	07/04/95
	\$	⊽	28.29	8.72	75.9	6750.07	W53 284	06/26/95	006	07/03/95	07/04/95
۷-7	\$	⊽	28.01	8.90	70.2	6254.79	W53 285	06/26/95	1215	07/03/95	07/04/95
	\$	⊽	27.54	8.73	68.2	80.6209	W53 286	06/26/95	1638	07/03/95	07/04/95
	\$	⊽	25.80	8.38	63.2	5585.52	W53 287	06/26/95	1639	07/03/95	07/04/95
	\$	⊽	29.25	6.45	60.1	5386.28	W53 288	06/27/95	8%6	07/03/95	07/04/95
	\$	⊽	26.54	8.82	47.7	4430.23	W53 292	06/27/95	1614	07/03/95	07/04/95
	\$	⊽	31.86	10.18	57.0	5354.66	W53 293	06/28/95	905	07/03/95	07/04/95
	\$	⊽	29.52	9.98	44.5	4567.64	W53 294	06/59/95	1130	07/03/95	07/04/95
	\$	⊽	16.49	4.68	49.3	3935.96	W53 295	06/59/95	1338	07/03/95	07/05/95
	\$	⊽	1.40	0.82	0.2	52.01	W53 296	06/29/95	1645	07/03/95	07/05/95
	\$	⊽	1.29	0.82	0.2	48.78	W53 297	06/59/95	1645	07/03/95	07/05/95
	\$	⊽	0.08	<.07	·.	8.91	W53 298	06/30/95	1115	07/03/95	07/05/95
MDES	8	•	20.	.07	0.1	.07					
)	t	•	•								

ANALYST INITIALS # C

MICROSEEPS 16-Nov-95

MICROSEEPS, Inc.

220 William Pitt Way, Pittsburgh, PA 15238

Phone: (412) 826-5245 Fax: (412) 826-3433

ourps Madin Radian Corp. Address: (3201 C5+ 1) Proj. Manager, uikuto Company Name:

5000+th 21.2-0378 Phone # : 9c ₹ / Proj. Location: Proj. Number:

Sampler's signature:

CHAIN-OF-CUSTODY RECORD

Page 1/2

V # C / ... C # X/

Note: Enter proper letters in Requested Analyses columns below.

Analysis Options

Note: If analysis D, E, or K is selected, scratch (option) NOT wanted.

# A C1 - C4 # B Hydrogen & Helium # C Permanent Gases (CH4, CO, CO2, N2, O2) Mercury (Soil) or (Air **) TO-14 by GC/MS (Ambient) or (Source **) F 601 & 602 Compounds	
--	--

- An additional 22 ml vial of sample is required when requested in combination with another analysis.
- Available upon request.

Collection	tion	Number of	# summs	Sample	Sample							
Date	Time	Containers	if Can. used	Туре	Identification	Reg	Requested Analyses	alyses	(Other)		Remarks	S
6-25-15 1156	1156	/		Sc)(905	1~1	F			acetone, MIBK, X/fene S)	ok, Xyene S		Same andlyses
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Relinquished by:	l by :		Company:		Date: Time:	Received by :	l by :		Company:		Date:	Time:
										L		

PINK COPY: Submitter

YELLOW COPY: Laboratory

WHITE COPY: Laboratory to return.

ここうしゅ

Dave Mosdea or Clerk Show lock MICROSEEPS, Inc.

220 William Pitt Way, Pittsburgh, PA 15238

Phone: (412) 826-5245 Fax: (412) 826-3433

Company Name:

Address: 3201 CSt. Suite 405, Anchorage AK Proj. Manager:

Ellsworth AFB 2- Mose Proj. Location: Proj. Number:

Phone #: 727/562-0375

James Alachin Radian Corp.

Sampler's signature:

CHAIN-OF-CUSTODY RECORD Proe 2 /2 Note: Enter proper letters in Requested Analyses columns below.

Analysis Options

Note: If analysis D, E, or K is selected, scratch (option) NOT wanted.

•	* A C1 -C4			ø	G Chlorinated HC
# B	Hydrogen	Hydrogen & Helium		•	B BTEX
Ø.	* C Permanent Gases	it Gases	(CH4, CO, CO2, N2, O2)		BTEX & C5 - C10
A	Mercury	D Mercury (Soil) or (Air **)	(Air **)	X	TPH (CS-C10) or (C4-C12)
E	TO-14 by	GC/MS	E TO-14 by GC/MS (Ambient) or (Source **)	1	L C11 - C18
B	601 & 602	F 601 & 602 Compounds	spu	Other	Other Specify below.

An additional 22 ml vial of sample is required when requested in combination with another analysis.

Available upon request.

Collection	Number of	*Summs*	Sample	Sample				
Date Time	Containers	if Can. used	Туре	Identification	Requested Analyses	(Other)	Remarks	
6-28-45 0902	۲ /		2-phase	~ 11-A		acetone, MIBK, * yle n	acetone, MIKK, Nenes - all gast ples	(÷
5191 56 17-9			1)	<i>\$</i> 721~ ∧	· ·	P/oH "	Hold sample V-17, Do not analyze	} %
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				,				
Results to: //								_
24/2	JAMES MACHINA	> // +						
Relinguished by Calle		Company:		Date: Time: (5/70/95/700)	Received by Holden	MCPOSEEPS Date 183 KS	Date; Time:	
Relifiquished by:		Company:		Date: Time:	Received by :	Company:	Date: Time:	
Relinquished by:		Company:		Date: Time:	Received by :	Company:	Date: Time:	

PINK COPY: Submitter

YELLOW COPY : Laboratory

WHITE COPY : Laboratory to return.